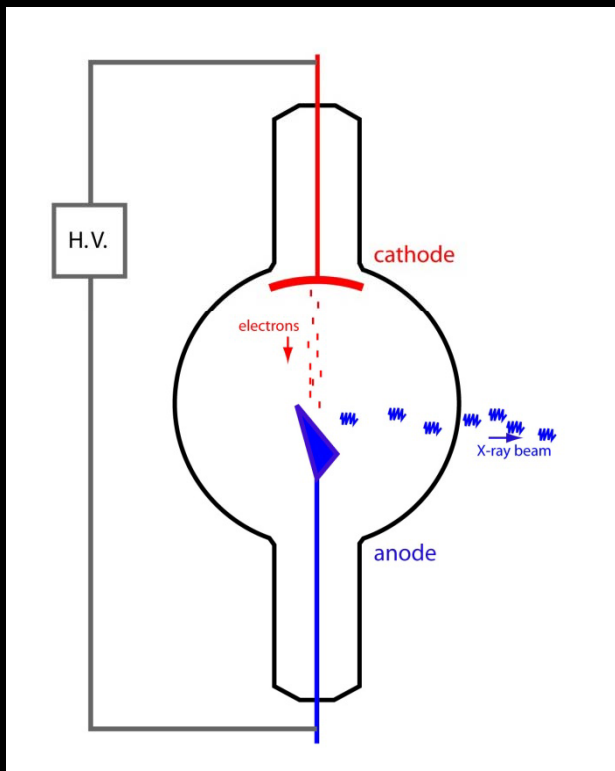


A normal chest X-ray

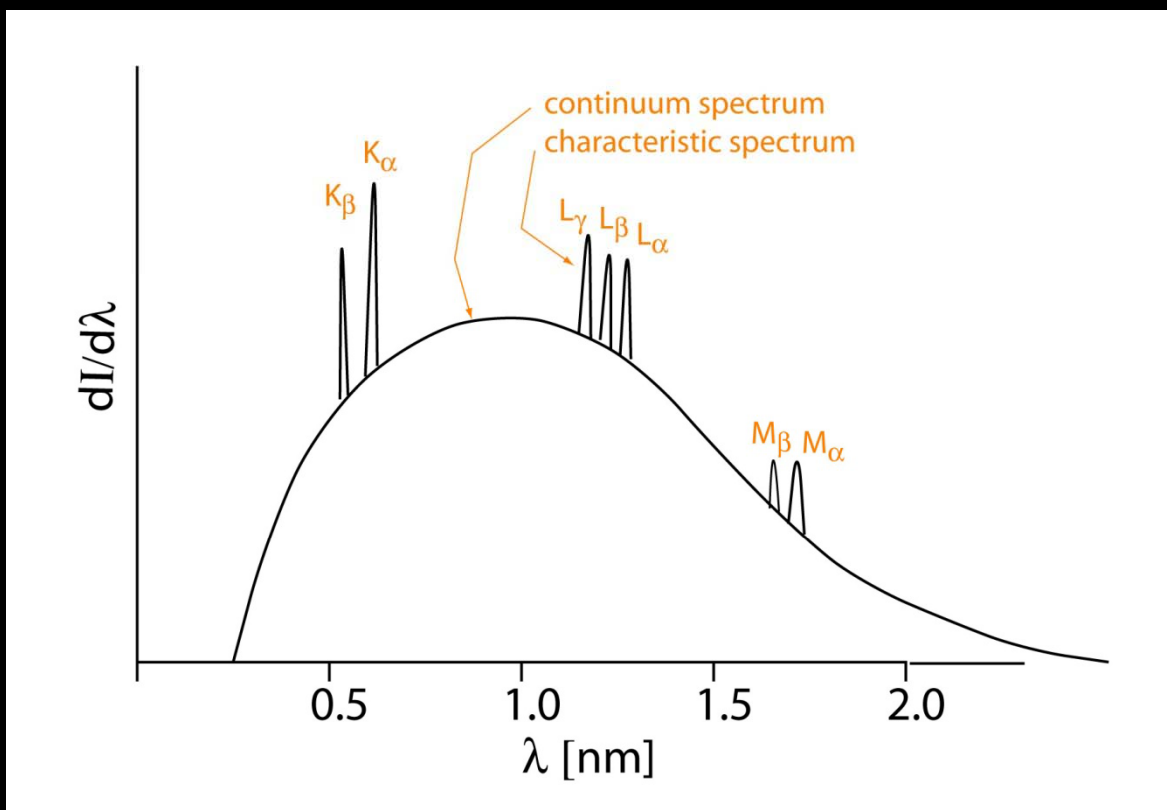


**Interactions of X rays
and matter**

Production of X rays

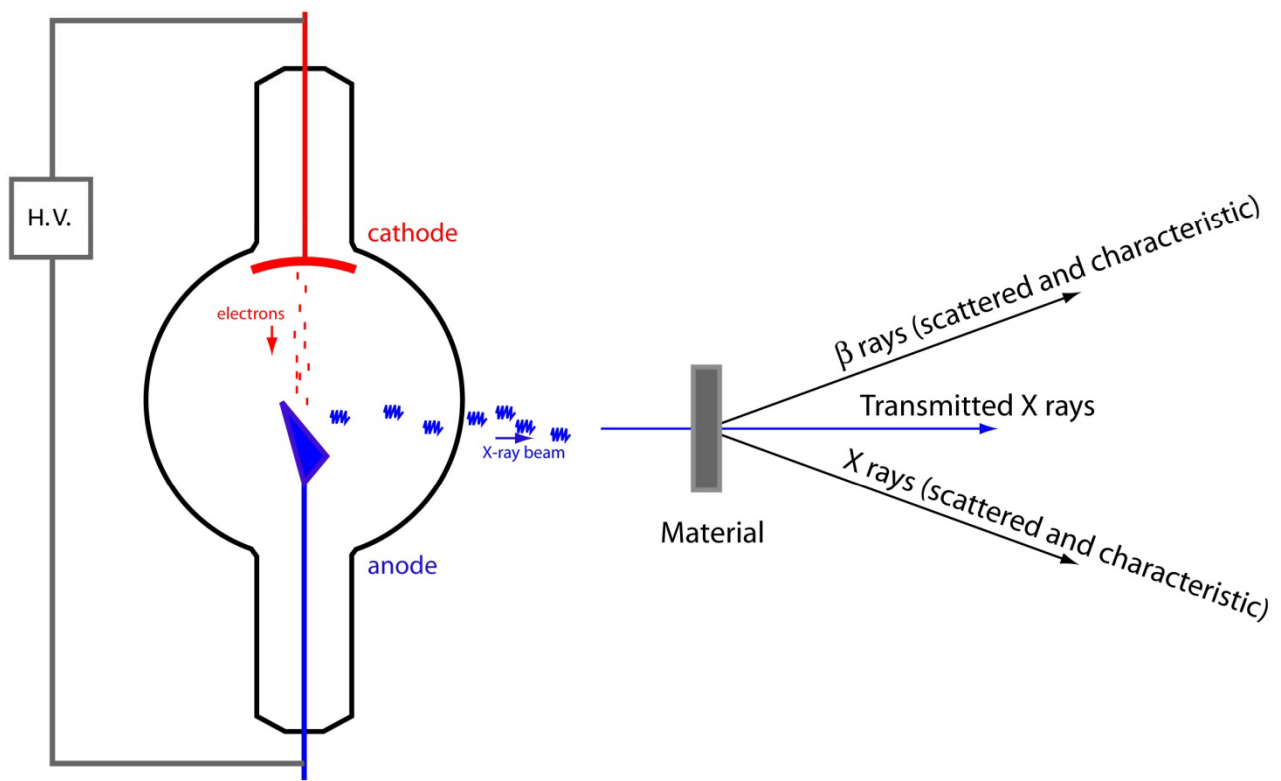


Example.
Suppose H.V. = 10 kV.
Then calculate λ_{\min} .



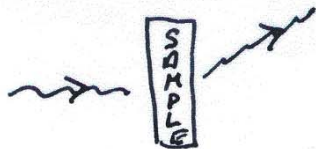
Interactions of X rays and matter

FIGURE 17 Interaction of X rays with matter



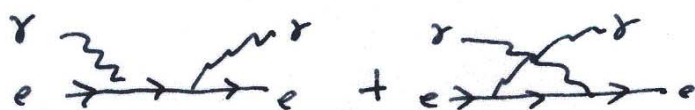
Interaction of X-rays with matter

Scattering



- Thomson scattering

$$\gamma + e \rightarrow \gamma + e \quad \text{with } E_\gamma \ll mc^2$$



$$\frac{d\sigma}{d\Omega} = \frac{1}{2} r_e^2 (1 + \cos^2 \theta) \quad \text{where } r_e = \frac{e^2}{4\pi\epsilon_0 mc^2}$$

$$\sigma_T = \int \frac{d\sigma}{d\Omega} \sin\theta d\theta d\phi = \frac{8\pi}{3} r_e^2$$

(J.J. Thomson, theory)
published ~1906)

$$r_e = 2.8 \times 10^{-15} \text{ m}$$

$$\sigma_T = 6.6 \times 10^{-29} \text{ m}^2 = 0.66 \text{ barns}$$

- Compton scattering ($E_\gamma \gtrsim mc^2$)

The X-ray loses some energy

$$\frac{E_\gamma'}{E_\gamma} = \frac{mc^2}{mc^2 + E(1 - \cos\theta)}$$

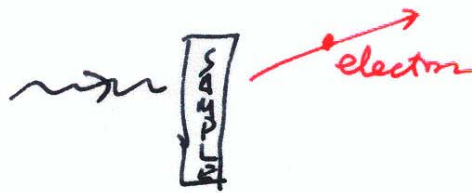
$$\lambda' - \lambda = \frac{h}{mc} (1 - \cos\theta)$$

$$\underline{E = hf = \frac{hc}{\lambda}}$$

(Arthur Compton, 1923)

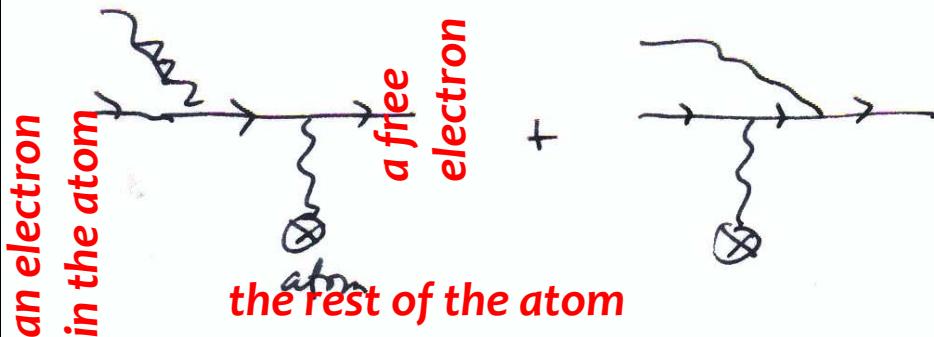
Interactions of X-rays with matter

Absorption



- Photoelectric effect

... on an atomic scale

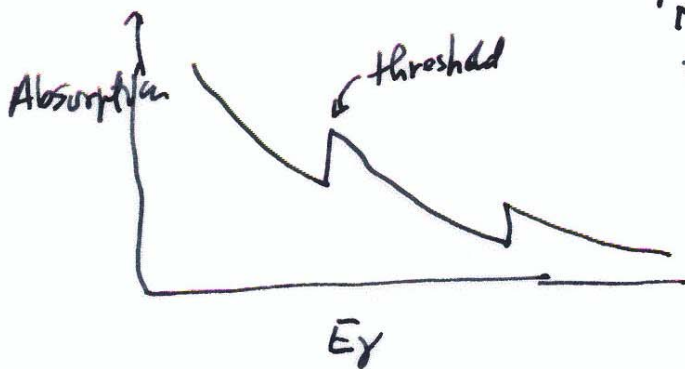


$$hf - B = \frac{1}{2}mv^2$$

(neglects recoil energy of the atom)

$$hf = B + \frac{1}{2}mv^2 \geq B$$

↑ minimum γ energy for this process.

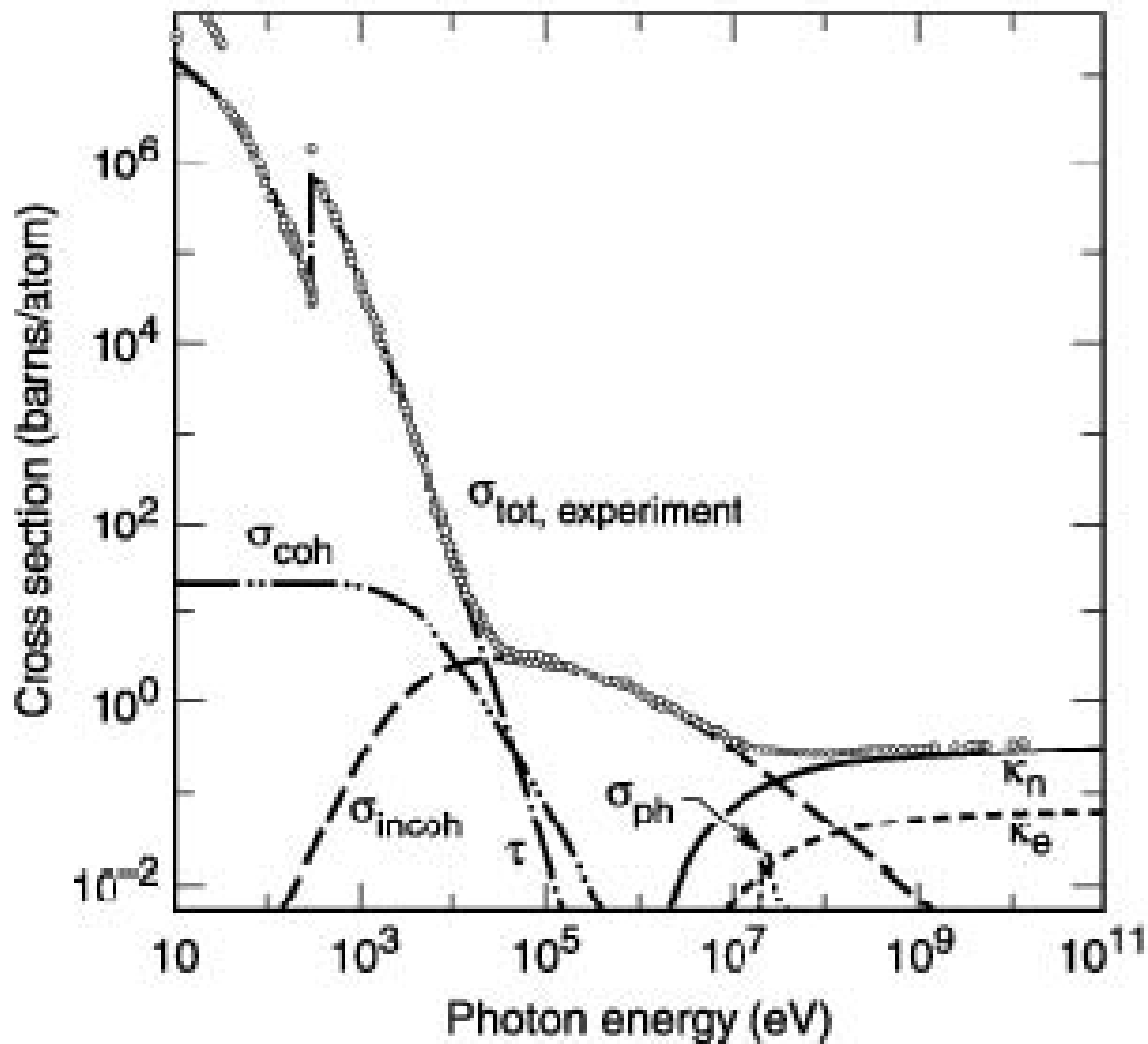


- Pair Production: $\gamma + \text{atom} \rightarrow e^- e^+ + \text{atom}$



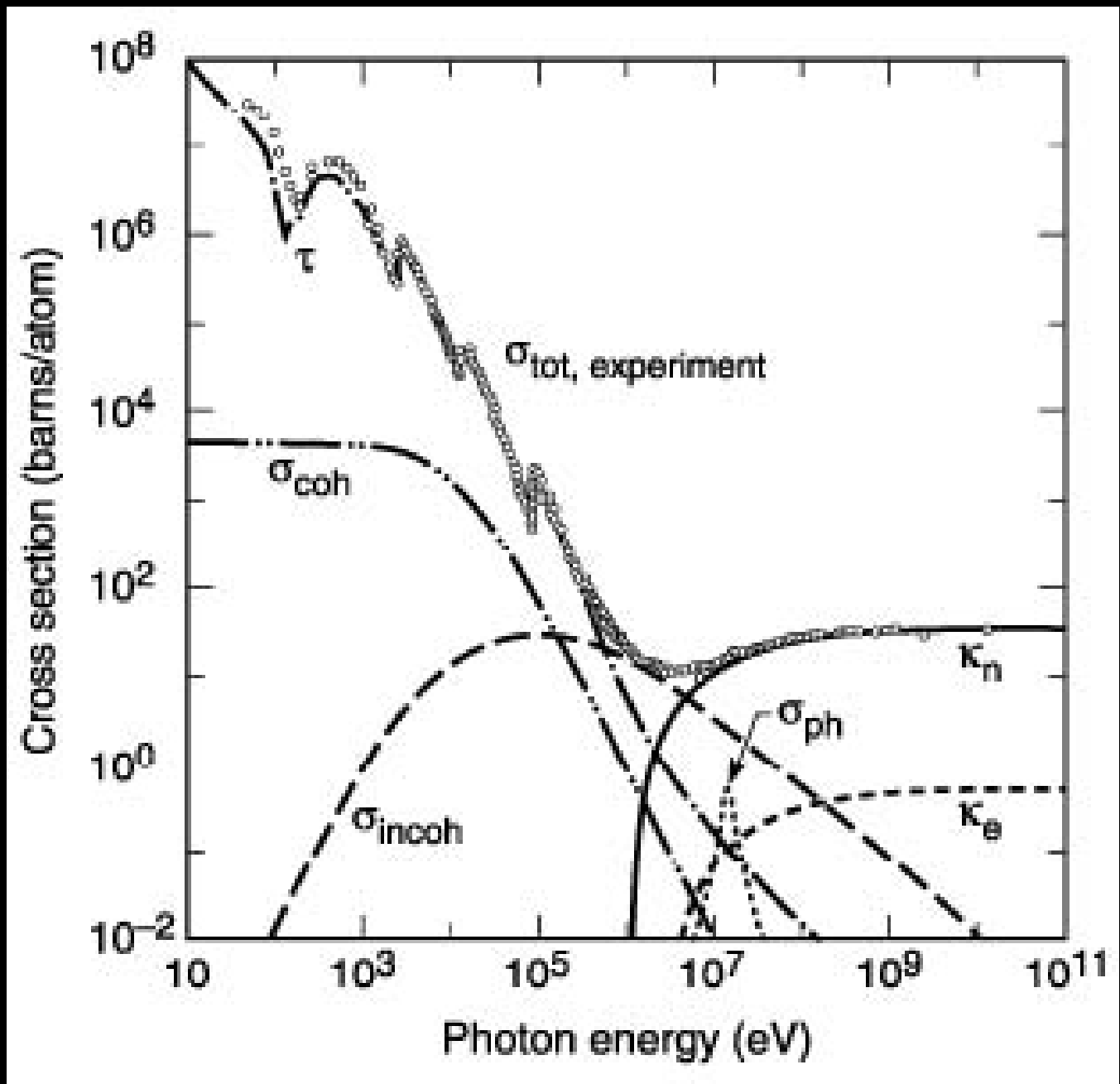
Can occur if $E_\gamma > 2mc^2 = 1.02 \text{ MeV}$

Carbon cross sections



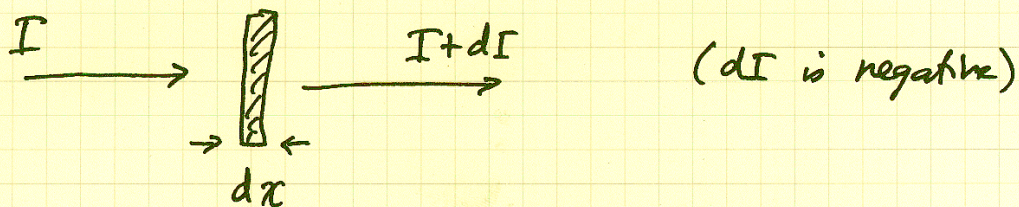
X-ray Data Booklet
xdb.lbl.gov

Lead cross sections



X-ray Data Booklet
xdb.lbl.gov

The X-ray absorption coefficient (Sec 1.18)



$$\begin{aligned} \text{Intensity} &= \# \text{ ~~X rays~~ photons per unit area per unit time} \\ &= \frac{d^2 N}{dA dt} \quad (\text{Units: } m^{-2} s^{-1}) \end{aligned}$$

dI is negative because photons are absorbed; e.g., the photoelectric effect $\gamma + \text{atom} \rightarrow e + \text{ion}$.

This is a random process so $dI = -\mu I dx$
where μ is a constant \rightarrow [units of μ : m^{-1}]

$$\mu(dx) = \frac{\text{fraction of intensity absorbed in length } dx}{|dI| / I}$$

$$\boxed{\frac{dI}{dx} = -\mu I}$$

μ : linear absorption coefficient
units: m^{-1}