

Physics 410 - 2003
Thermal Physics

Problem Set 1

1. Consider a particle of mass m confined in a two-dimensional square box of size L . The potential energy has the form

$$U(x, y) = \begin{cases} 0, & 0 < x, y < L \\ \infty, & \text{otherwise} \end{cases}$$

Find the 8 lowest energy levels and their multiplicities. (5 pt)

2. Derive the expression for the binomial coefficient $\binom{n}{m}$ in the expansion of $(x + y)^n$ (4 pt) [*Hint*: you can use mathematical induction]. Consider a system of N spins. Show that the number of combinations in which there are N_{\uparrow} spins pointing upward and $N_{\downarrow} = N - N_{\uparrow}$ spins pointing downward is given by $\binom{N}{N_{\uparrow}}$ (4 pt)

3. Consider a system of N spins, each of which has equal probability to point upward or downward. Find the probability that N_{\uparrow} of the spins are pointing upward. (4 pt).

4. The probability density for a classical molecule of mass m to have a given velocity v_x along the axis x is

$$p(v_x) = C \exp[-mv_x^2/2k_B T],$$

where T is temperature and k_B is the Boltzmann constant. Find the constant C (3 pt).

Find the probability for a molecule to have the velocity v_x lying between $-(k_B T/m)^{1/2}$ and $4(k_B T/m)^{1/2}$ (5 pt).

You need to have 20 points out of 25 (5 points are extra credit).