







## Lifetime of Stars

- Lifetime = Amount of fuel/Rate of consumption
  - Lifetime of a tank of gas for a car
  - For a star

    - Amount of fuel = mass
      Rate of consumption = luminosity
- Lifetime = mass / luminosity
- Stars have a finite life. The sun will not live forever!

Spectral Class	Abs Mag	Luminosity [Lsun]	Mass [Msun]	Lifetime [Tsun]
03	-6	25000	40	1/600
G2 (sun)	5	1	1	1
M0	10	1/100	0.3	30

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Lifetime = n Stars have a The sun will - Life of su O3 stars	nass / luminos finite life. not live fore un is 10Byr	sity ver!		
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#### **Cluster of Stars**

- In a cluster of stars
  - All stars were born at the same time.
  - Some are massive and live a short life.
    - On a human scale: 20T if the sun scales to 100lb.
    - On a human scale: 5 wk if the sun scales to 70yr.
  - Some have little mass.

Spectral Class	Abs Mag	Luminosity [Lsun]	Mass [Msun]	Mass	Lifetime [Tsun]	Lifetime
O3	-6	25000	40	20T	1/600	5wk
G2 (sun)	5	1	1	100lb	1	70yr
M0	10	1/100	0.3	30lb	30	2000yr

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# 19<sup>th</sup> Century "Energy Crisis"

- Luminosity of sun L=4×10<sup>26</sup>Watt
- Mass m=2×10<sup>30</sup>kg
- How long will the sun last if the energy is produced by burning coal? C+O<sub>2</sub>→CO<sub>2</sub>
  - Life time = m×(E/m)/L
  - E/m=9MJ/kg
  - 1500 years
- Earth is much older than that.



### E=mc<sup>2</sup>

- Crisis: No solution with physics of 19th century.
- Einstein's new theory (1906)
  - Energy can change into mass, and mass can change into energy.
  - $E = m c^{2}$ .
  - Energy = mass  $\times$  (speed of light)<sup>2</sup>.
- Changing a little mass produces a lot of energy. Compare kinetic energy  $\frac{1}{2}$  m v<sup>2</sup> with m c<sup>2</sup>.
  - Speed of light c = 300,000 km/s
  - Air in blast furnace moves at 0.2 km/s



- Chemical reaction  $C+O_2 \rightarrow CO_2$ -  $E=m c^2/100,000,000,000$ . One part in 100 billion of mass disappears and changes into energy.
- Sun contracts by 10%
  - $E=m c^2/1,000,000$ . One part in a million of mass disappears and changes into energy.
- H fuses to produce He
  - E=m c<sup>2</sup>/140. A part in 140 of the mass disappears and changes into energy.

#### Nuclear fusion In a nuclear reaction, converting a significant fraction of the mass to energy is possible. Hans Bethe figured out the nuclear physics of how this happens. Hans Bethe $4 \, {}^{1}\text{H} \rightarrow {}^{4}\text{He} + \text{neutrinos} + 2e^{+} + \text{energy}$ 1906-2005 4 hydrogen nuclei fuse One helium nucleus is produced Which is heavier? A box of hydrogen and a box of 1. helium, neutrinos, and positrons made from the hydrogen? A. Box containing H B. Box containing the products: He, neutrinos, and positrons C. The two boxes have the same mass.





