ISP209 Spring 2001	Homework 3
Due: Thurs., Feb. 1, 2:40 pm,	in Rm 118PA.

Name: _____ ID:_____

Homework Problems

1. Which force, electromagnetic(E), gravitational(G), weak nuclear(WN) or strong	
nuclear(SN), is primarily responsible for the following: (all correct for credit)	
fusion of Deuterium and Tritium. SN	<u>pumping</u> water from a well. <u>E</u>
smoking a cigarette <u>E</u>	beta decay of Tritium to Helium-3. WN
<u>bleaching</u> a shirt. <u>E</u>	sawing wood. <u>E</u>
<u>cutting</u> a piece of paper. <u>E</u>	toasting a muffin. <u>E</u>
<u>cooking</u> a chicken. <u>E</u>	defrosting in a microwave oven. <u>E</u>
<u>crushing</u> a nut. <u>E</u>	growing of hair. <u>E</u>

2. True or False (all correct for credit)

- \mathbf{T} F Two force vectors are "equal" only if the magnitude and direction are the same.
- \overline{T} \overline{F} Two force vectors can "balance" if the magnitude and direction are the same.
- $\underline{\mathbf{T}}$ F Two force vectors cannot "balance" and be "equal" at the same time.
- $\overline{\mathbf{T}}$ F Two force vectors with "equal" magnitudes can point in opposite directions.
- $\overline{\mathbf{T}}$ F Two force vectors cannot stretch an object if they are "equal".

3. Convert the pressure 1×10^5 N/m² to a pressure in lb/in² (1 in = 2.54 cm). <u>14 lb/in.²</u>

$$\times 10^5 \text{ N/m}^2 = (1 \times 10^5 \text{ N/m}^2)(0.22 \text{ lb/N})[(2.54 \text{ cm/1 in.})(1 \text{ m/100cm})]^2$$
$$= (0.22 \times 10^5 \text{ lb/m}^2)[.0254 \text{ m/in}]^2 = \underline{14 \text{ lb/in}^2}$$

- 4. What are the units of a spring constant? Force per unit length (force/unit length)
- 5. On the graph below, plot the applied force vs. compression of a spring, spring constant $k=1.5\times10^8$ N/m, for 11 equally spaced compression values from zero (increasing in 10



steps of 0.5×10^{-6} m) to 5×10^{-6} m. It should look similar to text *Fig. 3.5*.

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Show work for all questions.

Attach extra sheets if necessary.

6. A long spring obeys Hooke's law and can be stretched 10 cm by a force of 20 N.

a) Plot of force vs. stretch of this spring has what slope (w/units)? <u>2 N/cm</u>

b) Force applied is 50 N. How far does the spring stretch (w/units)? 25 cm

Plotted on the right, is data for a spring following Hooke's law only after an initial force, F_0 , is applied; $F = kx + F_0$. (Note: the zero of each scale is not shown.)

7. What is the spring constant, *k* (slope of the line), in the region shown? <u>5 N/cm</u>

$$k = \text{slope} = \frac{(70 - 60) \text{ N}}{(12 - 10) \text{ cm}} = 5 \text{ N/cm}$$

8. Use *k*, and one point, *F* and *x*, on the graph to find F_0 , force applied to the spring when it begins to stretch; $F_0 = \underline{10 \text{ N}}$. (assume spring is linear down to x = 0).

$$F = kx + F_0$$

 $F_0 = F - kx = (70 \text{ N}) - (5 \text{ N/cm})(12 \text{ cm}) = 10 \text{ N}$



- 9. In lbs, what force is equal to 1 N? about 1/4 lb
- 10. A force of 21 N, stretches a spring by 7 cm. What is its spring constant? <u>3 N/cm</u>.
- 11. What force (maximum) do you apply to dental floss in use (in lb. & N)? Do not guess, use logic! For example, tell me how many five pound bags of sugar can you can lift with a piece of dental floss. What will the dental floss do to your hand if you tried to lift 20 bags of sugar? maximum force on dental floss = <u>about 10 lb</u>, or <u>40 N</u>
- 12. An additional 10 N of force increases the stretch of a spring from 5 cm to 7 cm.
 - a) What is the spring constant (w/ units) of this spring? 5 N/cm
 - b) What additional force (w/ units) stretches this spring another 3 cm? 15 N

$$F = k \ x = (5 \text{ N/cm})(3 \text{ cm}) = 15 \text{ N}$$

13. A rod has a spring constant $k = 2 \times 10^7$ N/m (note units). How far will it compress (in mm!) under a force of 2×10^3 N? <u>0.1 mm</u>

$$x = \frac{F}{k} = \frac{2 \times 10^3 \text{ N}}{2 \times 10^7 \text{ N/m}} = 1 \times 10^{-4} \text{ m} \left(10^3 \text{ mm/m} \right) = 0.1 \text{ mm}$$

14. The density of lead is 11 g/cm³ (each cubic centimeter of lead has a mass of 11 g). What is the mass (in kg) of 1 m³ of lead? (1 m³ is a cube, 100 cm on a side).

density =
$$(11 \text{ g/cm}^3)(1 \text{ kg/1000 g})(100 \text{ cm/m})^3 = 1.1 \times 10^4 \text{ kg/m}^3$$
 $m = 1.1 \times 10^4 \text{ kg}.$

15. One cubic centimeter of water has a mass of 1 gram ($1g = 10^{-3}$ kg), i.e., the density is $1g/cm^3$. What is the mass of one cubic meter of water?

density =
$$(1 \text{ g/cm}^3)(1 \text{ kg}/1000 \text{ g})(100 \text{ cm/m})^3 = 1 \times 10^3 \text{ kg/m}^3$$
 $m = 1 \times 10^3 \text{ kg}$