

# LECTURE # 8

Note Title

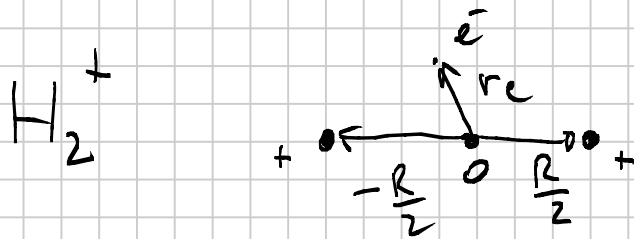
9/22/2008

LAST TIME: MOLECULES

BORN - OPPENHEIMER

APPROX

PARAMETER



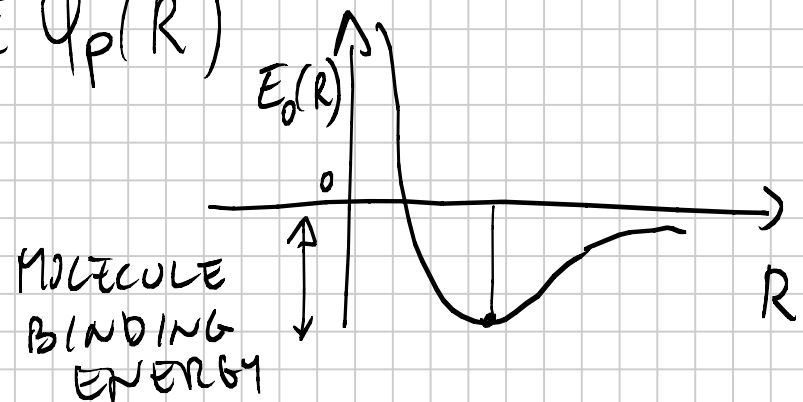
$$\Psi(r_e, R) \sim \Psi_p(R) \Psi_e^R(r_e)$$

① "CLAMP" PROTONS AT DISTANCE  $R \rightarrow \forall R$ , SOLVE SE FOR  $e^- \rightarrow \Psi_{e,m}^R, E_m(R)$

② Eq. PROTONS:

$$\left( -\frac{\hbar^2 \nabla_R^2}{2\mu_p} + E_m(R) \right) \Psi_p(R) = E \Psi_p(R)$$

HOW DO YOU FIND  $E_m(R)$ ?



$$H = -\frac{\hbar^2}{2m} \nabla_{\vec{r}}^2 - \frac{e^2}{|\vec{r} + \frac{\vec{R}}{2}|} - \frac{e^2}{|\vec{r} - \frac{\vec{R}}{2}|} \quad r = r_e$$

MOLECULAR ORBITALS AS LINEAR COMBINATION OF ATOMIC ORBITALS

MO - LCAO

⊕  $\psi_{1s}(\vec{r} + \frac{\vec{R}}{2}) \equiv a(\vec{r}) = 1s \text{ HYDROGEN}$

⊕  $\psi_{1s}(\vec{r} - \frac{\vec{R}}{2}) \equiv b(\vec{r})$  "

$$\psi_{MO}(\vec{r}) = c_a a(\vec{r}) + c_b b(\vec{r})$$

$$|c_a| = |c_b|$$

g = GERADE (EVEN)

$$\psi_g(\vec{r}) = N (a(\vec{r}) + b(\vec{r}))$$

SYMMETRIC UNDER INVERSION WITH RESPECT

$$\psi_u(\vec{r}) = N' (a(\vec{r}) - b(\vec{r}))$$

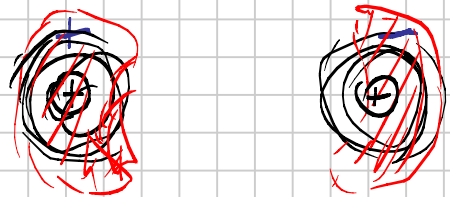
TO THE COM (ORIGIN)

u = UNGERADE (ODD)



GERADE

$$\psi_g = \frac{1}{\sqrt{1+s}} \cdot \frac{1}{\sqrt{2}} (a+b)$$



UNGERADE

$$\psi_u = \frac{1}{\sqrt{1-s}} \cdot \frac{1}{\sqrt{2}} (a-b)$$

$$S = \int d\vec{r} a^*(\vec{r}) b(\vec{r}) = \text{OVERLAP OF ATOMIC ORBITALS}$$

$$E_n(R) \begin{cases} \rightarrow E_g(R) = \langle \psi_g | H | \psi_g \rangle \\ \downarrow E_u(R) = \langle \psi_u | H | \psi_u \rangle \end{cases}$$

$$\psi_g(r)$$

$$(\sigma_g 1s)$$

"EQUIVALENT"

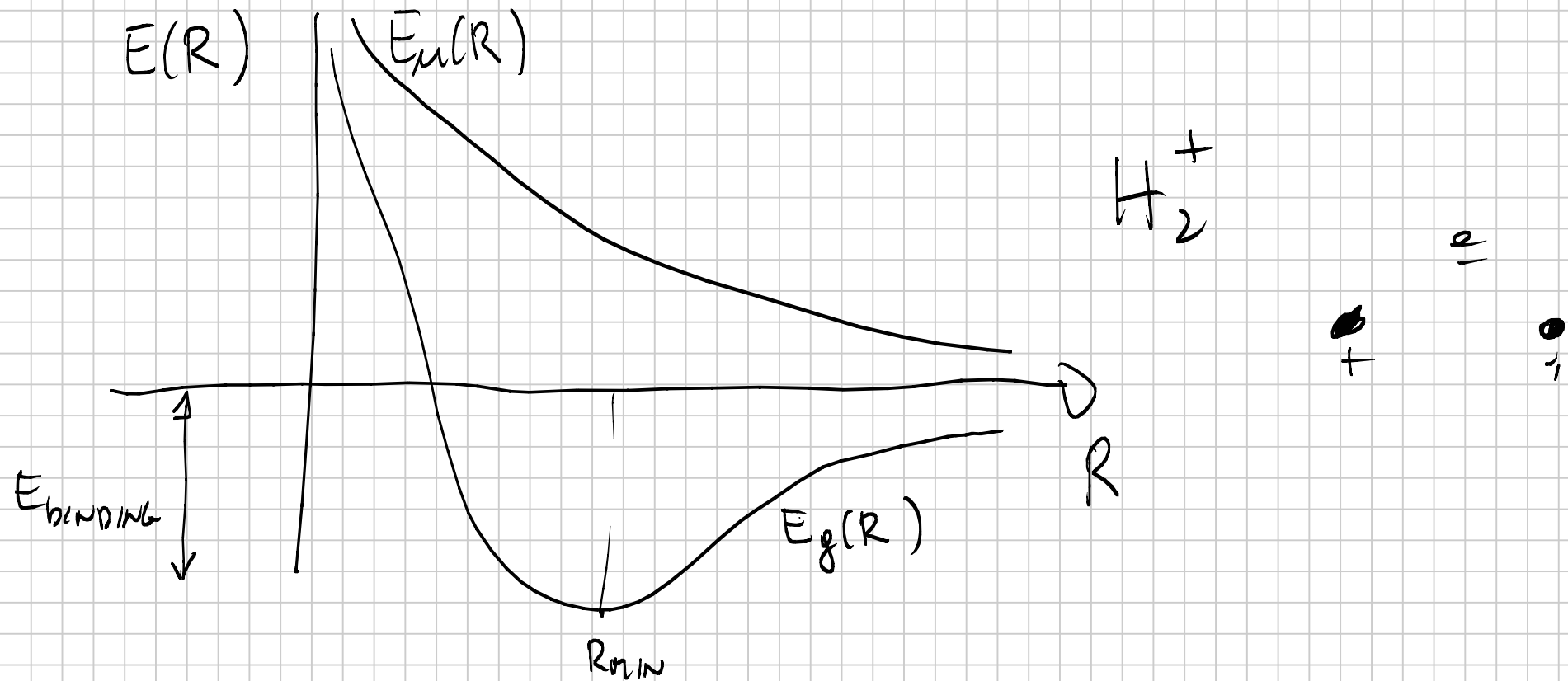
$$\psi_{1s}(u)$$

$$(1s)$$

$$\psi_u(\vec{r})$$

$$(\sigma_u 1s)$$

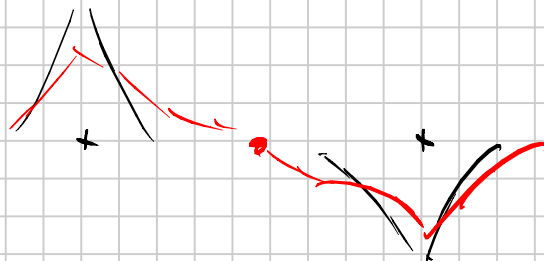
$$(1s)^2 (2s)^2 \dots$$

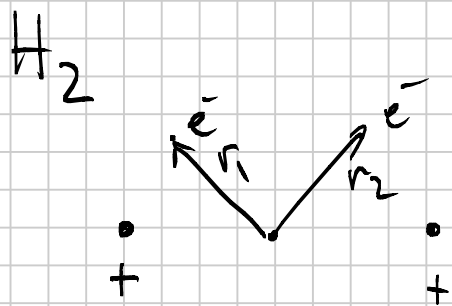


MO  $\rightarrow$   $E_b \sim 1.76 \text{ eV}$   
 $R_{\text{MIN}} \sim 1.32 \text{ \AA}$

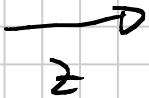
EXPERIMENT  
 $E_b \sim 2.79 \text{ eV}$   
 $R_{\text{MIN}} \sim 1.06 \text{ \AA}$

UNGERADE NOT STABILE





$$\psi_e^R(r_1, r_2)$$



$$(\sigma_g 1s)^2$$

( $\rightarrow$  NOT A GOOD QUANTUM #)  
 $L_{TOT}$

$L_z$  (TOTAL) IS A GOOD QUANTUM #  $L_z = \sum_i l_{z_i}$

$S_z$  " "

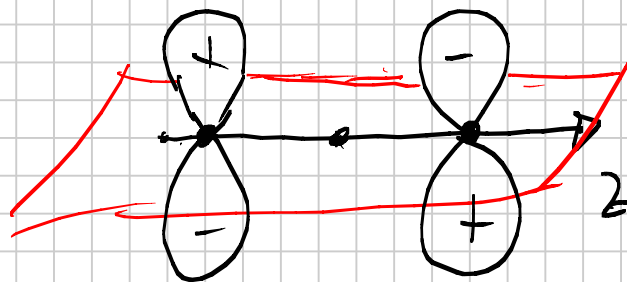
ATOMS	s	p	d	f	...	} $L_{TOT}$ CONSERVED FOR ATOMS
	S	P	D	F	..	

MOLECULES (AXIAL SYMMETRY)	g	$\pi$	$\delta$	$\phi$	} $L_z$ CONSERVED
	$\Sigma$	$\Pi$	$\Delta$	$\Phi$	

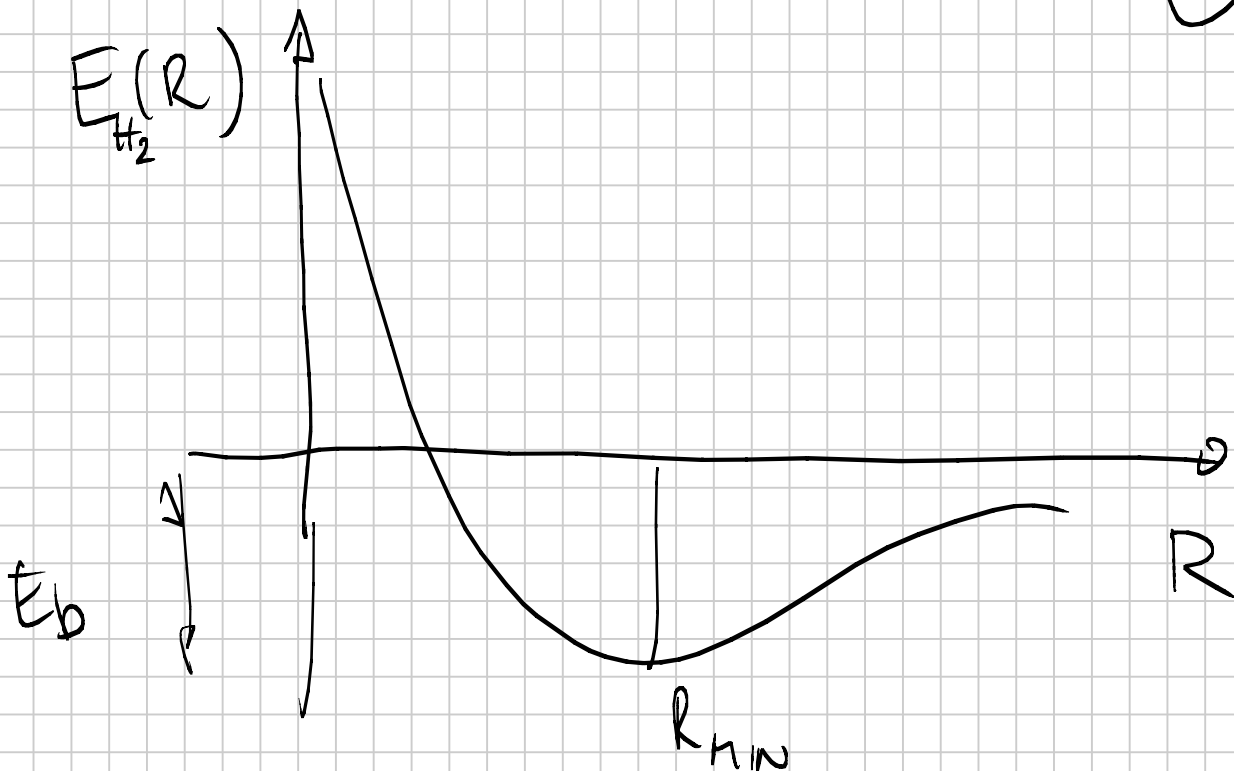
H<sub>2</sub><sup>+</sup>  $(\sigma_g 1s)^2 \sum_g \rightarrow$  SYMMETRIC WITH RESPECT TO A PLANE PASSING BY AXIS

H<sub>2</sub>

$$(\sigma_g 1s)^2 \sum_g^+$$



GERADE AND  
—  
UNGERADE WITH RESPECT TO PLANE



MOLECULAR ORBITAL

$$E_b \sim 2.65 \text{ eV}$$

$$R_{min} \sim .8 \text{ \AA}$$

EXPERIMENT

$$E_b \sim 4.72 \text{ eV}$$

$$R_{min} \sim .7 \text{ \AA}$$

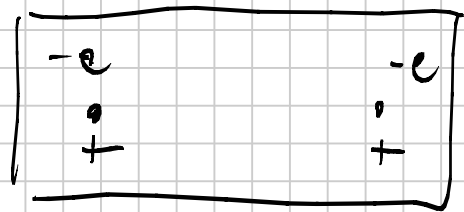
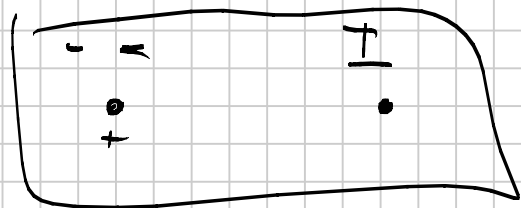
MOLECULAR ORBITAL FOR H<sub>2</sub> (A.K.A. HUND-MOLLIKEN)

CAN BE IMPROVED USING

# HEITLER-LONDON APPROXIMATION

$$MO \quad (\sigma_g | s)^2 = \psi_g(r_1) \psi_g(r_2) = (a(r_1) + b(r_1))(a(r_2) + b(r_2))$$

$$\psi_{MO}(r_1, r_2) = \overset{\text{I}}{a(r_1)} a(r_2) + \overset{\text{II}}{a(r_1)} b(r_2) + b(r_1) \overset{\text{II}}{a(r_2)} + b(r_1) \overset{\text{I}}{b(r_2)}$$



HEITLER-LONDON  
KEEP ONLY TERMS OF TYPE II

$$\psi_{HL}(r_1, r_2) \sim N [a(r_1) b(r_2) + a(r_2) b(r_1)]$$

$$E_b \sim 3.14 \text{ eV}$$

$$R_{min} \sim .8 \text{ \AA}$$

1D DIMENSIONALE

$H_2$



$$\frac{e^2}{\hbar} \rightarrow e^2 \delta(x)$$

$$\psi_{1S}(x) = e^{-|x|}$$