
PHY492: Nuclear & Particle Physics

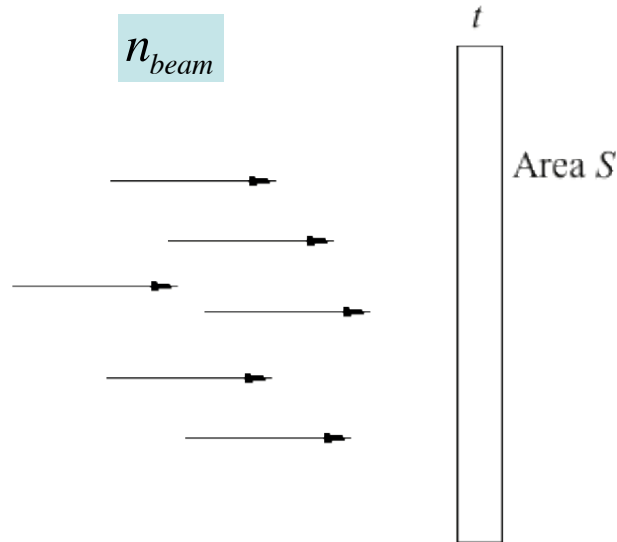
Lecture 3

Homework 1

Nuclear Phenomenology

Measuring cross sections in thin targets

beam particles/s



scattered particles
detected per second

$$m_T = \rho t S$$

mass of target

$$n_{moles} = \frac{m_T}{A}$$

of target moles
atomic mass A

$$n_{nuclei} = n_{moles} A_0$$

target nuclei

$$\frac{n_{nuclei}}{S}$$

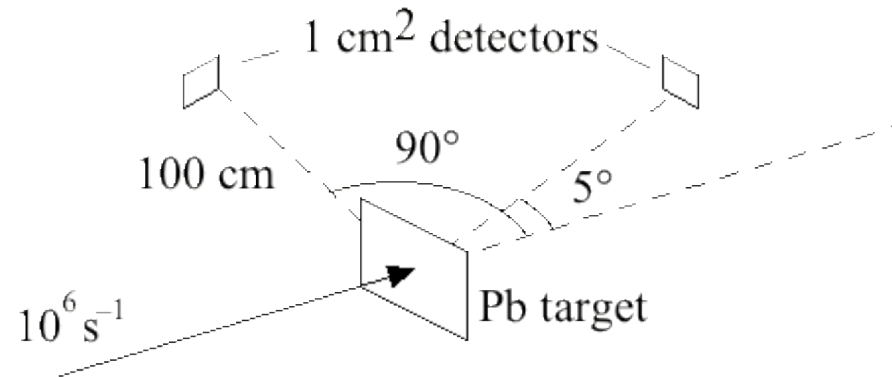
target nuclei
per unit area

$$n_{detected} = n_{beam} \frac{n_{nuclei}}{S} \frac{d\sigma}{d\Omega}(\theta, \phi) d\Omega_{detector}$$

Problem 1.4

Rutherford scattering of 10 MeV α particles on Lead foil, 0.1 cm thick

What is the counting rate at 90° in the lab, 100 cm away, in a counter 1 cm^2 if there are 10^6 α particles/s hitting the target?



S = area of target is not given?

$$\begin{aligned} dn &= N_0 \frac{N}{S} \frac{d\sigma}{d\Omega}(\theta, \phi) d\Omega \\ &= N_0 A_0 \frac{\rho t}{A} \frac{d\sigma}{d\Omega}(\theta, \phi) d\Omega \end{aligned}$$

$$\rho = \text{mass/cm}^3$$

$$\rho t = \text{mass/cm}^2$$

$$\frac{\rho t}{A} = \text{moles/cm}^2$$

$$A_0 \frac{\rho t}{A} = \text{atoms/cm}^2$$

$$\frac{d\sigma}{d\Omega}(\theta) = \left(\frac{ZZ'e^2}{4E} \right)^2 \sin^{-4}(\theta/2)$$

$$dn = 0.5 \text{ s}^{-1}$$

What about at 5° ?

$$dn = 34,000 \text{ s}^{-1}$$

$$\begin{aligned} \frac{ZZ'e^2}{4E} &= \frac{ZZ'e^2 (\hbar c)}{4E (\hbar c)} = \frac{ZZ'}{4E} \alpha \hbar c \\ &= \frac{ZZ'}{4E} \frac{197 \text{ MeV} \cdot \text{fm}}{137} \end{aligned}$$

Problem 1.1: Rutherford scattering total cross section

Differential
Cross section

$$\frac{d\sigma}{d\Omega}(\theta) = \left(\frac{ZZ'e^2}{4E} \right)^2 \frac{1}{\sin^4 \frac{\theta}{2}}$$

Total
Cross section

$$\sigma_{total} = \int d\Omega \frac{d\sigma}{d\Omega}(\theta)$$

$$d\Omega = \sin \theta d\theta d\phi$$

in D&F Eq. 1.32

$$\sin^2 \frac{\theta_b}{2} = \frac{1}{1 + \frac{4b^2 E^2}{(ZZ'e^2)^2}}$$

Scattering at angles larger than θ_b

$$\sigma_{TOT}(\theta > \theta_b) = 8\pi \left(\frac{ZZ'e^2}{4E} \right)^2 \int_{\sin^2 \frac{\theta_b}{2}}^1 \frac{dx}{x^3}$$

$$\begin{aligned} &= 4\pi \left(\frac{ZZ'e^2}{4E} \right)^2 \left(\frac{1}{\sin^2 \frac{\theta_b}{2}} - 1 \right) = 4\pi \left(\frac{ZZ'e^2}{4E} \right)^2 \frac{4b^2 E^2}{(ZZ'e^2)^2} \\ &= 4\pi \left(\frac{ZZ'e^2}{4E} \right)^2 \left(\frac{2E}{ZZ'e^2} \right)^2 b^2 = \pi b^2 \end{aligned}$$

Area of disk radius b

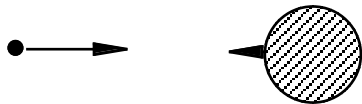
Problem 1.3: N.R. two body scattering

$$\zeta = \frac{m_{\text{beam}}}{m_{\text{target}}}$$

$\zeta = 0.05$

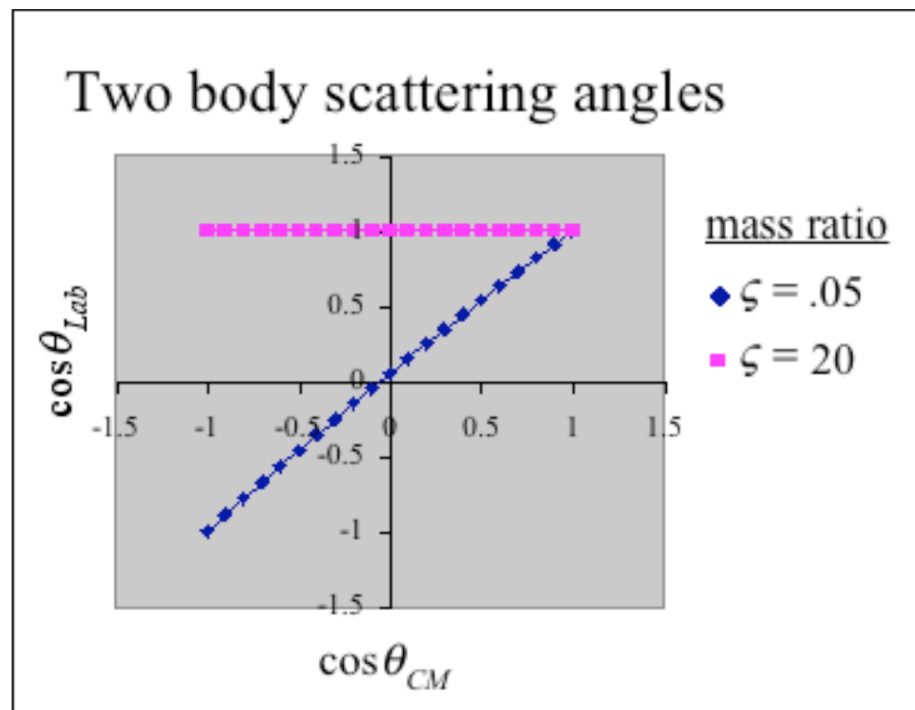


Target stationary
(laboratory frame)



Net zero momentum
(center of mass frame)

$\zeta = 20$



Problem 1.5: Isotropic Center of Mass scattering

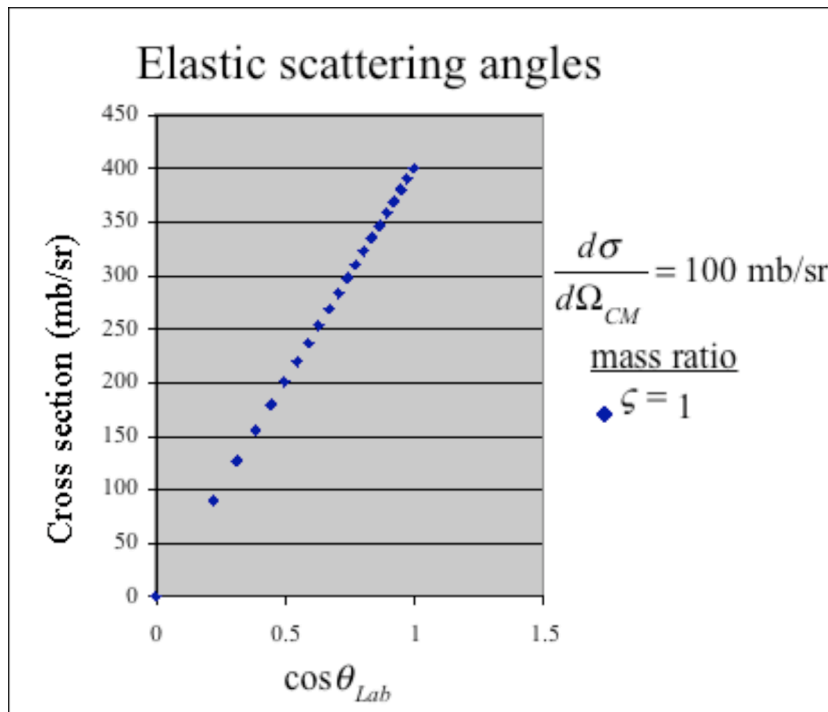
Angle transformation

$$\cos\theta_{\text{Lab}} = \frac{\cos\theta_{\text{CM}} + \zeta}{(1 + 2\zeta\cos\theta_{\text{CM}} + \zeta^2)^{1/2}}$$

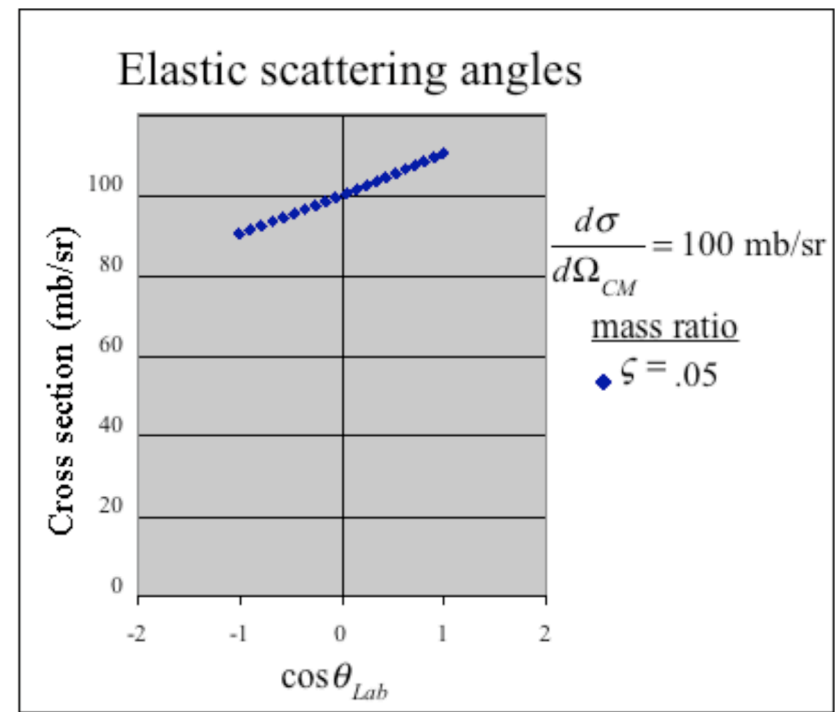
Cross section transformation

$$\frac{d\sigma}{d\Omega_{\text{Lab}}} = \frac{d\sigma}{d\Omega_{\text{CM}}} \frac{(1 + 2\zeta\cos\theta_{\text{CM}} + \zeta^2)^{3/2}}{|1 + \zeta\cos\theta_{\text{CM}}|}$$

No backward scattering



Lab scattering nearly isotropic



Problem 1.7: Relativistic electron

Electron mass $m = 0.511 \text{ MeV}/c^2$

Given electron momentum $pc = mc^2$

Total energy $E = \sqrt{p^2 c^2 + m^2 c^4} = \sqrt{2} mc^2$

Velocity $\beta = \frac{pc}{E} = \frac{1}{\sqrt{2}}$

Gamma $\gamma = (1 - \beta^2)^{-\frac{1}{2}} = \sqrt{2}$

Kinetic energy $T = E - mc^2 = (\sqrt{2} - 1) mc^2$

Problem 1.9: Ultra-relativistic limit $\tilde{\beta} \rightarrow 1$

Ultra-relativistic limit

$$\tan \theta_{Lab} = \frac{\tilde{\beta} \sin \theta_{CM}}{\gamma_{CM} (\tilde{\beta} \cos \theta_{CM} + \beta_{CM})} \rightarrow \frac{\sin \theta_{CM}}{\gamma_{CM} (\cos \theta_{CM} + 1)}$$

Maximum at

$$\theta_{CM} = \frac{\pi}{2}$$

$$\tan \theta_{Lab} = \frac{1}{\gamma_{CM}}$$

$$\begin{aligned} \theta_{Lab} &= \tan^{-1} \left(\frac{1}{\gamma_{CM}} \right) \\ &= \tan^{-1} (.1) = 5.7^\circ; \quad \gamma_{CM} = 10 \\ &= \tan^{-1} (.01) = 0.57^\circ; \quad \gamma_{CM} = 100 \end{aligned}$$

Problem 1.11: Scattering rates into a detector

Alpha T=8 MeV, flux = 10^4 s^{-1} , 0.1 mm Gold foil, $\rho = 19.3 \text{ g/cm}^3$

$$\Delta\theta = .05 \text{ rad}, \theta = 90^\circ = \pi \frac{90}{180} \text{ rad} = 1.57 \text{ rad} \text{ (note: } d\theta = \Delta\theta \text{ if } \frac{\Delta\theta}{\theta} \ll 1 \text{ is satisfied!)}$$

$$\Delta\theta = .05 \text{ rad}, \theta = 5^\circ = \pi \frac{5}{180} \text{ rad} = 0.087 \text{ rad} \text{ (} d\theta \neq \Delta\theta \text{ because } \frac{\Delta\theta}{\theta} \ll 1 \text{ not satisfied!)}$$

$$\text{at } 90^\circ \quad \frac{d\sigma}{d\Omega}(\theta) = \left(\frac{ZZ'e^2}{4E} \right)^2 \frac{1}{\sin^4 \frac{\theta}{2}} = 2.02 \text{ b/sr}; \quad \text{rate} = 3.5/\text{s}$$

$$\text{at } 5^\circ \quad \frac{d\sigma}{d\Omega}(\theta) = 21,000/\text{s} \text{ is greater than beam rate!}$$

but $d\theta$ is not correct!

Nuclear Numerology

Nuclei are specified by three numbers A , Z , and N , and are labeled by the element X , in the periodic table with that Z :

Z = # of protons (atomic number)

N = # of neutrons (determines the isotope)

A_ZX or ${}^AX^Z$ or AX_Z

$A = Z + N$ (atomic weight, specifies the isotope)

X = Mnemonic for element in periodic table

For example,
Calcium 40 :

${}^{40}\text{Ca}$

Z is not specified except by looking in
a Periodic Table of Elements

${}^{40}_{20}\text{Ca}$

$Z = 20$, and therefore $N = A - Z = 20$

Neutron deficit

${}^{34}\text{Ca}$

Stable isotopes

${}^A\text{Ca}$, $A = 40, 42, 43, 44, 46, 48$

Neutron excess

${}^{57}\text{Ca}$

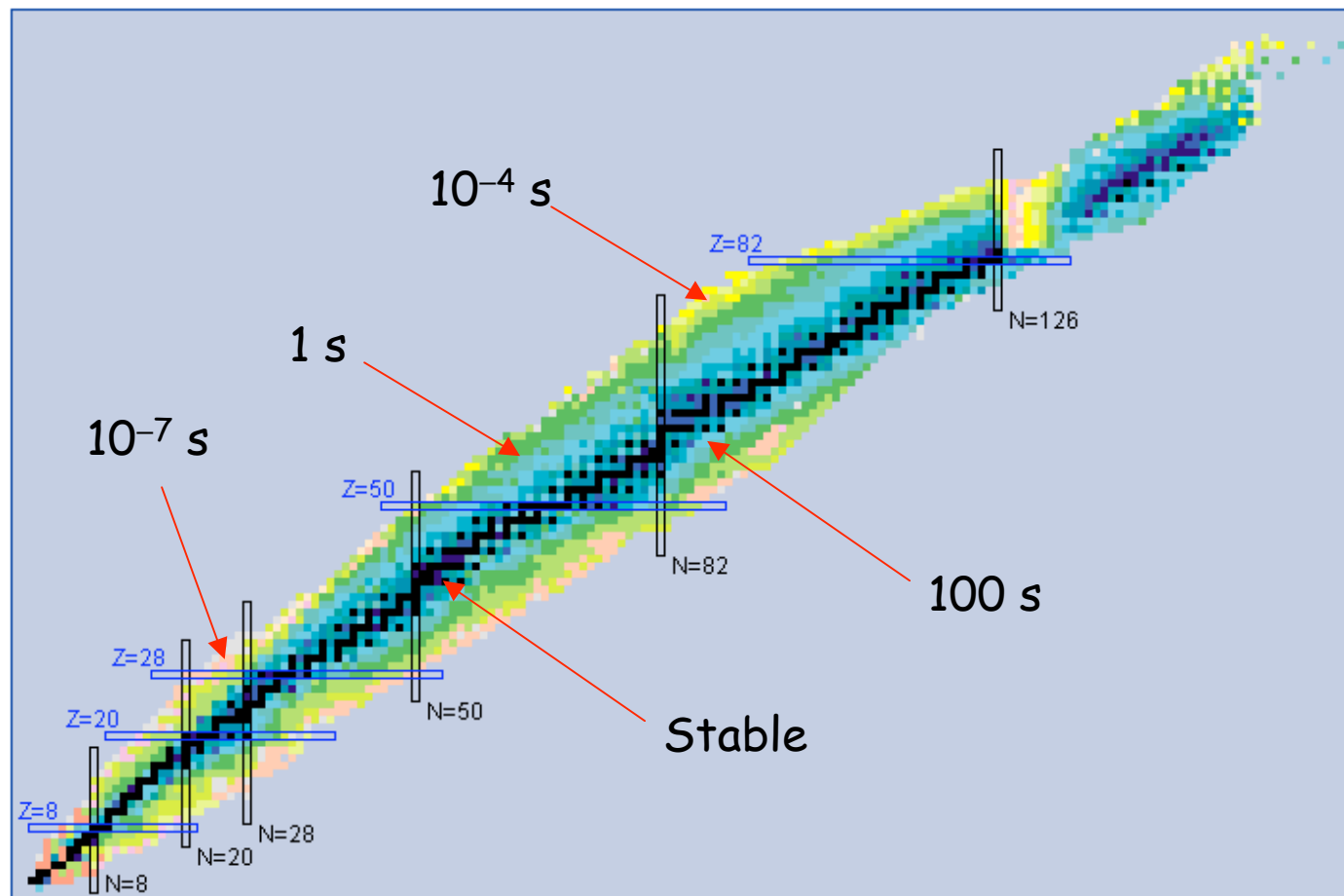
Alice's Restaurant

<http://www.nndc.bnl.gov/>

For binding energies

<http://t2.lanl.gov/data/map.html>

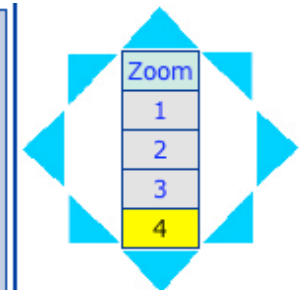
Half Lives (www.nndc.bnl.gov)



Ground and isomeric state information for $^{46}_{20}\text{Ca}$

E(level) (MeV)	J π	Δ (MeV)	$T_{1/2}$	Abundance	Decay Modes
0.0	0+	-43.1349	> 0.28E+16 y	0.004% 3	2 β^-

A [list of levels](#) and a [level scheme](#) are available



Nucleus: go

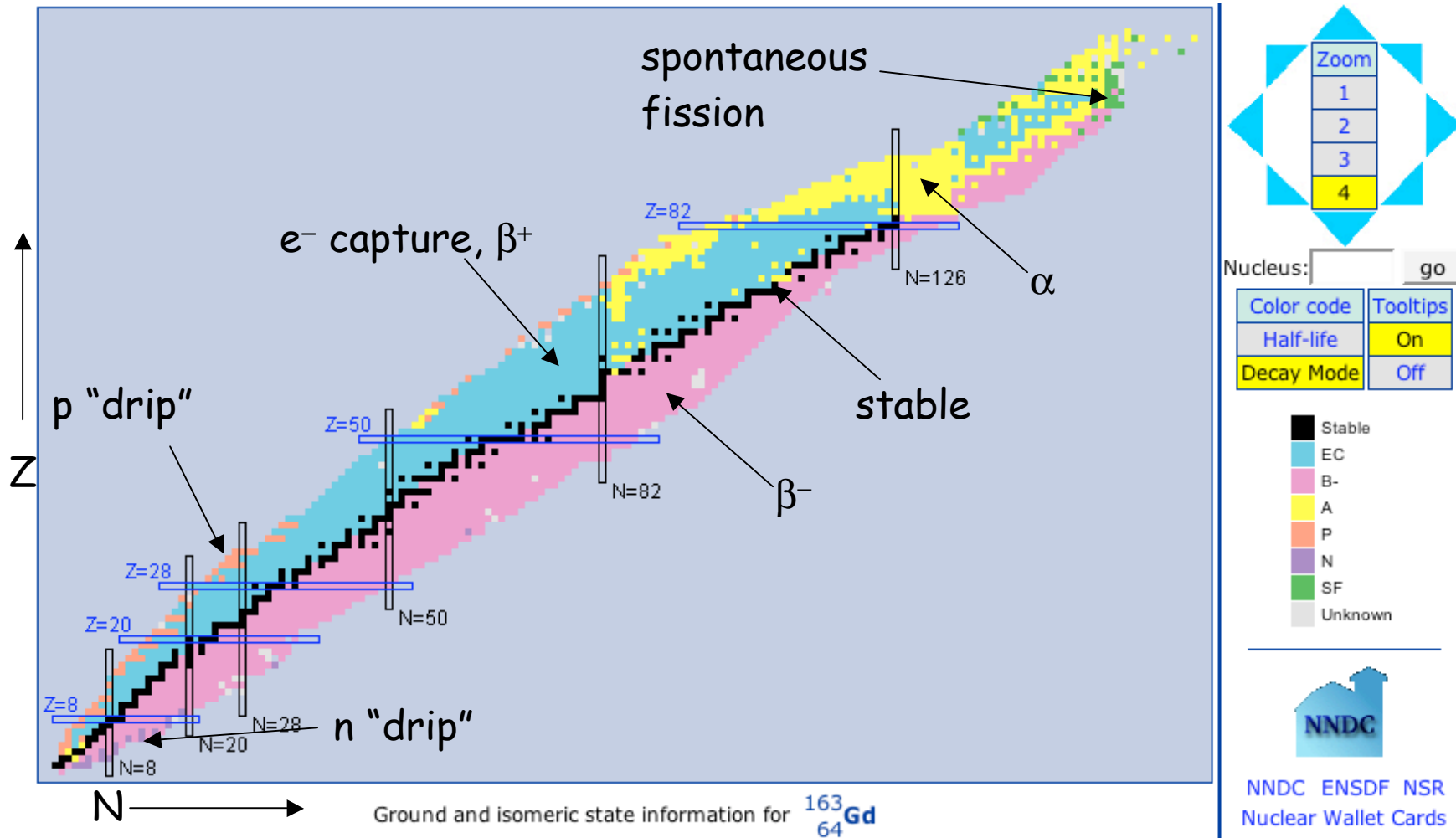
Color code	Tooltips
Half-life	On
Decay Mode	Off

> 10+15 s	10-01 s
10+10 s	10-02 s
10+07 s	10-03 s
10+05 s	10-04 s
10+04 s	10-05 s
10+03 s	10-06 s
10+02 s	10-07 s
10+01 s	10-15 s
10+00 s	< 10-15 s
unknown	



NNDC ENSDF NSR
Nuclear Wallet Cards

Decay modes (www.nndc.bnl.gov)



E(level) (MeV)	J π	Δ (MeV)	T _{1/2}	Decay Modes
0.0	(5/2 ⁻ , 7/2 ⁺)	-61.4880 Syst	68 s 3	β^- : 100.00 %

A list of levels, a level scheme and decay radiation information are available

NNDc ENSDF NSR
Nuclear Wallet Cards

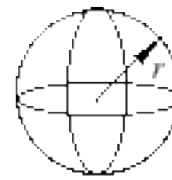
Physics of nuclei

Topics to be covered

- size and shape →
- mass and binding energy
- charge distribution
- angular momentum (spin)
- symmetries (parity)
- magnetic moments
- radioactivity
- energy levels
- reactions

Nuclear size

Binding energy per nucleon is < 1% of the nucleon mass. The protons and neutrons in the nucleus retain their particle properties. Assume nuclei are spherical and have a constant density (not compressed).



$$\left. \begin{array}{l} V \sim r^3 \\ V \sim A \end{array} \right\} r \sim A^{\frac{1}{3}}$$

Nuclear radius

$$r = (1.2 \times 10^{-15} \text{ m}) A^{\frac{1}{3}}$$