## PHYS851 Quantum Mechanics I, Fall 2009 HOMEWORK ASSIGNMENT 12

**Topics Covered:** Motion in a central potential, spherical harmonic oscillator, hydrogen atom, orbital electric and magnetic dipole moments

- 1. [20 pts] A particle of mass M and charge q is constrained to move in a circle of radius  $r_0$  in the x-y plane.
  - (a) If no forces other than the forces of constraint act on the particle, what are the energy levels and corresponding wavefunctions?

If the particle is forced to remain in the x-y plane, then it can only have angular momentum along the z-axis, so that  $\vec{L} = L_z \vec{e}_z$  and  $L^2 = L_z^2$ . The kinetic energy can be found two ways:

Method 1: Using our knowledge of angular momentum. We start by choosing  $\phi$  as our coordinate

$$H = \frac{L^2}{2I} = \frac{L_z^2}{2Mr_0^2}$$
(1)

so that the eigenstates are eigenstates of  $L_z \to -i\hbar$ partial<sub> $\phi$ </sub>, from which we see know that the energy levels are then  $E_m = \frac{\hbar^2 m^2}{2Mr_0^2}$ , where  $m = 0, \pm 1, \pm 2, \pm 3...$ , and the wavefunctions are  $\langle \phi | m \rangle = \frac{1}{\sqrt{2\pi}} e^{im\phi}$ .

Method 2: Solution from first principles. We start by choosing s as our coordinate, where s is the distance measured along the circle. The classical Lagrangian is then

$$\mathcal{L} = \frac{M\dot{s}^2}{2} \tag{2}$$

the canonical momentum is  $p_s = \partial_s L = M\dot{s}$ . The Hamiltonian is then

$$H = p\dot{s} - \mathcal{L} = \frac{p_s^2}{2M} \tag{3}$$

promoting s and  $p_s$  to operators, we must have  $[S, P_s] = i\hbar$ , so that in coordinate representation, we can take  $S \to s$ , and  $P_s \to -i\hbar\partial_s$ , which gives

$$H = -\frac{\hbar^2}{2M}\partial_s^2 \tag{4}$$

the energy eigenvalue equation is then

$$-\frac{\hbar^2}{2M}\partial_s^2\psi(s) = E\psi(s) \tag{5}$$

or equivalently

$$\partial_s^2 \psi(s) = -\frac{2ME}{\hbar^2} \psi(s) \tag{6}$$

This has solutions of the form:

$$\psi(s) \propto e^{\pm i \frac{\sqrt{2ME}}{\hbar}s} \tag{7}$$

single-valuedness requires

$$\psi(s + 2\pi r_0) = \psi(s) \tag{8}$$

which means

$$\frac{\sqrt{2ME}}{\hbar}2\pi r_0 = 2\pi m \tag{9}$$

where m is any integer. This gives

$$E = \frac{\hbar^2 m^2}{2M r_0^2} \tag{10}$$

so that

$$\psi_m(s) = \frac{e^{ims/r_0}}{\sqrt{2\pi r_0}} \tag{11}$$

Both methods agree because  $s = r_0 \phi$ .

(b) A uniform, weak magnetic field of amplitude  $B_0$  is applied along the z-axis. What are the new energy eigenvalues and corresponding wavefunctions?

Using the angular momentum method, we now need to add the term  $-\frac{qB_0}{2M}L_z$  to the Hamiltonian to account for the orbital magnetic dipole moment, which gives

$$H = \frac{L_z^2}{2Mr_0^2} - \frac{qB_0}{2M}L_z$$
(12)

so that the eigenstates are still  $L_z$  eigenstates,  $\psi_m(\phi) = \frac{e^{im\phi}}{\sqrt{2\pi}}$ , where  $m = 0, \pm 1, \pm 2, \ldots$ , but the degeneracy is lifted so that

$$E_m = \frac{\hbar^2 m^2}{2M r_0^2} - \frac{q B_0 \hbar}{2M} m$$
(13)

(c) Instead of a weak magnetic field along the z-axis, a uniform electric field of magnitude  $E_0$  is applied along the x-axis. Find an approximation for the low-lying energy levels that is valid in the limit  $qr_0E_0 \gg \hbar^2/Mr_0^2$ .

Hint: try expanding around the potential about a stable equilibrium point.

Here we need to add the electric monopole energy. The electrostatic potential of a uniform E-field along  $\vec{e}_x$  is  $\phi(\vec{r}) = -E_0 x$ , so that the potential energy is  $U = -qE_0 x$ . The full Hamiltonian of the particle is then given by

$$H = \frac{L_z^2}{2Mr_0^2} - qE_0r_0\cos(\phi)$$
(14)

The stable equilibrium point is at  $\phi = 0$ . Expanding to second-order about the equilibrium then gives

$$H = -\frac{\hbar^2}{2Mr_0^2}\partial_{\phi}^2 - qE_0r_0 + \frac{qE_0r_0}{2}\phi^2$$
(15)

This is just a harmonic oscillator Hamiltonian, with  $M_{eff} = Mr_0^2$ , and  $\omega = \sqrt{\frac{qE_0}{2Mr_0}}$ , so that the energy levels are

$$E_n = -qE_0r_0 + \hbar \sqrt{\frac{qE_0}{2Mr_0}} \left(n + \frac{1}{2}\right)$$
(16)

where  $n = 0, 1, 2, \ldots$  This approximation must be valid only when the level spacing is small compared to the depth of the cos potential, so that

$$\hbar \sqrt{\frac{qE_0}{2Mr_0}} \ll qE_0 r_0 \tag{17}$$

which is equivalent to

$$\frac{\hbar^2}{2Mr_0^2} \ll qE_0r_0\tag{18}$$

2. [10 pts] Write out the fully-normalized hydrogen wavefunctions for all of the 3p orbitals. Expand out any special functions in terms of elementary functions. You can look these up in a book or on-line, but keep in mind that you will be penalized if your expression is not properly normalized.

We have

$$\psi_{n,\ell,m}(r,\theta,\phi) = \sqrt{\frac{8(n-\ell-1)!}{2n(a_0n)^3(n+\ell)!}} e^{-r/a_0n} \left(\frac{2r}{a_0n}\right)^\ell L_{n-\ell-1}^{(2\ell+1)} \left(\frac{2r}{a_0n}\right) Y_\ell^m(\theta,\phi)$$
(19)

Using Mathematica, I then get for n = 3 and  $\ell = 1$ ,

$$\psi_{3,1,1}(r,\theta,\phi) = \frac{1}{81a_0^{7/2}\sqrt{\pi}} e^{-r/3a_0}(6a_0-r)r\sin\theta e^{i\phi}$$
(20)

$$\psi_{3,1,0}(r,\theta,\phi) = \frac{\sqrt{2}}{81a_0^{7/2}\sqrt{\pi}}e^{-r/3a_0}(6a_0-r)r\cos\theta$$
(21)

$$\psi_{3,1,-1}(r,\theta,\phi) = \frac{1}{81a_0^{7/2}\sqrt{\pi}}e^{-r/3a_0}(6a_0-r)r\sin\theta e^{-i\phi}$$
(22)

1

Normalization checks out:

Untitled-1

$$In[823]:= Clear[\psi, n, 1, m, r, \theta, \phi, a]$$

$$In[863]:= \psi[n_{-}, 1_{-}, n_{-}, r_{-}, \theta_{-}, \phi_{-}] := \sqrt{\frac{8(n-1-1)!}{2n(an)^{3}(n+1)!}} Exp[\frac{-r}{an}] (\frac{2r}{an})^{1} LaguerreL[n-1-1, 21+1, \frac{2r}{an}] SphericalHarmonicY[1, m, \theta, \phi]$$

$$In[864]:= \psi311 = PullSimplify[\psi[3, 1, 1, r, \theta, \phi]]$$

$$Out[864]= \sqrt{\frac{1}{a^{n}}} e^{-\frac{r}{3}r^{1+0}r} (-6a+r) Sin[\theta] (\sqrt{\frac{1}{a^{n}}} e^{-\frac{r}{3}r^{1+0}r} \sqrt{\frac{2}{n}} (6a-r) r Cos[\theta] (\sqrt{\frac{1}{a^{n}}} e^{-\frac{r}{3}r^{1+0}} (6a-r) r Sin[\theta] (\sqrt{\frac{1}{a^{n}}} e^{-\frac{r}{3}r^{1+0}} (6a-r) (r ) sin[\theta] (\sqrt{\frac{1}{a^{n}}} e^{-\frac{r}{3}r^{1+0}} (6a-r) (r ) (\sqrt{\frac{1}{a^{n}}} e^{-\frac{r}{3}r$$

3. [20 pts] Numerically compute the matrix elements of the z-component of the orbital electric and magnetic dipole moments for the  $|200\rangle \rightarrow |100\rangle$ ,  $|210\rangle \rightarrow |100\rangle$ , and  $|211\rangle \rightarrow |100\rangle$  transitions in hydrogen. Be sure to show your work.

For the electric dipole moments, we need to compute  $e\langle i|Z|f\rangle = e\langle i|R\cos\Theta|f\rangle$ . The selection rules are  $m_f = m_i$  and  $L_f = L_i \pm 1$ . Of these three transitions, only  $|210\rangle \rightarrow |100\rangle$  satisfies these selection rules. Using the wavefunction from 12.2, and mathematica, and taking  $a_0 = 5.20 \times 10^{-10}$ m and  $e = -1.6 \times 10^{-19}$ C, we find

$$\langle 200|eZ|100\rangle = 0 \tag{23}$$

$$\langle 210|eZ|100\rangle = \int_{0} dr r^{2} \int_{0} d\theta \cos\theta \int_{0} d\phi \psi_{2,1,0}^{*}(r,\theta,\phi)r \cos\theta\psi_{1,0,0}(r,\theta,\phi)$$
  
= 6.305 × 10<sup>-29</sup>Cm (24)

$$\langle 211|eZ|100\rangle = 0 \tag{25}$$

For the magnetic dipole moments, we need  $\mu = \frac{e}{2m_e}L_z$ , so the selection rule is  $m_i = m_f$ . The dipole moment is then  $\mu = \frac{e\hbar}{2m_e}m_\ell$ . This gives zero for all transitions. Note that when spin is included, there will can be non-zero magnetic dipole transitions between these levels.

4. [15 pts] Based on the classical relation E = T + V, where E is the total energy, T is the kinetic energy, and V is the potential energy, what is the probability that the velocity of the relative coordinate exceeds the speed of light for a hydrogen atom in the 1s state? What about the 2s state? Based on these answers, which of the two energy levels would you expect to have a larger relativistic correction?

Using H = T + V and  $T = \frac{1}{2}mv^2$ , we find

$$v = \sqrt{\frac{2}{m}(E - V)}$$

so for the hydrogen system with principle quantum number n this gives

$$v^{2}(r) = \frac{2}{m} \left[ -\frac{\hbar^{2}}{2ma_{0}^{2}} \frac{1}{n^{2}} + \frac{e^{2}}{4\pi\epsilon_{0}r} \right]$$

Setting this equal to  $c^2$  and solving for  $r_c$  gives

$$r_c(n) = \frac{ma_0^2 n^2 e^2}{2\pi\epsilon_0 (m^2 a_0^2 c^2 n^2 + \hbar^2)}$$

with the parameters (from Google)  $m = 9.10 \times 10^{-31}$ kg,  $a_0 = 5.29 \times 10^{-11}$ m,  $e = 1.60 \times 10^{-19}$ C,  $\epsilon_0 = 8.85 \times 10^{-12}$ C<sup>2</sup>N<sup>-1</sup>m<sup>-2</sup>,  $c = 3.00 \times 10^8$ ms<sup>-1</sup>, and  $\hbar = 1.05 \times 10^{-34}$ Js, we find:

For n = 1:  $r_c(1) = 5.62 \times 10^{-15}$  m

For n = 2:  $r_c(2) = 5.62 \times 10^{-15}$  m

So we see that dependence on n is very weak.

The probability to be within this radius, however, depends strongly on n. For n = 1, we have

$$P(r < r_c(1)) = \int_0^{r_c(1)} dr \, R_{10}^2(r) = 4 \int_0^{r_c(1)/a_0} dx \, e^{-2x} x^2 = 8.00 \times 10^{-13}$$

for n = 2 we have

$$P(r < r_c(2)) = \int_0^{r_c(2)} dr \, R_{20}^2(r) = 2 \int_0^{r_c(2)/a_0} dx \, e^{-2x} x^2 (1 - x^2) = 4.00 \times 10^{-13}$$

Therefore we would expect the ground-state to have the larger relativistic correction.

5. [10 pts] Consider the Earth-Moon system as a gravitational analog to the hydrogen atom. What is the effective Bohr radius (give both the formula and the numerical value). Based on the classical energy and angular momentum, estimate the n and m quantum numbers for the relative motion (take the z-axis as perpendicular to the orbital plane).

The Bohr radius for Hydrogen is given by

$$a_0 = \frac{4\pi\epsilon_0\hbar^2}{me^2}$$

From wikipedia I found  $M_M = 7.35 \times 10^{22}$ kg,  $M_E = 5.97 \times 10^{24}$ kg,  $r_M = 3.84 \times 10^8$ m, and  $v_M = 1.022 \times 10^3 \text{ms}^{-1}$ 

To compute the Bohr radius for the moon, we just need to make the substitutions

$$m \to \mu = \frac{M_M M_E}{M_M + M_E} = \frac{7.35 \times 10^{22} \cdot 5.97 \times 10^{24}}{7.35 \times 10^{22} + 5.97 \times 10^{24}} = 7.2610^{22} \text{kg}$$
$$\frac{e^2}{4\pi\epsilon_0} \to GM_M M_E = (6.67 \times 10^{-11})(5.97 \times 10^{24})(7.35 \times 10^{22}) = 2.93 \times 10^{37}$$

This gives

$$a_M = \frac{\hbar^2}{GM_M^2 M_E} = 4.67 \times 10^{-129} \mathrm{m}$$

The classical energy is

$$E = \frac{1}{2}\mu v_E^2 - \frac{GM_M M_E}{r_M} = -3.83 \times 10^{28} \text{J}$$

Solving

$$E = -\frac{\hbar^2}{2\mu a_M^2 n^2}$$

for n gives

$$n = \frac{\hbar}{\sqrt{-2\mu a_M^2 E}} = 2.77 \times 10^{68}$$

To calculate m, we take  $L_z = \mu v_M r_M$  and us

$$m = \frac{L_z}{\hbar} = \frac{\mu v_M r_M}{\hbar} = 2.74 \times 10^{68}$$

Just for fun:

For a transition from n to n-1, the energy released is

$$\Delta E = -\frac{\hbar^2}{2\mu a_M^2} \left[ \frac{1}{n^2} - \frac{1}{(n-1)^2} \right] = -\frac{\hbar^2}{2\mu a_M^2} \frac{(n-1)^2 - n^2}{n^2(n-1)^2} = \frac{\hbar^2}{2\mu a_M^2} \frac{2n-1}{n^2(n-1)^2} \approx \frac{\hbar^2}{2\mu a_M^2} \frac{2}{n^3} \frac{2n-1}{n^2(n-1)^2} \approx \frac{\hbar^2}{2\mu a_M^2} \frac{2n-1}{n^3}$$

This gives a numerical result of  $\Delta E = 2.76 \times 10^{-40}$  J. With  $\lambda = 2\pi\hbar c/\Delta E$  we find  $\lambda = 7.10 \times 10^{14}$  m. Using 1lyr = 9.46 × 10<sup>15</sup> m we find that  $\lambda = 0.075$  light years. The lunar month is 27.21 days, or 0.074 years. Coincidence?