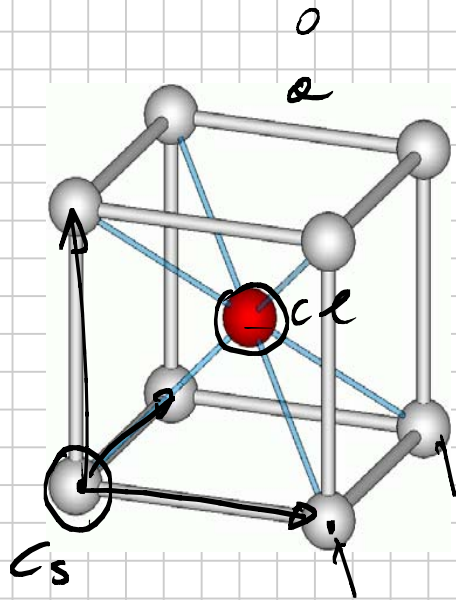


Lecture # 11

Note Title

10/15/2007



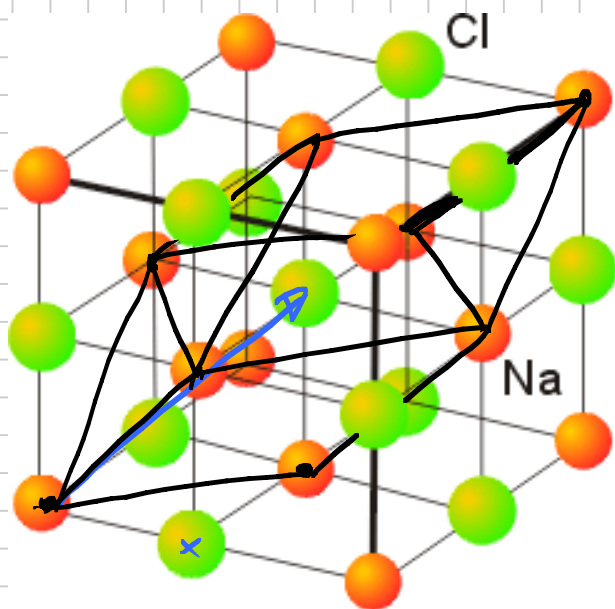
Not a BCC!
Simple cubic

$$\begin{aligned} \vec{a}_1 &= a(100) \\ \vec{a}_2 &= a(010) \\ \vec{a}_3 &= a(001) \end{aligned}$$

Volume of unit cell
 $= a^3$

$$\left. \begin{aligned} \vec{b}_{Cs} &= (0,0,0) \\ \vec{b}_{ce} &= \frac{a}{2}(111) \end{aligned} \right\}$$

Density: $\frac{2 \text{ atoms}}{a^3}$



Not a SC

Look at Na \rightarrow FCC

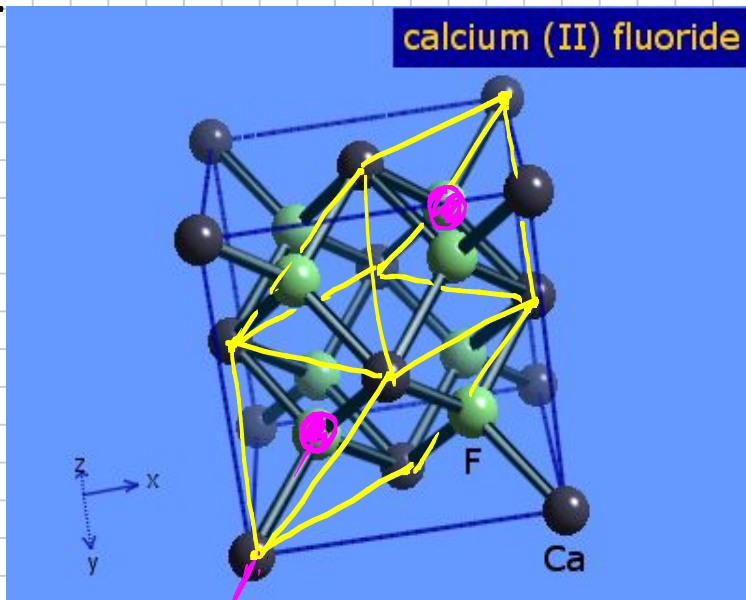
$$\begin{aligned} \vec{a}_1 &= \frac{a}{2}(110) \\ \vec{a}_2 &= \frac{a}{2}(011) \\ \vec{a}_3 &= \frac{a}{2}(101) \end{aligned}$$

Volume = $\frac{a^3}{4}$

$$n = \frac{2}{a^3/4} = \frac{8}{a^3}$$

$$\left. \begin{aligned} \vec{b}_{Na} &= (0,0,0) \rightarrow \text{FCC} \\ \vec{b}_{ce} &= \frac{a}{2}(111) \rightarrow \text{FCC} \end{aligned} \right\}$$

① Ca make an FCC



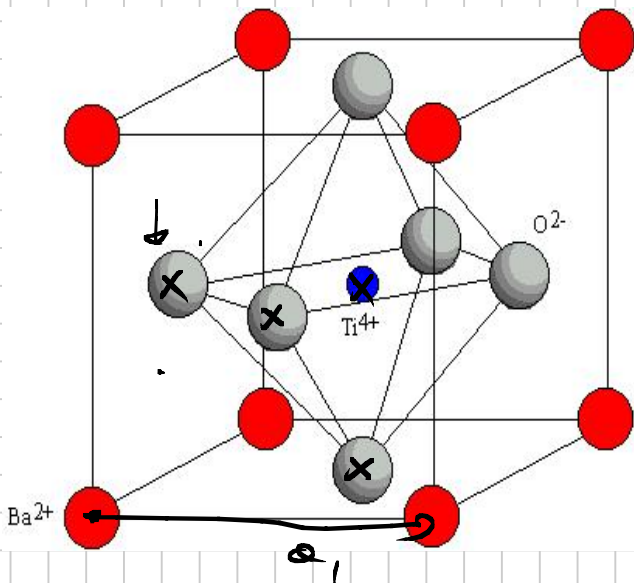
$$\begin{aligned} a_1 &= \frac{a}{2} (110) \\ a_2 &= \frac{a}{2} (011) \\ a_3 &= \frac{a}{2} (101) \end{aligned}$$

$$\vec{b}_{Ca} = (0, 0, 0)$$

$$\vec{b}_F = \frac{a}{4} (111)$$

$$\vec{b}_P = \frac{3a}{4} (111)$$

$$\frac{3 \text{ atoms}}{a^3/4}$$



Simple Cubic

$$b_{Ba} = (000)$$

$$b_o = \frac{a}{2} (110)$$

$$b_o = \frac{a}{2} (101)$$

$$b_o = \frac{a}{2} (011)$$

$$b_{Ti} = \frac{a}{2} (111)$$

5 Atoms

$$\frac{5 \text{ Atoms}}{a^3}$$

Hexagonal lattice

HCP crystal structure

RECIPROCAL

LATTICE

$$\{ \vec{R}_i \}$$

Breuer's



$$\{ \vec{K}_j \}$$

Breuer's

$$e^{i \vec{K}_j \cdot \vec{R}_i} = 1$$

$\vec{K}_i \rightarrow$ wave with
same symmetry
of lattice

$$\vec{R}_i = (m_1 \vec{a}_1 + m_2 \vec{a}_2 + m_3 \vec{a}_3)$$

$$\vec{K}_j = (m_1 \vec{b}_1 + m_2 \vec{b}_2 + m_3 \vec{b}_3)$$

$$\boxed{\vec{b}_1 \quad \vec{b}_2 \quad \vec{b}_3}$$

$$\vec{R}_i \cdot \vec{K}_j = 2\pi k$$

$$\left(m_1 \vec{a}_1 + m_2 \vec{a}_2 + m_3 \vec{a}_3 \right) \cdot \left(m_1 \vec{b}_1 + m_2 \vec{b}_2 + m_3 \vec{b}_3 \right) = 2\pi k$$

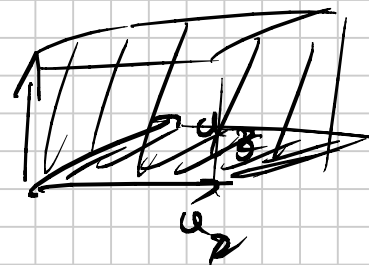
$$\left\{ \begin{array}{l} \vec{b}_1 \perp \vec{a}_2 \text{ and } \vec{a}_3 \Rightarrow \vec{b}_1 = c (\vec{a}_2 \times \vec{a}_3) \\ \vec{b}_1 \cdot \vec{a}_1 = 2\pi \Rightarrow c \vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3) = 2\pi \end{array} \right.$$

$$\vec{b}_1 = \frac{2\pi (\vec{a}_2 \times \vec{a}_3)}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)}$$

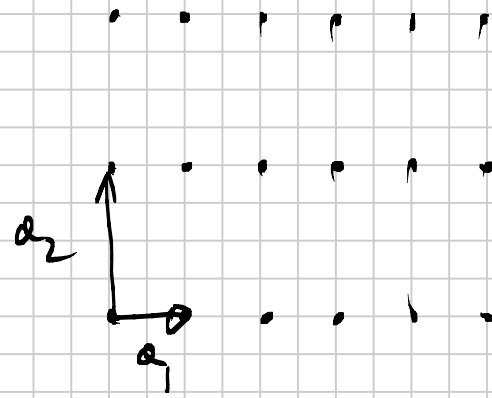
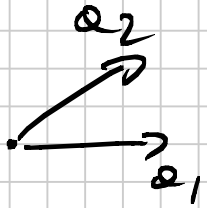
$$\vec{b}_i = \frac{2\pi (\vec{a}_j \times \vec{a}_k)}{\vec{a}_i \cdot (\vec{a}_2 \times \vec{a}_3)}$$

$$ijk \in \{1, 2, 3\}$$

$$\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3) = \text{Volumen umittel cell}$$



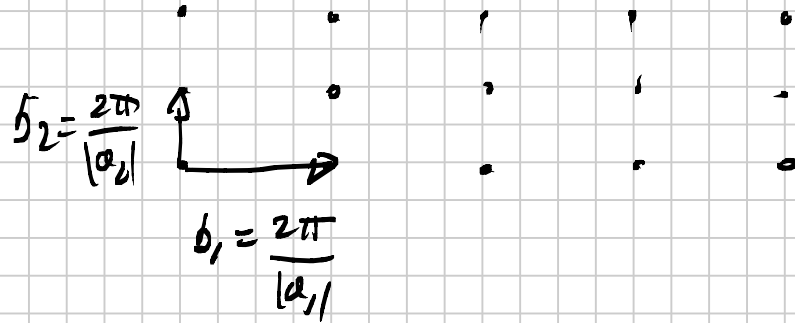
For a 2D lattice (add \vec{e}_3 as \vec{e}_3)



Direct

$$b_1 \cdot a_1 = 2\pi$$

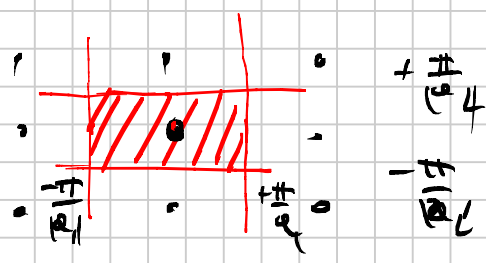
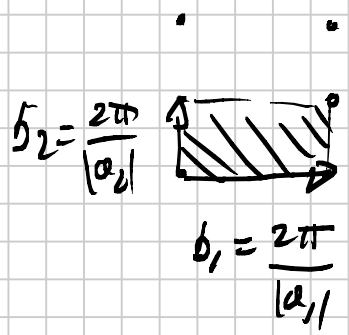
$$b_2 \cdot a_2 = 2\pi$$



IN 3D

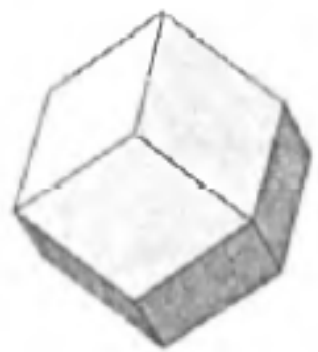
S Cubic	a	\longrightarrow	S Cubic	$\frac{2\pi}{a}$	
FCC	a	\longrightarrow	BCC	$\frac{4\pi}{a}$	
BCC	a	\longrightarrow	FCC	$\frac{4\pi}{a}$	
HEX	a	\longrightarrow	HEX	$\frac{2\pi\sqrt{4}}{a \cdot 3}$	

Unit Cell of Reciprocal Lattice



WIGNER-SEITZ
UNIT CELL OF
RECIPROCAL LATTICE

↓
1^o BRILLOUIN ZONE

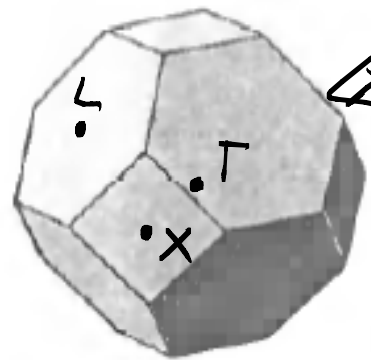


(a)

DODECAHEDRON

REC FCC

↓
DIRECT BCC



(b)

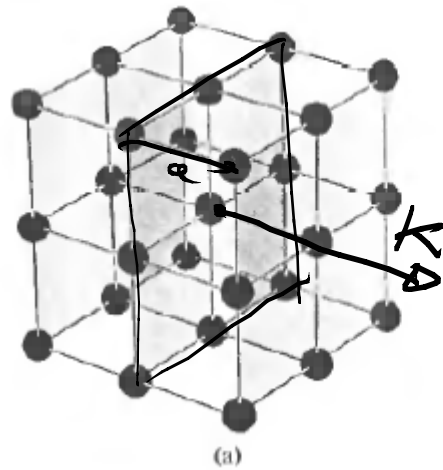
OCTAHEDRON

DIRECT = FCC



RECIPROCAL BCC

LINK BETWEEN \vec{K} AND FAMILY OF PLANES IN A BRAVAIS LATTICE



$$\vec{K} \perp (111)$$

$\vec{K} \perp$ TO FAMILY
OF PLANES

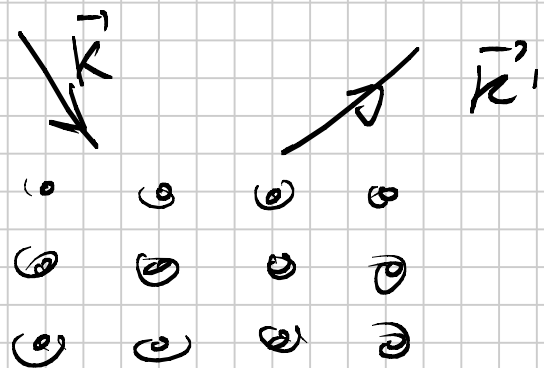
$$|\vec{K}| = \frac{2\pi}{d} \quad d \text{ distance between planes}$$

Reciprocal lattice points

important in Scattering

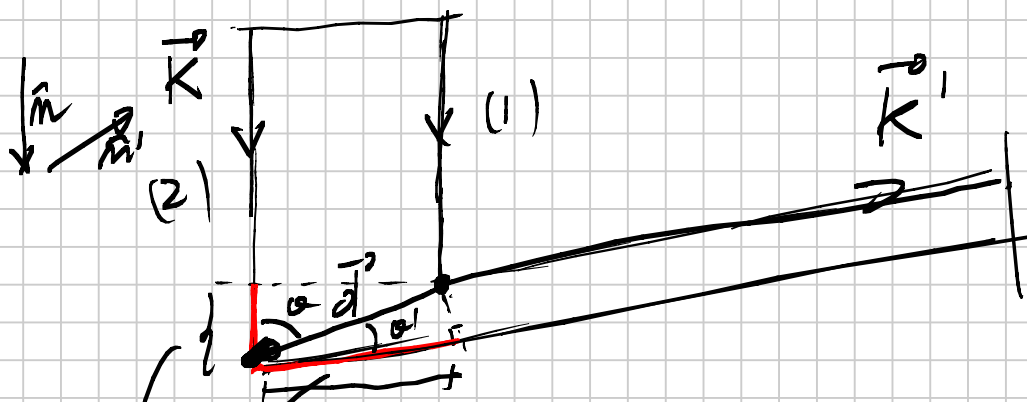
K_i :

X-RAYS



Look first at 2 atoms

Difference in paths
 $\delta = m\lambda$



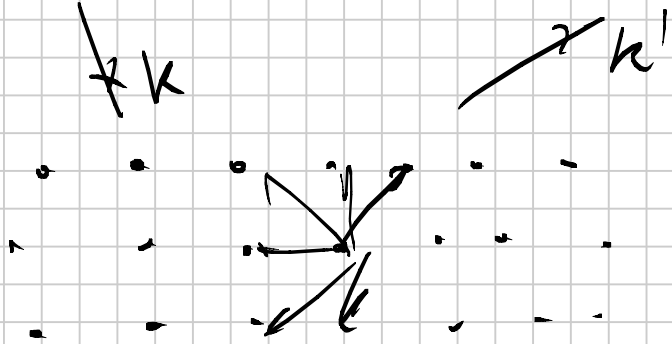
$$d \cos \theta = \vec{d} \cdot \hat{n}$$

$$d \cos \theta' = -\vec{d} \cdot \hat{n}'$$

$$\delta = \vec{d} \cdot (\hat{n} - \hat{n}') = m\lambda$$

$$\frac{2\pi}{\lambda}$$

$$|\vec{k}| = \frac{2\pi}{\lambda} = |\vec{k}'| \Rightarrow \boxed{\vec{d} \cdot (\vec{k} - \vec{k}') = 2\pi m}$$



$$\boxed{\vec{R}_i \cdot (\vec{k} - \vec{k}') = 2\pi m \quad \forall R_i}$$

\Downarrow
 $(\vec{k} - \vec{k}') = \vec{K}$ is a vector of
 RECIPROCAL LATTICE

Chap 4 Chap 7
 Chap 5

}
Ashcroft
Mermin