

Mass of the Electron

Motivation for the Experiment

The aim of this experiment is to measure the mass of the electron. The mass will be deduced from a measurement of the charge to mass ratio, e/m , combined with a value for the charge taken from elsewhere.

Table I gives the values for the mass and charge of the electron for comparison purposes. They are known to much higher accuracy than we can hope to measure in our experiments and, therefore, they will be considered to be exact.

Table I
Electron Charge and Mass

<u>Quantity</u>	<u>Symbol</u>	<u>Value</u>	<u>Units</u>
Mass	m_e	9.109×10^{-31}	kg
Charge	e	1.602×10^{-19}	C

I. The Charge to Mass Ratio (e/m) for the Electron

Introduction

In 1897 J. J. Thompson discovered the first "elementary particle", the electron, by measuring the ratio of its charge to mass in a manner similar to the experiment that we will perform. Given the mass and charge in Table I we expect to measure a value that is close to:

$$\frac{e}{m_e} = \frac{1.602 \times 10^{-19} \text{C}}{9.109 \times 10^{-31} \text{kg}} = 1.759 \times 10^{11} \frac{\text{C}}{\text{kg}} \quad (\text{I-1})$$

Theory

Both conceptually and experimentally this part of the experiment is quite simple. A hollow glass sphere is evacuated to a high vacuum and filled with a small amount of Neon gas or Mercury vapor at a pressure of about 1 millitorr. Inside the sphere, a heated wire (a "filament") emits a large number of electrons which are accelerated by a potential difference V , to acquire kinetic energy:

$$\frac{1}{2} m_e v^2 = e \cdot V \quad (\text{I-2})$$

and, therefore, a velocity:

$$v = \sqrt{\left[2V \cdot \frac{e}{m_e} \right]} \quad (\text{I-3})$$

The moving electrons strike and ionize the Ne atoms which give off light when they recombine, producing a visible beam along the electron track. The sphere is placed in a region where a magnetic field, \mathbf{B} , is produced by a current in two coils of wire. \mathbf{B} is applied perpendicular to the velocity vector of the electrons resulting in a magnetic force:

$$\mathbf{F} = e \mathbf{v} \times \mathbf{B} \quad (\text{I-4})$$

The force is perpendicular to both the velocity and the magnetic field vectors and produces an acceleration \mathbf{a} , whose magnitude is given by:

$$|a| = \frac{|F|}{m_e} = \frac{v^2}{r} \quad (\text{I-5})$$

where r is the radius of the circular path of the electron. (Recall that such a force is called a **centripetal** force and the corresponding acceleration a **centripetal** acceleration with both vectors directed toward the center of the circle.)

Therefore:

$$m_e \frac{v^2}{r} = e \cdot v \cdot B \quad (\text{I-6})$$

which yields:

$$\frac{e}{m_e} = \frac{v}{B \cdot r} \quad (\text{I-7})$$

Substituting for v from Eq. 3, we obtain:

$$\frac{e}{m_e} = \frac{1}{B \cdot r} \sqrt{\frac{2eV}{m_e}} \quad (\text{I-8})$$

which simplifies to:

$$\frac{e}{m_e} = \frac{2V}{(B r)^2} \quad (\text{I-9})$$

This equation is correct **IF** V and B are both constant in space and time so that all the electrons have the same velocity v , and follow the same circular path of radius r .

V is applied between two conducting equipotential surfaces by a stable power supply and is therefore well defined.

Two coils of radius R , separated by a distance d , each having N turns with current I passing through them in the same sense, generate a magnetic field on their common axis, half way between them, equal to:

$$B_0 = \frac{\mu_0 N I}{R} \left[\frac{1}{1 + \left(\frac{d}{2R}\right)^2} \right]^{\frac{3}{2}} \quad (\text{I-10})$$

here $\mu_0 = 4\pi \times 10^{-7}$ henries/per meter is the permeability constant. You should verify that when $d = R$, this expression reduces to that for a Helmholtz pair, $(4/5)^{3/2} \mu_0 NI/R$. This configuration provides a rather uniform field, especially on the mid-plane between the coils.

Extra Credit: correction for field variation. Equation I-10 gives the field at midpoint of the central axis of the coils. There are corrections of a few % to the field seen by the electron in its orbit as it moves in the plane equidistant from the coils, because the field changes once you move away from the center. To avoid large corrections, confine your measurements to radii less than about 70% of the radius of the coils thereby limiting corrections to 5% or less.

To correct for the effects of the field away from the center, we need to know how the field varies as a function of the radius s in the mid plane between the two coils. Figure 1 shows this variation as a function of the fractional radius, s/R , where s is the distance from the central axis of the Helmholtz pair to the point of interest. Note that in each case the field is normalized to the value at $s = 0$. Note that because our electrons travel in a circle starting from the accelerating electrode (which has a fixed position), these

orbits are not at a constant s , except for one particular value of r . Thus, using this correction is approximate, and requires estimating an average value of s for the orbit. To do so will require a good sketch of the apparatus.

EC Questions: Explain the systematic errors due to neglecting this correction: would e/m be too low or too high? Can you give an upper bound for how large the effect might be in your data? How would this vary with the orbit radius r ?

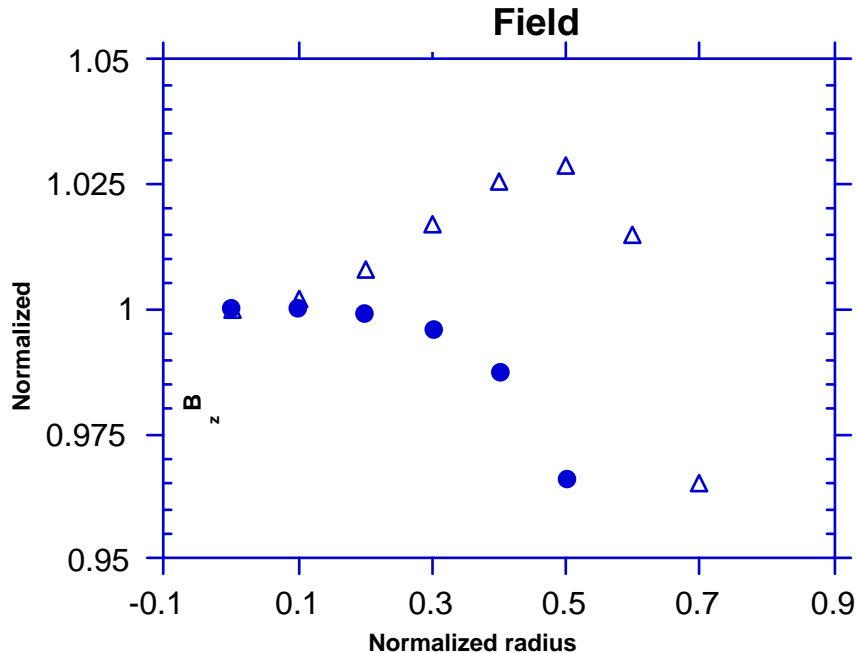


Fig. I-1: Normalized magnetic field in mid plane for Helmholtz pair (closed circles) and a different configuration (open triangles) as a function of the normalized distance from the axis, s/R .

Apparatus

e/m tube

Magnetic field coils with 124 turns per coil: $N = 124$.

Power supplies and meters

The Experiment

The e/m tube consists of the glass bulb, the hot wire filament, the accelerating electrodes and a set of reference marks to indicate the diameter of the electron path which is made visible by ionization of the Neon gas.

The Kepco power supply provides the current for the magnet coils while the Heathkit power supply provides the accelerating voltage. The Pasco power supply provides the ac voltage to heat the filament.

Measure the radius of one of the coils as well as their separation in order to calculate the central magnetic field from Eq. I-10. Think about exactly what distance you should measure, i.e. how to deal with the thickness of the coils.

The power supplies are connected as shown in the schematic (Fig. I-2 below). The two coils are in series with each other so that each coil has exactly the same current. There is an ammeter in series with the coils, and a voltmeter in parallel with the accelerating voltage and across the heater voltage. Use an

AC frequency different from 60 Hz to avoid beat frequency oscillations in the beam. Record the heater voltage used.

Turn down all the voltages to their minimum, then turn on the power supplies. Let the heater voltage (marked “amplitude” on the Pasco supply) warm up a few minutes and then apply about +150 V accelerating voltage (on the Heathkit supply). Adjust the heater voltage so that you obtain a fairly sharp beam of electrons (not too fuzzy). This controls the intensity of the beam. You may have to raise it a fair bit to get a beam started and then you can lower it back down. In any case, stop raising the voltage when the orbit stops changing (we do not understand the coupling!). Once you’re satisfied with the electron beam, don’t touch the heater current knob until the end of the experiment.

Adjust the current creating the magnetic field until the electron beam forms a circular path. Adjust the bulb orientation in the magnetic field so that the electron path is circular and not spiral.

Compute the actual magnetic field at each radius using Eq I-10 and I-9.

Collect data for several different voltages and magnet currents. The “rungs” on the ladder in the tube are 2 cm apart; the top tick is 10cm above beam. The easiest way to take data is to set the accelerating voltages to 150V, 200V and 250V and then to adjust the magnetic field currents so that the top of the electron beam is either level with one of the rungs, or halfway between two rungs. Take as many readings (magnetic field currents) for each accelerating voltage as possible.

Turn down all the voltages, and then turn off the power supplies: this helps the tubes live longer.

Calculate your e/m value before leaving the lab!

Plot your values for e/m_e as a function of the radius using different symbols for the different accelerating voltages. Do a simple statistical analysis to (mean, standard deviation for each voltage) determine the errors in your measurement and put these as error bars on your plot.

Question 1: Which of your input measured quantities (I, V, radius, coil dimensions, etc) has the largest fractional uncertainty?

Question 2: Estimate your fractional uncertainty in e/m by finding the fractional error contribution due to your largest fractional uncertainty, and discuss whether this uncertainty does a reasonable job of predicting the standard deviation of your measurements. *Hint:* if you need the uncertainty of the magnetic field, estimate it by considering the expression in brackets in Eq I-10 to be a constant.

Question 3: Do your data vary by more than the statistical error bars? If so, these variations may indicate the presence of a systematic error in your measurements. Do your data suggest the presence of such a systematic error? If they do, you will need to assign a reasonable value to these systematic errors based on your data and take them into account when you compute your value for m_e at the conclusion of this lab.

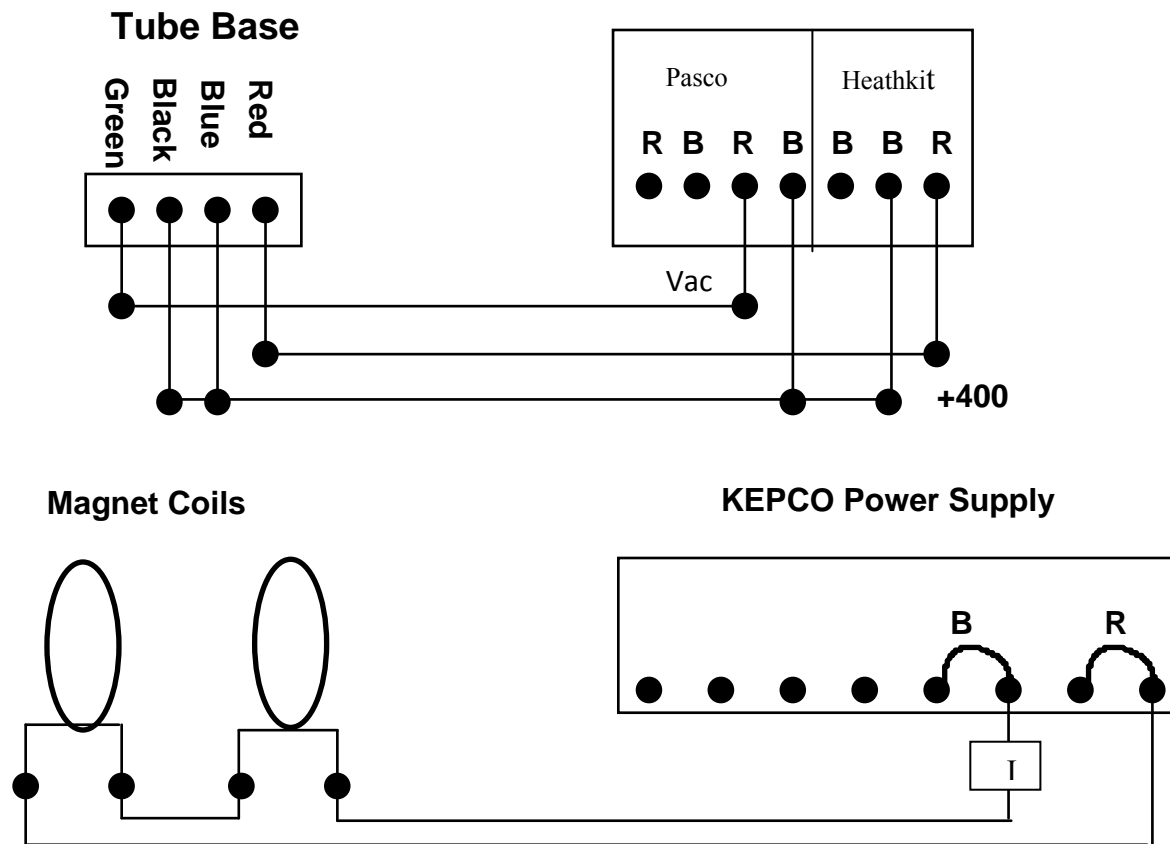


Fig. I-2: Schematic Circuit Diagram

- Notes: 1) The coils are connected “in series” but the correct direction of current flow is chosen to ensure the right direction of beam deflection.
- 2) In the upper part of the diagram, lines entering a black circle represent wires connected at that point, but lines that cross each other at 90 degrees represent non-intersecting wires.

II. The Charge of the Electron

The classic method for measuring electron charge is described in the appendix (so you know what you’re missing), but we won’t be performing the measurement ourselves. The essence of the method is that small droplets of liquid may carry a net charge. By observing their motion in an electric field, the net charge can be deduced, and by observing many droplets, the smallest difference in charge between two droplets can be calculated. This smallest charge difference is the unit of charge, the charge of an electron.

For the purposes of changing your e/m value into a measurement of the electron mass, let us assume you did a measurement of the electron charge, with the result of $1.71 \text{ E-19 Coulomb}$, with a fractional uncertainty of 5%.

The Combined analysis

Using the best estimate for e/m (mean, std dev, sdm) from part I of the experiment and your best estimate of e from part II, compute the mass of the electron, m_e , and the uncertainty in the mass, δm_e . Indicate whether or not your measurement agrees with the known value.

The Write-up

Include tables and graphs where appropriate. Give a summary table including your final results for the charge and mass of the electron, including an error analysis. Compare with the given values and discuss discrepancies, if any.

Appendix: Measurement of the Charge of the Electron

The charge of the electron will be measured using a variation of the Millikan oil drop experiment.

Description of the Apparatus

The apparatus is designed so that all necessary components are contained in one unit. It is mounted on a metal base, which supplies all necessary power inputs when it is connected to a 110 V AC, 50/60 Hz outlet and consists of the following parts:

- A storage bottle with spray bulb pump for producing spheres of latex liquid (diameter $\sim 1/1000$ mm).
- A 30X scale microscope which has a resolution of 0.2 mm between the small divisions and 1 mm between the large divisions.
- An electrode assembly.
- Appropriate controls, including a polarity-reversing switch, a potentiometer for fine control of plate voltage, and a voltmeter indicating the plate voltage applied.

See Figure II-1 for an illustration of the apparatus with a key listing the parts.

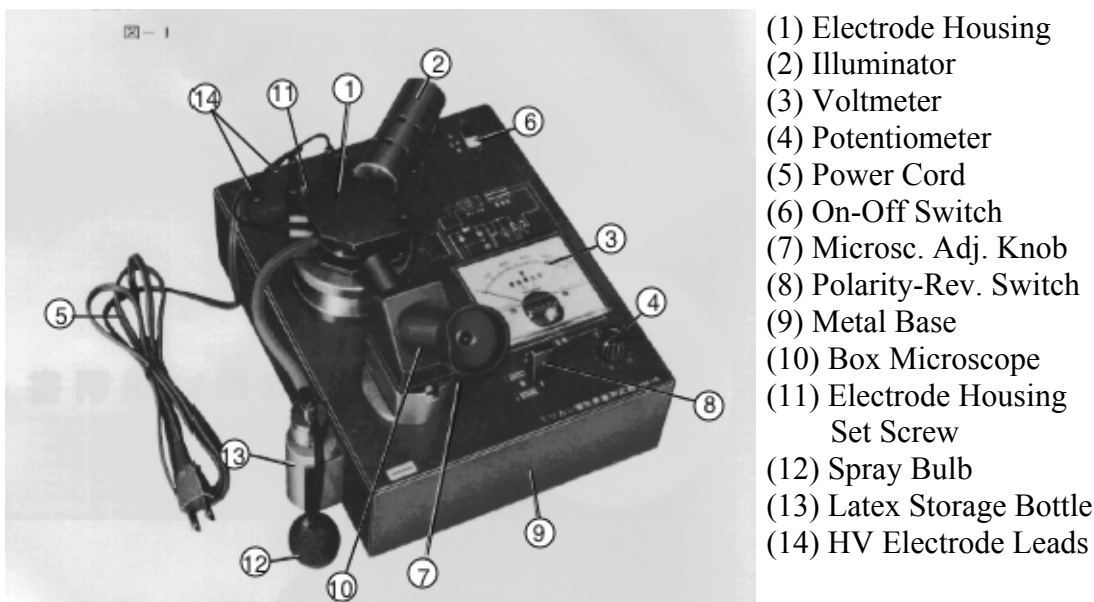


Fig. II-1: The Millikan Apparatus

Theory

The experiment is named for R. A. Millikan, the American physicist who devised it. (Millikan's original experiment used drops of oil, while this apparatus uses spheres of latex liquid.) Millikan wanted to determine whether electrical charge occurred in discrete units and, if it did, whether there was such a thing as an elementary charge.

In the Millikan experiment, a small charged ball made of latex moves vertically between two metal plates. This sphere is too small to be seen by the naked eye, so the projector and microscope are used to enable the user to see the sphere as a small dot of light. When there is no voltage applied to the plates, the sphere falls slowly and steadily under the influence of gravity, quickly reaching its terminal velocity. When a voltage is applied to the plates, the terminal velocity of the sphere is affected not only by the force of gravity but also by the electric force acting on the sphere.

When the experimenter knows the density of the latex ball, the terminal velocity of the ball falling under the influence of gravity alone, and the charge on the plates of the Millikan apparatus, it is possible to find the force produced by the electric charge on the ball. A series of observations will produce a group of terminal velocity values which are seen to be multiples of a lowest value. From this data, it is possible to determine the elementary unit of charge.

Consider a latex sphere of mass m and charge q , falling under the influence of gravity between two horizontal plates. In falling, the sphere is subjected to an opposing force due to air resistance. The speed of the sphere quickly increases until a constant terminal speed is reached, at which time the weight of the sphere, mg , minus the buoyant force is exactly equal to the air resistance force. The value of the air resistance force on a sphere was first derived by Sir George Stokes and is given as $6\pi\eta r s$ where η is the coefficient of viscosity of air, r is the radius of the sphere and s is its terminal speed. If the buoyant force of the air is neglected, the equation of motion of the sphere is:

$$mg - 6\pi\eta r s = 0 \quad (\text{II-1})$$

Now suppose that the metal plates are connected to a source of constant potential difference such that an electric field of intensity E is established between the plates and a latex sphere of charge q is made to move upwards. The direction of the electric field must depend on the sign of the charge q , which may be either positive or negative. The resultant upward force on the charge is $Eq - mg$ and this force causes the sphere to move upwards with a terminal speed s^+ . The equation of motion is:

$$Eq - mg = 6\pi\eta r s^+ \quad (\text{II-2})$$

If now the polarity of the electric field is reversed, the sphere will move downward under the combined force of gravity and the electrostatic force. The equation of motion is:

$$Eq + mg = 6\pi\eta r s^- \quad (\text{II-3})$$

Note that the forces are now additive and that the terminal speed is achieved in the opposite direction than in the previous case.

The effect of gravity can now be eliminated by adding the equations of motion yielding:

$$2Eq = 6\pi\eta r (s^+ + s^-) \quad (\text{II-4})$$

If the terminal speeds are changed to velocities by incorporating the proper sign convention of upwards (+) and downwards (-), then one must change the symbols from s to v and add a minus (-) before the term for the downward velocity. We will not need to write that equation here because Eq. (II-4) will be sufficient for our purposes. (The corresponding equation with velocities can also lead to mistakes if one forgets that the downward velocity should have a negative sign.)

Performing the Measurement

Make sure the electrode housing is set as in Figure II-2. This requires unscrewing the electrode housing set screws and removing the top upper electrode housing plate. Clean the top and bottom plate with H_2O + Qtips and kimwipes. Get a new chamber and bulb from the front of room. There are two small glass windows. Do not touch the glass! Then place the chamber on the bottom plate so that light is shining through the "dark hole". Hold onto the chamber, line up the screws from the top with the holes and carefully place the top snugly on the chamber, then snug up the screws, but not too hard. Be sure the alignment is still good.

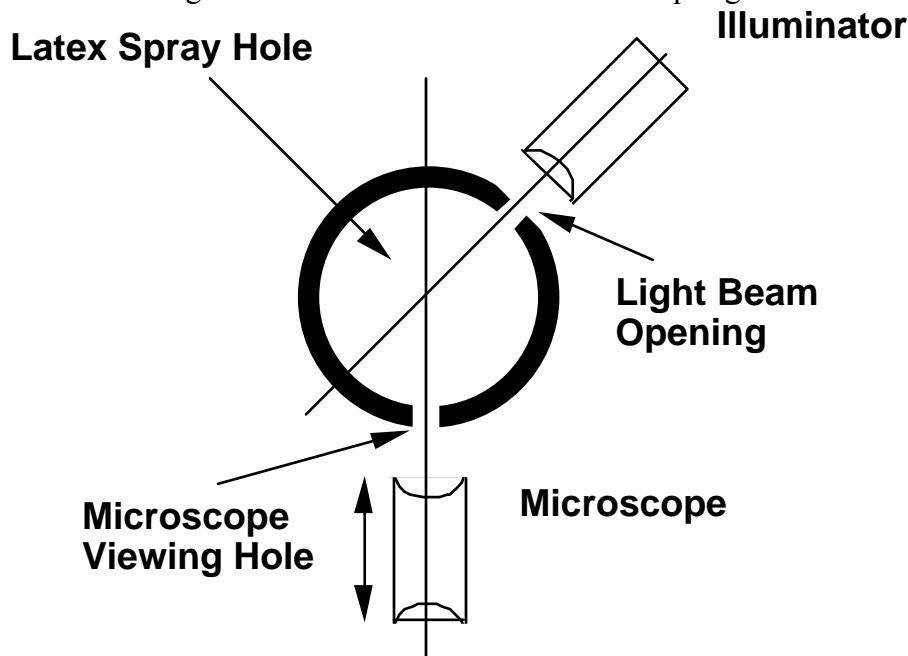


Fig. II-2: Millikan Apparatus Schematic Setup

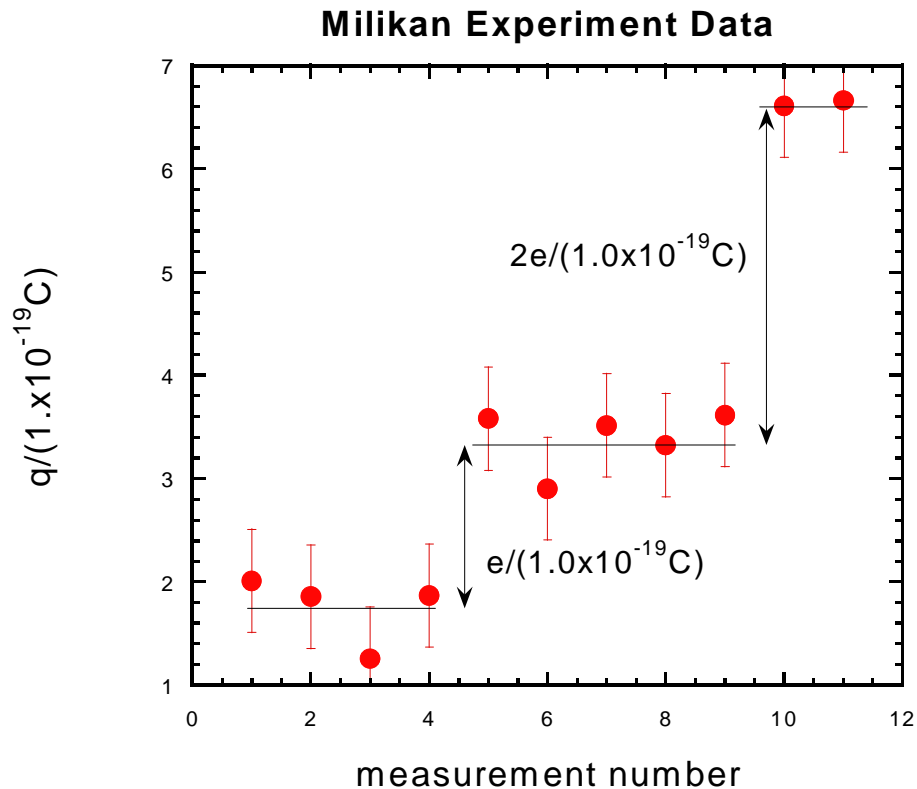
Inject latex spheres into the chamber by using the spray bulb pump. Shake your bottle of latex before using, and **write down the droplet size marked on the bottle**. To feed the latex spheres in, dip the yellow tube into the latex, and cover the air hole of the spray bulb pump with a finger and squeeze the bulb (relatively hard). Note that the spheres will not be injected unless the air hole is covered. **After squeezing, take your thumb back off the hole to allow air back in.** Otherwise you waste the fluid by drawing it back into the bulb, and flood the chamber. Spraying is usually difficult the first few times. We recommend that you begin your measurements after making several test sprays. After using the apparatus, **clean the spray tubing** with water, Qtips and Kleenex.. If the spray is not cleaned after each lab section, the residue of latex will harden, preventing smooth operation of the spray.

1. Position the 3-way polarity switch in its mid position, establishing a “no charge” condition. Turn the “on/off” switch to “on”. The illuminating lamp should light.
2. Adjust the microscope by rotating the focus-adjusting knob until an approximate mid position is established. The eyepiece divisions should be distinguishable from the background.
3. Adjust the electrode voltage to 450 volts using the voltage-adjusting knob, but keep the polarity switch in the mid position.
4. Spray latex into the apparatus and carefully look for the “dots of light” in the microscope. If, after several pumps on the atomizer bulb the latex spheres are still not visible, adjust the microscope carefully to attempt to focus in on the spheres. Don't just keep spraying; ask for help if you can't see the dots of light.

Obtaining a suitable drop requires patience, for drops continue to enter the region between the plates for several seconds after spraying has stopped. Select a small drop which takes 20 seconds or more to move between two large divisions in the eyepiece. Though one person can do this experiment alone, it is helpful if two work together, one taking the readings of the times of rise and fall of the latex sphere and the other recording these readings.

Record the rise time t^+ and the fall time t^- of the latex sphere between two large divisions in the eyepiece (1 mm). Alongside, calculate the corresponding speeds s^+ and s^- , which are merely the distance between the two large divisions divided by the time the sphere took to travel that distance.

Continue to take similar data on as many different spheres as possible but, no less than three with different charges, and at least 3 s^+ and s^- measurements each. **Each student** should take their own data.



Calculate the charge. For two parallel plates as we have here, the E field is $E = V/d$, where V is the voltage across the plates and d , and distance between the plates, $E = V/d$, and the charge from Eq. (II-4) becomes:

$$q = 3\pi\eta r(s^+ + s^-) d/V \quad (\text{II-5})$$

For the latex spheres in air, $\eta = 1.8 \times 10^{-5}$, the latex sphere radius is r is **given on the bottle**, the plate voltage is V , and the distance between the plates, $d = 5 \times 10^{-3} \text{ m}$.

Divide the charge on each measurement by $1.0 \times 10^{-19} \text{ C}$, and plot the values for $q/(1.0 \times 10^{-19} \text{ C})$ on a graph similar to Fig. II-4. The data points should fall into groups, each group representing a different charge on the spheres. If the charge is an integer number times $1.6 \times 10^{-19} \text{ C}$, the groups will be bunched on the graph about that integer value times 1.6.

Draw an “eyeball” line through each group. Average the data in each group and use the differences between the average values for the groups to calculate e .