

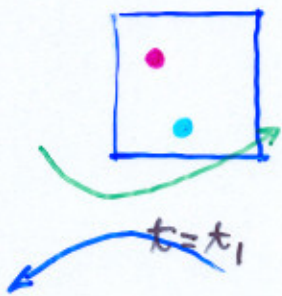
# CHAPTER 8 ATOMIC STRUCTURE & PERIODIC TABLE

So, Hydrogen was "easy" ... closed form

mathematically clean

How about Helium? - there's a story here.

Imagine a classical box with 2 classical balls.



totally tell the difference

classically.

you can easily LABEL and track the trajectories of both through time

Suppose they are quantum mechanical objects.

That's a tangle.

→ their wave functions overlap

⇒ regions where one cannot

distinguish the "phase" of either

The diagram shows two overlapping wave packets. The left packet is labeled with  $P_1$  and  $P_2$  above it and  $k_2$  below it. The right packet is labeled with  $P_1$  and  $P_2$  above it and  $k_1$  and  $k_2$  to its right. A blue wavy line is drawn between the two packets, and a red wavy line is drawn across the right packet.

+ all problems of uncertainty principle. & measurement

Remember my definition of an electron?

$$m = 0.511 \text{ MeV}/c^2$$

$$q = -e$$

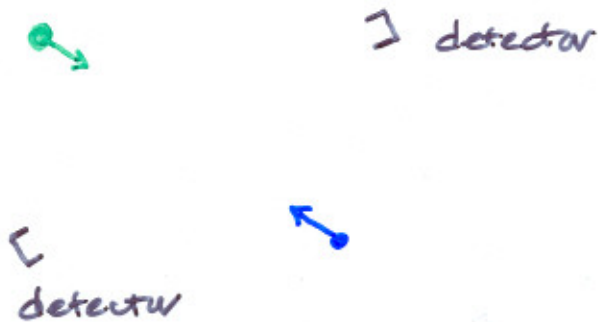
$$s = \frac{1}{2}$$

every object with these properties is an electron -

so every electron is absolutely IDENTICAL.

Consider a classical "scattering experiment" between  
CLASSICAL electrons

can distinguish between them!



"PLACE" becomes really problematic!

→ QM must take INDISTINGUISHABILITY into account.

How about a Schrödinger Equation describing 2

-- say -- electrons?

$\Psi_T$  = wavefunction of the whole system of  
2 electrons having coordinates  $x_1, y_1, z_1, x_2, y_2, z_2$

$V_T$  = potential of system

$E_T$  = energy of system.

$$-\frac{\hbar^2}{2m} \left( \frac{\partial^2 \Psi_T}{\partial x_1^2} + \frac{\partial^2 \Psi_T}{\partial y_1^2} + \frac{\partial^2 \Psi_T}{\partial z_1^2} \right) - \frac{\hbar^2}{2m} \left( \frac{\partial^2 \Psi_T}{\partial x_2^2} + \frac{\partial^2 \Psi_T}{\partial y_2^2} + \frac{\partial^2 \Psi_T}{\partial z_2^2} \right) + V_T(x_1, y_1, z_1, x_2, y_2, z_2) = E_T \Psi_T$$

Can separate the variables...

$$\Psi_T(x_1, y_1, z_1, x_2, y_2, z_2) = \phi(x_1, y_1, z_1) \eta(x_2, y_2, z_2)$$

a function describing  
particle 1 - wavefunction

function  
describing  
particle 2 -  
wavefunction



simplify the notation

$$\phi(x_1, y_1, z_1) \rightarrow \phi(1)$$

$$\eta(x_2, y_2, z_2) \rightarrow \eta(2)$$

So,

$$\textcircled{1} \quad \psi_T = \phi(1) \eta(2)$$

particle 1 in state  $\phi$

particle 2 in state  $\eta$

what if particle 1 was in state  $\eta$  &

particle 2 in state  $\phi$ ?

$$\textcircled{2} \quad \psi_T = \phi(2) \eta(1)$$

... these are different

MEASURABLES...

simplest is just the probability density

$$\textcircled{1} \quad \psi_T^* \psi_T = \phi^*(1) \eta^*(2) \phi(1) \eta(2)$$

$$\textcircled{2} \quad \psi_T^* \psi_T = \phi^*(2) \eta^*(1) \phi(2) \eta(1)$$

crucial point:

If we cannot distinguish between 1 and 2...

then  $\psi_T^* \psi_T$  should not be different  
for switching the labels..

$$\begin{array}{ccc} \phi^*(1) \eta^*(2) \phi(1) \eta(2) & \longrightarrow & \phi^*(2) \eta^*(1) \phi(2) \eta(1) \\ & \begin{array}{c} 1 \rightarrow 2 \\ 2 \rightarrow 1 \end{array} & \end{array}$$

Not the same.

eg.  $\phi^*(1) \Rightarrow \phi^*(x_1, y_1, z_1)$  a particular function  
evaluated at  $\vec{x}_1$

$\phi^*(2) \Rightarrow \phi^*(x_2, y_2, z_2)$  evaluated at  $\vec{x}_2$

There is a way out...

construct wavefunctions such that  $\psi_T^* \psi_T$   
doesn't change under exchange  
of particles.

Look at these...

$$\psi_S \equiv \frac{1}{\sqrt{2}} [\phi(1)\eta(2) + \phi(2)\eta(1)] \quad \text{symmetric}$$

$$\psi_A \equiv \frac{1}{\sqrt{2}} [\phi(1)\eta(2) - \phi(2)\eta(1)] \quad \text{antisymmetric}$$

These are linear combinations of

$$\begin{cases} \psi_T = \phi(1)\eta(2) \\ \psi_T = \phi(2)\eta(1) \end{cases} \quad \text{direct effect } E_T$$

no also solutions to Sch Eq. @ same  $E_T$

How do  $\psi_S$  and  $\psi_A$  behave?

$$\psi_S = \frac{1}{\sqrt{2}} [\phi(1)\eta(2) + \cancel{\phi(2)\eta(1)}] \xrightarrow[\substack{1 \rightarrow 2 \\ 2 \rightarrow 1}]{\eta(1)\phi(2)} \frac{1}{\sqrt{2}} [\phi(2)\eta(1) + \phi(1)\eta(2)] = \psi_S$$

$$\psi_A = \frac{1}{\sqrt{2}} [\phi(1)\eta(2) - \cancel{\phi(2)\eta(1)}] \xrightarrow[\substack{1 \rightarrow 2 \\ 2 \rightarrow 1}]{\eta(1)\phi(2)} \frac{1}{\sqrt{2}} [\phi(2)\eta(1) - \cancel{\phi(1)\eta(2)}] = -\psi_A$$

$$\psi_S^* \psi_S \xrightarrow[\substack{1 \rightarrow 2 \\ 2 \rightarrow 1}]{} \psi_S^* \psi_S \quad \& \quad \psi_A^* \psi_A \xrightarrow[\substack{1 \rightarrow 2 \\ 2 \rightarrow 1}]{} (-1)^2 \psi_A^* \psi_A$$

Wolfgang Pauli in 1925

suggested a solution to both an empirical situation and a logical one.

empirical — state of multi-electron spectra... results

logical — how come all of the electrons in B16 atoms don't collapse into the  $n=1$  state?

Pauli Exclusion Principle:

In multi-electron atoms there can be no more than one electron in the same quantum state.

A state: a wavefunction defined by all quantum numbers so far:

$$n, l, m_l, m_s$$



This isn't right, this isn't even wrong

I don't mind your thinking slowly; I mind your publishing faster than you think

**If I understand Dirac correctly, his meaning is this: there is no God, and Dirac is his Prophet.**

*Wolfgang was a typical night bird. He preferred the town, liked to spend evenings in some café, and would thereafter work on his physics with great intensity and great success. To Sommerfeld's dismay he would therefore rarely attend morning lectures and would not turn up until about noon.*

It was absolutely marvelous working for Wolfgang Pauli. You could ask him anything. There was no worry that he would think a particular question was stupid, since he thought all questions were stupid.

*At the moment physics is again terribly confused. ...I wish I had been a movie comedian or something of the sort and had never heard of physics.*

For quite a while I have set for myself the rule if a theoretician says 'universal' it just means pure nonsense.

You know, what Einstein has just said isn't so stupid.

Einstein wrote, "No one studying this mature, grandly conceived work would believe that the author is a man of twenty-one. One wonders what to admire most, the psychological understanding for the development of the ideas, the sureness of mathematical deduction, the profound physical insight, the capacity for lucid, systematic presentation, the knowledge of the literature, the complete treatment of the subject matter, or the sureness of critical appraisal.



I have done a terrible thing, I have postulated a particle that cannot be detected.





Comment on Heisenberg's radio advertisement<sup>1942</sup>  
"This is to show the world, that I can paint  
like Titian!"



Only technical details are missing.

J. Pauli



How about an ANTISYMMETRIC wave function  
where 1 and 2 are in the same state  
-- say  $\phi$ ?

$$\psi_A = \frac{1}{\sqrt{2}} \left[ \phi(1)\phi(2) - \phi(2)\phi(1) \right] = 0$$

This insures it.

A system containing several electrons must be  
in a totally antisymmetric state.

another statement of  
Pauli E.P.

now... Helium. (for a minute)

ignore the fact that the 2 electrons repel  
one another

$$\psi_A = \frac{1}{\sqrt{2}} \left[ \phi(1)\eta(2) - \eta(1)\phi(2) \right]$$

$\phi$  and  $\eta$  include both the  $n, l, m_l$   
characterizations  $\rightarrow$  functions of  $x, y, z$

also some characterization of the spins. NOT  $x, y, z$ .

total wavefunction = (space wavefunction)  $\times$  (spin w.f.)  
for any particle



$\psi_A$  or  $\psi_S$

Spin is different. not a function of  $x, y, z$  but  
something that depends on  $m_s = \pm 1/2$

Often use little cartoons...

$m_s = +1/2$	$\uparrow$	spin "up"
$m_s = -1/2$	$\downarrow$	spin "down"

The total spin state of the 2-electron system

$$\vec{S} = \vec{S}_1 + \vec{S}_2$$

length:  $S = \hbar \sqrt{s(s+1)}$

projection:  $S_z = m_s \hbar$

possible values:  $m_s = -s \dots +s$

what's  $s$ ? a combination of  $s_1$  and  $s_2$

$$s_1 = \uparrow \quad \text{or} \quad \downarrow$$

$$s_2 = \uparrow \quad \text{or} \quad \downarrow$$

can combine like

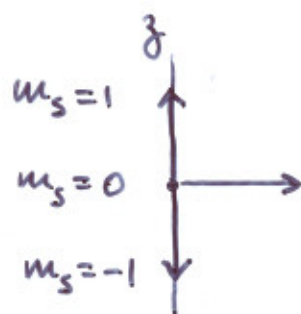
$$\begin{array}{c} s_2 \\ s_1 \end{array} \begin{array}{c} \uparrow \\ \uparrow \end{array} \quad \text{or} \quad \begin{array}{c} s_1 \\ s_2 \end{array} \begin{array}{c} \uparrow \\ \uparrow \end{array} \Rightarrow \left. \begin{array}{c} \uparrow \\ \uparrow \end{array} \right\} = 1$$

or

$$s_1 \uparrow \downarrow s_2 \quad \text{or} \quad s_2 \uparrow \downarrow s_1 \Rightarrow \cdot \} = 0$$

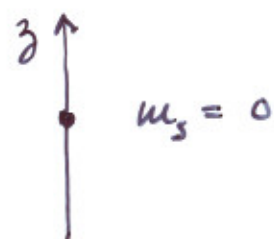
So,  $s$  can be either 0 or 1

When  $s = 1$



$$s_1, s_2: \begin{array}{c} \uparrow \frac{1}{2} \\ \uparrow \frac{1}{2} \end{array}$$

When  $s = 0$



$$s_1, s_2: \frac{1}{2} \downarrow \uparrow \frac{1}{2}$$

forget continuous functions of  $x, y, z$  -- can  
still write SPIN WAVEFUNCTIONS

$$\Sigma = (\uparrow, \downarrow) \quad (m_{s_1}, m_{s_2})$$

$$m_{s_1} = \frac{1}{2}$$

$$m_{s_2} = -\frac{1}{2}$$

Can also form  $A$  &  $S$

$$\Sigma_S = \frac{1}{\sqrt{2}} [ (\uparrow, \uparrow) ]$$

$$\Sigma_S = \frac{1}{\sqrt{2}} [ (\uparrow, \downarrow) + (\downarrow, \uparrow) ]$$

$$\Sigma_S = [ (\downarrow, \downarrow) ]$$

} a "triplet"  
three symmetric  
states

symmetric in the exchange of  $m_{s_1} \rightarrow m_{s_2}$

$m_{s_2} \rightarrow m_{s_1}$

and:

$$\Sigma_A = \frac{1}{\sqrt{2}} [ (\uparrow, \downarrow) - (\downarrow, \uparrow) ]$$

} a singlet

one antisymmetric  
state