

# PHYSICS 215 - Thermodynamics and Modern Physics

## Practice Midterm Exam 2

Fall 2007

1. [6 points] An interstellar spaceship travels from the earth to a distant star 12 light years away (as measured in the earth's frame). The trip takes 15 years as measured on the ship.
  - (a) What is the speed of the ship relative to the earth?
  - (b) When the ship arrives at its destination, it sends a radio signal back to the earth. How much time elapses on earth between the departure of the spaceship and the arrival of the signal?
  
2. [6 points] A beam of X rays has an energy of 10 keV.
  - (a) What is the wavelength of these X rays?
  - (b) The X rays are then Compton-scattered from electrons at rest. The scattered beam is observed at an angle of  $90^\circ$  relative to the incident beam. What is the wavelength of the scattered X rays?
  - (c) What is the energy of the scattered photons?
  - (d) What is the kinetic energy of the scattered electrons

[Extra credit, 2 points] What is the angle of the scattered electrons?
  
3. [6 points] In an experiment done by scattering  $\alpha$  particles on a thin gold foil, students find that 10,000  $\alpha$  particles are scattered at an angle of  $\theta = 45^\circ$ .
  - (a) How many  $\alpha$  particles would they expect to see scattered at  $\theta = 60^\circ$ ?
  - (b) If the gold foil ( $Z = 79$ ) is replaced by an aluminum foil ( $Z = 13$ ) with a similar number of scattering nuclei per unit area, how many  $\alpha$  particles will now be scattered at  $45^\circ$ ?
  
4. [6 points] An electron, initially at rest, is accelerated across a potential difference of 5 kV.
  - (a) What is its kinetic energy?
  - (b) What is its total energy??
  - (c) What is its momentum?
  - (d) What is its de Broglie wavelength?

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**Useful Constants:** Avogadro's number,  $N_A = 6.02E23 \text{ mol}^{-1}$   
Speed of light,  $c = 3.00E8 \text{ m/s}$   
Charge of an electron,  $-e = -1.6E-19 \text{ C}$   
Mass of the electron,  $m_e = 9.1E-31 \text{ kg} = 511 \text{ keV}/c^2$   
Mass of the proton,  $m_p = 1.67E-27 \text{ kg} = 938 \text{ MeV}/c^2$   
Planck's constant,  $h = 6.63E-34 \text{ J.s} = 4.14E-15 \text{ eV.s}$   
Planck's reduced constant,  $\hbar = h/2\pi = 1.05E-34 \text{ J.s} = 6.58E-16 \text{ eV.s}$   
Compton Wavelength of the electron,  $\lambda_c = h/m_e c = 2.4263E-12 \text{ m}$

## Useful Formulae:

Potential energy lost by a charge  $q$  in traversing a potential difference of  $V$  is  $U = qV$

Wave relation:  $v = \nu\lambda$  where  $v$  = velocity,  $\nu$  = frequency,  $\lambda$  = wavelength.

$$\beta = v/c \quad \gamma = 1/\sqrt{(1 - \beta^2)}$$

Length Contraction:  $L' = L/\gamma$  Time Dilation,  $T' = \gamma T$

Addition of Velocities:  $v' = (v + u)/(1 + vu/c^2)$

Relativistic Doppler Effect:  $\nu' = \frac{\nu\sqrt{(1 - \beta)}}{\sqrt{(1 + \beta)}}$

Momentum – Energy relations:  $E^2 = p^2 c^2 + m^2 c^4$   
 $E = \gamma m c^2$        $p = \gamma m v$        $K = E - m c^2$

Planck's Relation:  $E = h\nu$

Einstein's Photoelectric Law:  $h\nu = K + \phi$

Compton Effect:  $\Delta\lambda = \lambda' - \lambda = (1 - \cos\theta)h/m_e c$

Bohr Quantization Relation:  $L = mvr = n\hbar$

Atomic Radii:  $r_n = n^2 a_0 / Z$  Atomic Energies:  $E_n = -Z^2 E_0 / n^2$   
where  $a_0 = 5.29E-11 \text{ m}$  where  $E_0 = 13.6 \text{ eV}$

Rutherford Scattering:  $N(\theta) = \frac{N_i n t e^4 Z_1^2 Z_2^2}{16 (4\pi\epsilon_0)^2 r^2 K^2 \sin^4(\theta/2)}$

de Broglie Wavelength:  $\lambda = h/p$