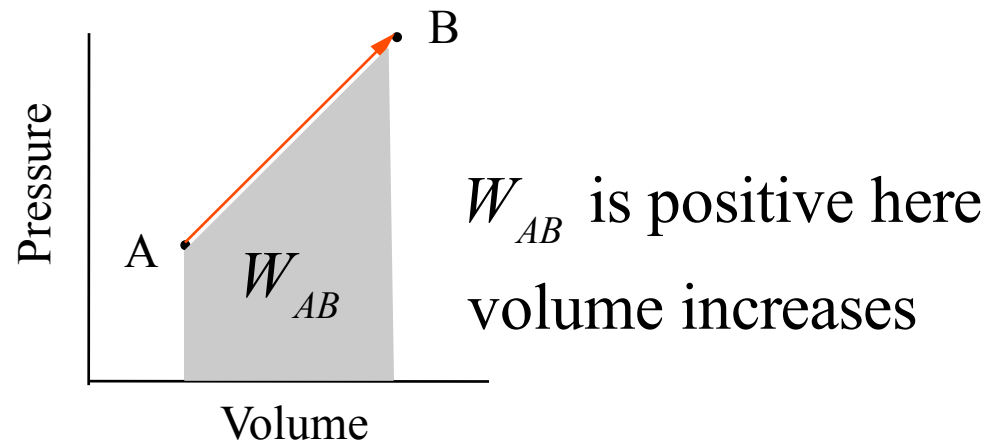


Chapter 15

Thermodynamics ***continued***

15 Work

The area under a pressure-volume graph is the work for any kind of process.



Clicker Question 15.3

Consider the pressure-volume graph shown for an ideal gas that may be taken along one of two paths from state A to state B. Path “1” is directly from A to B via a constant volume path. Path “2” follows the path A to C to B. How does the amount of work done along the two paths compare?

The area under a pressure-volume graph is the work for any kind of process.

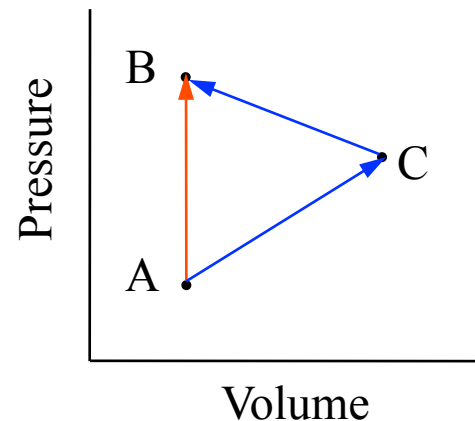
a) $W_1 = W_2 \neq 0$

b) $W_1 = W_2 = 0$

c) $|W_1| < |W_2|$

d) $|W_1| > |W_2|$

e) One needs P, V and T at each point to compare W .



Clicker Question 15.3

Consider the pressure-volume graph shown for an ideal gas that may be taken along one of two paths from state A to state B. Path “1” is directly from A to B via a constant volume path. Path “2” follows the path A to C to B. How does the amount of work done along the two paths compare?

The area under a pressure-volume graph is the work for any kind of process.

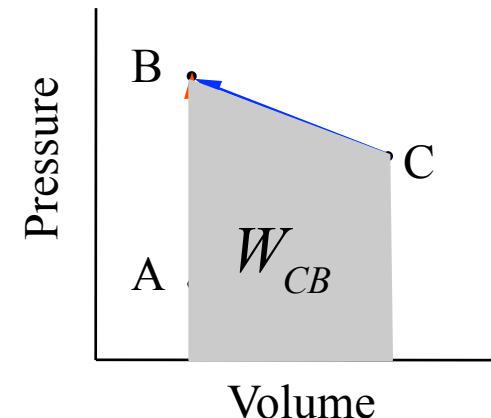
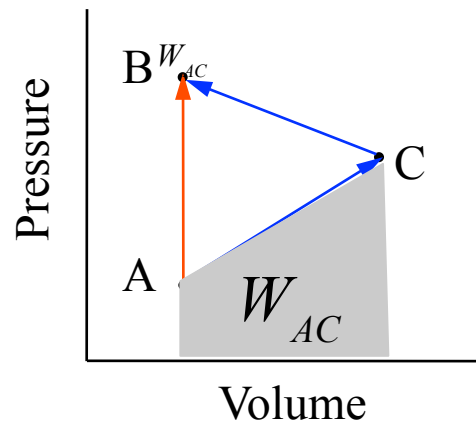
a) $W_1 = W_2 \neq 0$

b) $W_1 = W_2 = 0$

c) $|W_1| < |W_2|$

d) $|W_1| > |W_2|$

e) One needs $P, V \dots$



$$W_1 = 0$$

W_{AC} is positive and W_{CB} is negative.

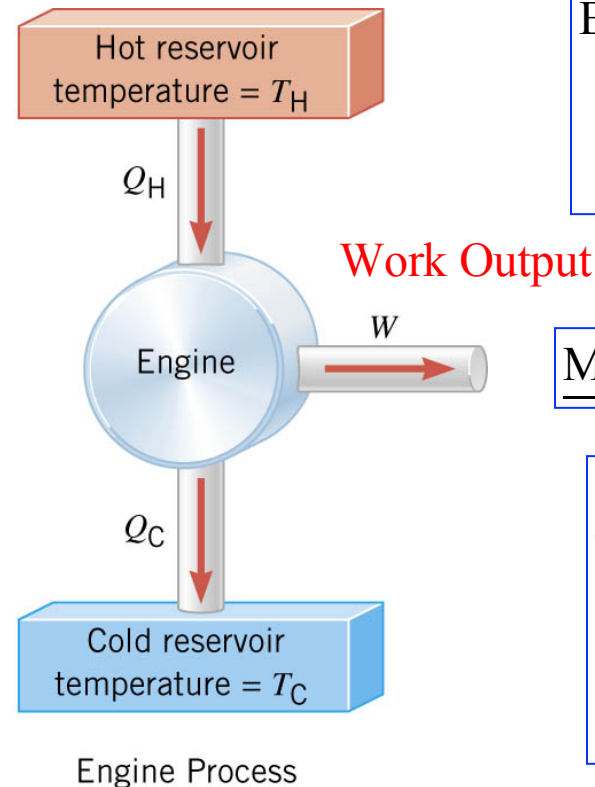
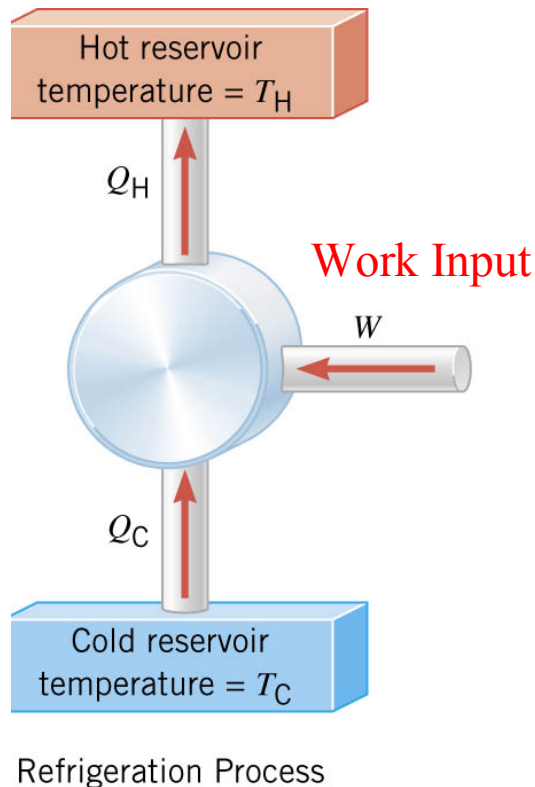
But $|W_{AC}| < |W_{CB}|$, so $W_{AC} + W_{CB} \neq 0$

15.10 Refrigerators, Air Conditioners, and Heat Pumps

Refrigerators, air conditioners, and heat pumps are devices that make heat flow from cold to hot. This is called the **refrigeration process**.

Heat-pump
coefficient of
performance
 $= \frac{|Q_H|}{|W|}$

Refrigerator
coefficient of
performance
 $= \frac{|Q_C|}{|W|}$



Engine Efficiency
 $e = \frac{|W|}{|Q_H|}$

Maximum efficiency

$$e_{Carnot} = 1 - \frac{T_C}{T_H}$$

$$= \left[\frac{W}{Q_H} \right]_{Carnot}$$

It is **NOT** possible to cool your kitchen by leaving the refrigerator door open or to cool your room by putting a window air conditioner on the floor by the bed.

The heat output into the room equals the heat removed PLUS the amount of work necessary to move the heat from a cold reservoir to a hot one.

15.10 Refrigerators, Air Conditioners, and Heat Pumps

Example 10 A Heat Pump

An ideal, or Carnot, heat pump is used to heat a house at 294 K. How much work must the pump do to deliver 3350 J of heat into the house on a day when the outdoor temperature is 273 K?

$$\begin{aligned} e_{\text{Carnot}} &= 1 - \frac{T_C}{T_H} = 1 - \frac{Q_C}{Q_H}; & Q_H &= Q_C + W \\ &= 1 - \frac{Q_H - W}{Q_H} = 1 - \left(1 - \frac{W}{Q_H} \right) = \frac{W}{Q_H} \end{aligned}$$

If $e_{\text{Heat-Pump}}$ is as good as e_{Carnot}

$$\begin{aligned} \frac{W}{Q_H} &= 1 - \frac{T_C}{T_H} \\ W &= Q_H \left[1 - \frac{T_C}{T_H} \right] = 3350 \text{ J} \left[1 - \frac{273}{294} \right] \\ &= 240 \text{ J} \end{aligned}$$

$$\begin{aligned} &\text{Heat-Pump} \\ &\text{Coefficient performance} \\ &\frac{|Q_H|}{|W|} = \frac{3350 \text{ J}}{240 \text{ J}} = 14 \end{aligned}$$

15.11 Entropy

In general, irreversible processes cause us to lose some, but not necessarily all, of the ability to do work. This partial loss can be expressed in terms of a concept called **entropy**.

$$\begin{array}{l} \text{Carnot} \\ \text{Engine} \\ \text{(is reversible)} \end{array} \quad \frac{|Q_C|}{|Q_H|} = \frac{T_C}{T_H} \quad \longrightarrow \quad \frac{|Q_C|}{T_C} = \frac{|Q_H|}{T_H}$$

Entropy change for Reversible (R) Processes

$$\Delta S = \left(\frac{Q}{T} \right)_R$$

Entropy (S) is a state-function of the system (like internal energy)

Consider the entropy change of a Carnot engine + surroundings. The entropy of the hot reservoir decreases and the entropy of the cold reservoir increases.

$$\begin{array}{l} \Delta S_H = (-) \left(\frac{Q_H}{T_H} \right) \quad \text{(heat leaves hot reservoir)} \\ \Delta S_C = (+) \left(\frac{Q_C}{T_C} \right) \quad \text{(heat enters cold reservoir)} \end{array}$$

NO Entropy change of surroundings

$$\Delta S_{total} = (+) \left(\frac{|Q_C|}{T_C} \right) (-) \left(\frac{|Q_H|}{T_H} \right) = 0$$

Reversible processes do not alter the entropy of the universe.

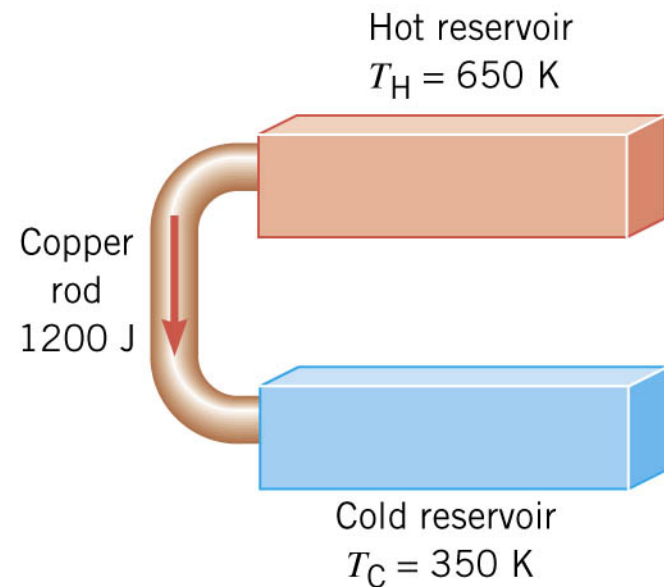
15.11 Entropy

Irreversible processes (e.g., Kinetic Friction)

Example 11 The Entropy of the Universe Increases

The figure shows 1200 J of heat spontaneously flowing through a copper rod from a hot reservoir at 650 K to a cold reservoir at 350 K. Determine the amount by which this process changes the entropy of the universe.

$$\begin{aligned}\Delta S_{\text{universe}} &= +\frac{|Q_C|}{T_C} - \frac{|Q_H|}{T_H} \\ &= +\frac{1200 \text{ J}}{350 \text{ K}} - \frac{1200 \text{ J}}{650 \text{ K}} = +1.6 \text{ J/K}\end{aligned}$$



No work was obtained from this heat transfer (0 efficiency engine)

Irreversible process lowers the amount of work possible between heat reservoirs

15.11 Entropy

Any irreversible process increases the entropy of the universe.

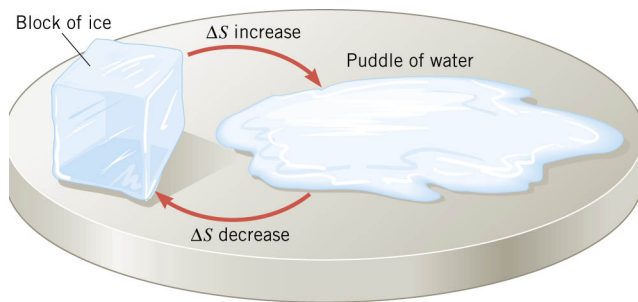
$$\Delta S_{\text{universe}} > 0$$

THE SECOND LAW OF THERMODYNAMICS STATED IN TERMS OF ENTROPY

The total entropy of the universe does not change when a reversible process occurs and increases when an irreversible process occurs.

$$W_{\text{unavailable}} = T_o \Delta S_{\text{universe}}$$

Melting Ice at 0°C (constant temperature) is a reversible process:



$$T_o = 273\text{K, Melting}$$

$$Q_{\text{ice}} = (+)m_{\text{ice}}L_f; \quad Q_{\text{surroundings}} = (-)m_{\text{ice}}L_f$$

$$\begin{aligned} \Delta S_{\text{universe}} &= \Delta S_{\text{melt-ice}} + \Delta S_{\text{surroundings}} \\ &= \frac{(+m_{\text{ice}}L_f}{T_o} + \frac{(-)m_{\text{ice}}L_f}{T_o} = 0 \end{aligned}$$

15.12 *The Third Law of Thermodynamics*

CORROLARY OF THE SECOND LAW OF THERMODYNAMICS

A perpetual motion machine is IMPOSSIBLE.

There will always be some irreversible process going on.

Most obvious irreversible process is kinetic friction.

Chapter 16

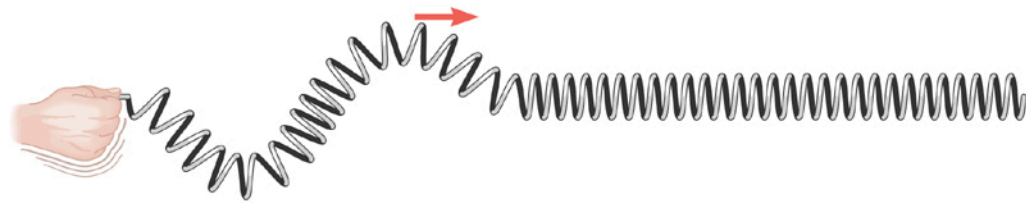
Waves and Sound

16.1 The Nature of Waves

1. A wave is a traveling disturbance.
2. A wave carries energy from place to place.



(a)



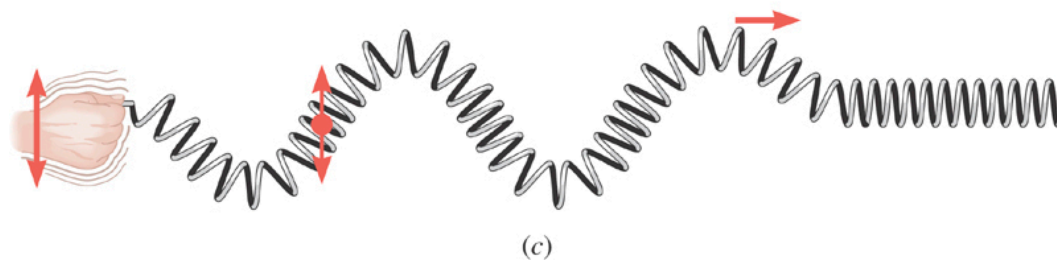
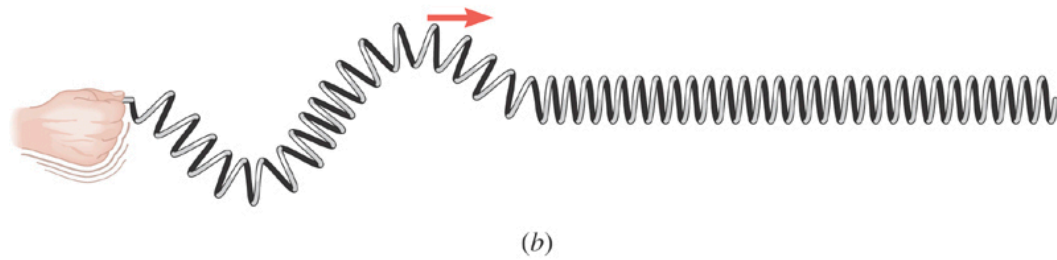
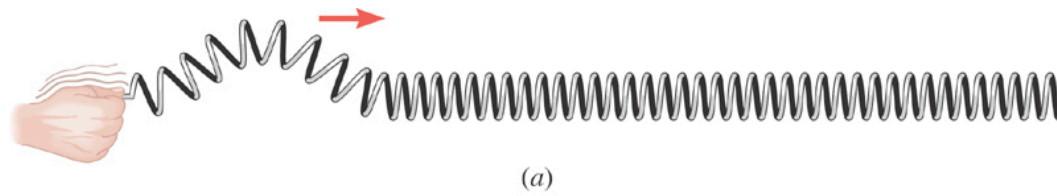
(b)



(c)

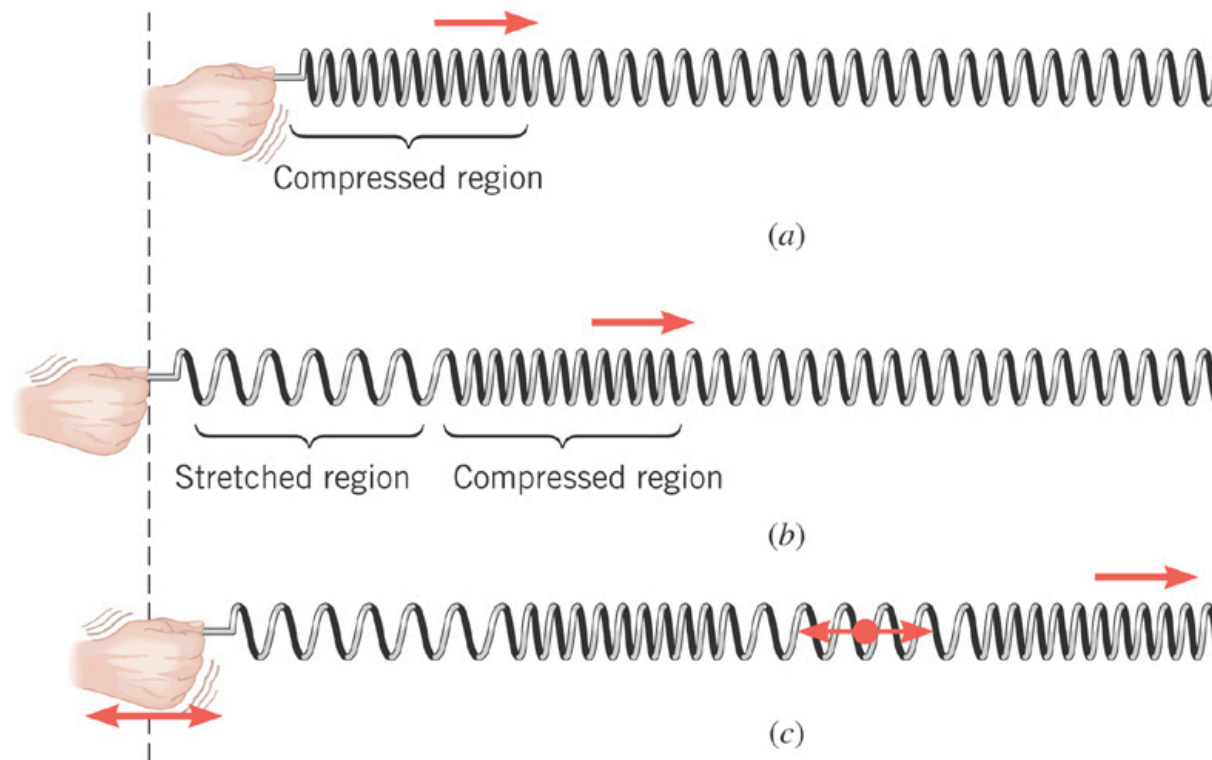
16.1 The Nature of Waves

Transverse Wave



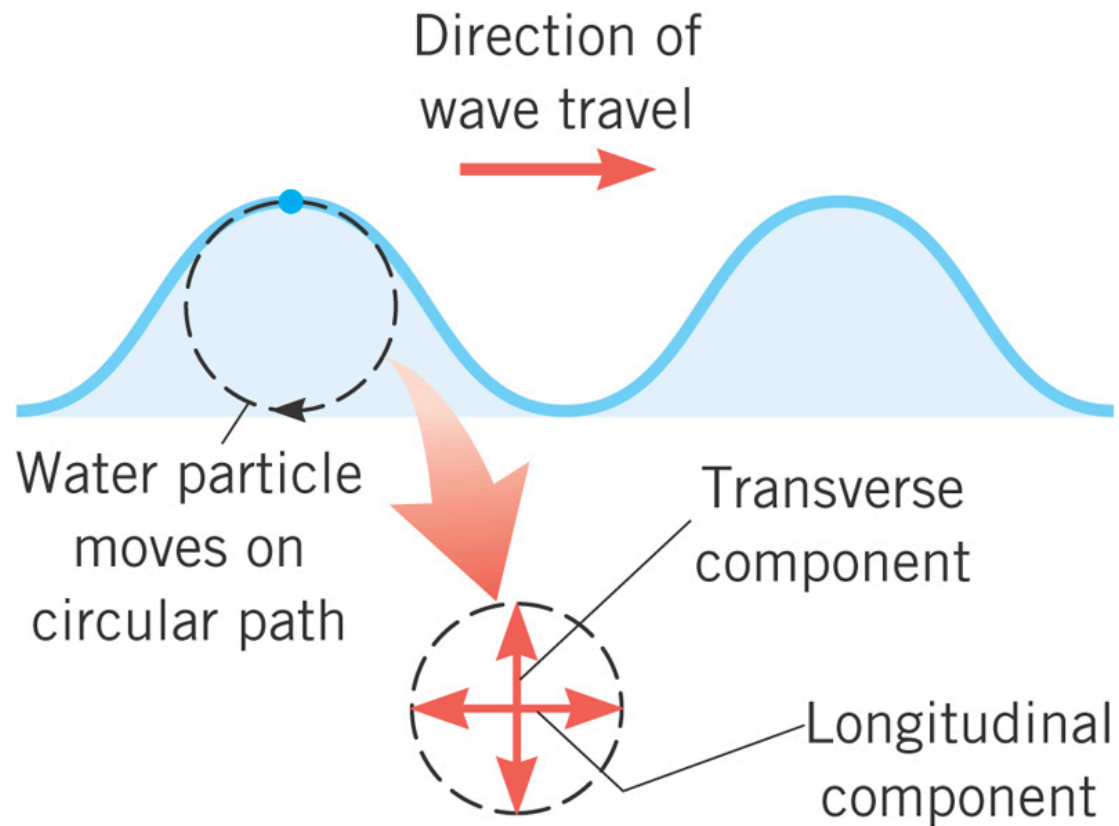
16.1 The Nature of Waves

Longitudinal Wave



16.1 The Nature of Waves

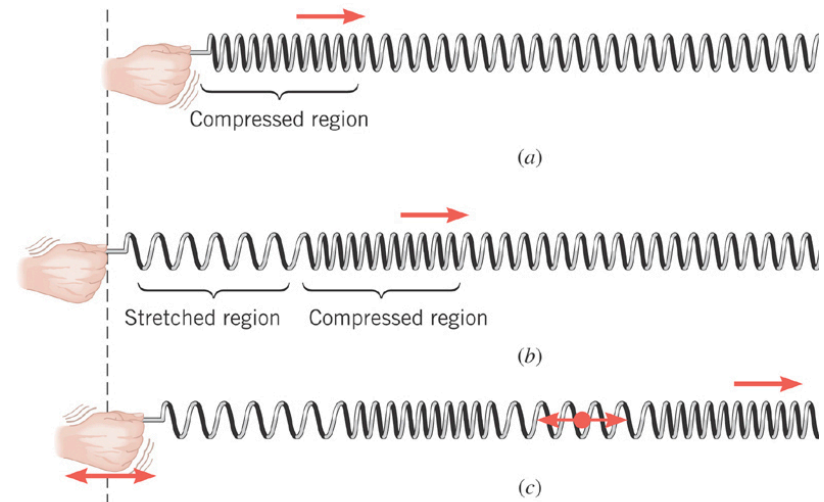
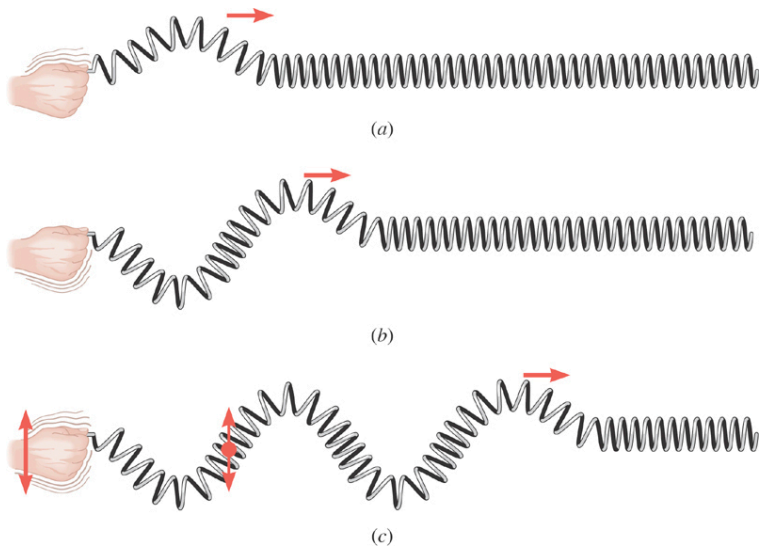
Water waves are partially transverse and partially longitudinal.



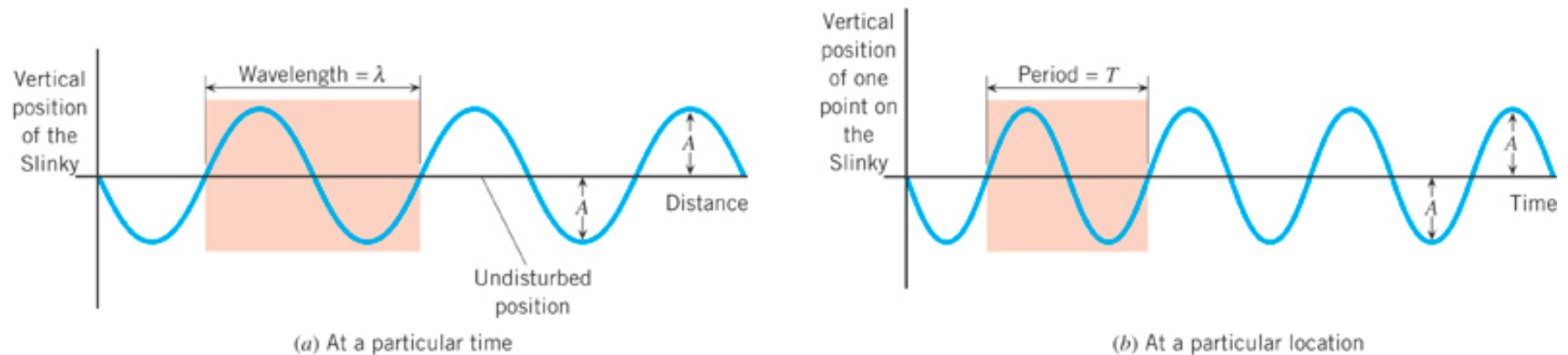
16.2 Periodic Waves

Periodic waves consist of cycles or patterns that are produced over and over again by the source.

In the figures, every segment of the slinky vibrates in simple harmonic motion, provided the end of the slinky is moved in simple harmonic motion.



16.2 Periodic Waves



In the drawing, one **cycle** is shaded in color.

The **amplitude** A is the maximum excursion of a particle of the medium from the particles undisturbed position.

The **wavelength** is the horizontal length of one cycle of the wave.

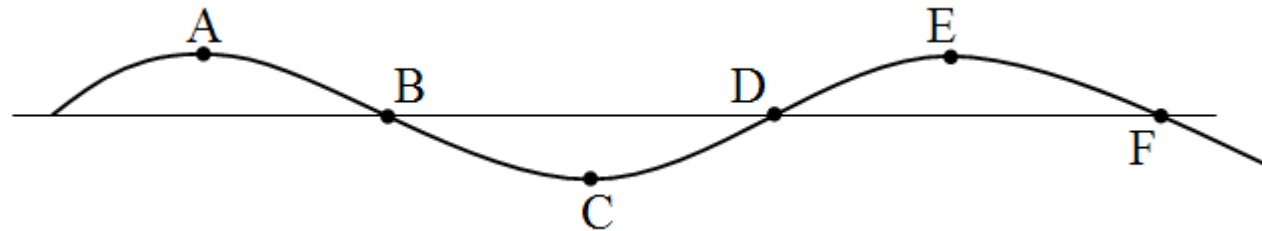
The **period** is the time required for one complete cycle.

The **frequency** is related to the period and has units of Hz, or s^{-1} .

$$f = \frac{1}{T}$$

Clicker Question 16.1

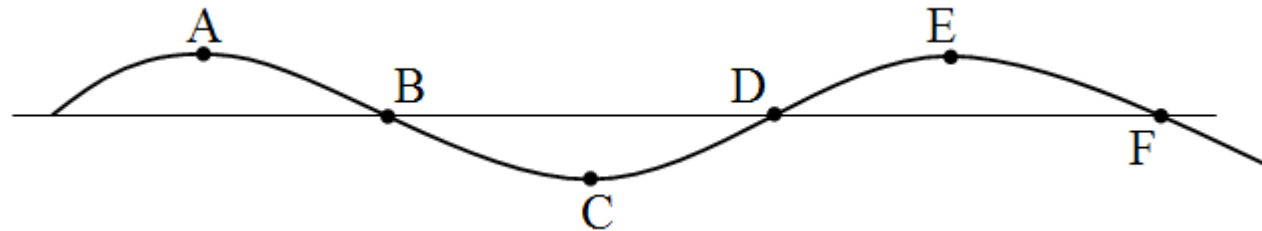
The drawing shows the vertical position of points along a string versus distance as a wave travels along the string. Six points on the wave are labeled A, B, C, D, E, and F. Between which two points is the length of the segment equal to one wavelength



- a) B to D
- b) A to E
- c) A to C
- d) A to F
- e) C to F

Clicker Question 16.1

The drawing shows the vertical position of points along a string versus distance as a wave travels along the string. Six points on the wave are labeled A, B, C, D, E, and F. Between which two points is the length of the segment equal to one wavelength



a) B to D

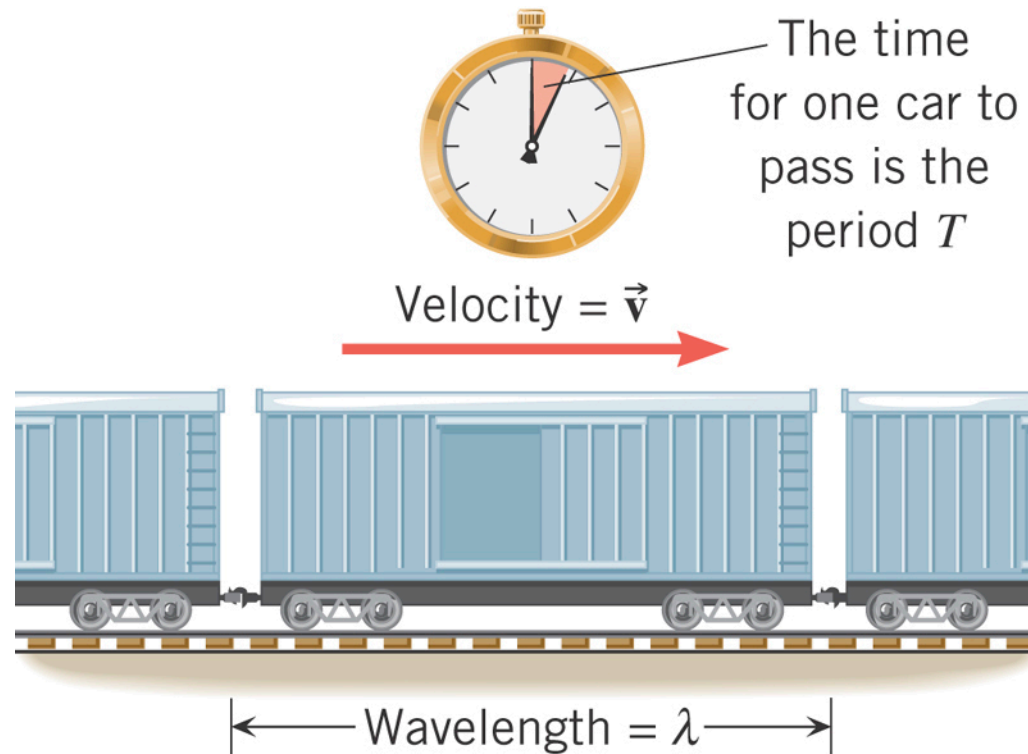
b) A to E

c) A to C

d) A to F

e) C to F

16.2 Periodic Waves



$$vT = \lambda; \quad f = \frac{1}{T}$$

$$v = \frac{\lambda}{T} = f\lambda \quad \Rightarrow \quad \lambda = \frac{v}{f}$$

16.2 Periodic Waves

Example 1 The Wavelengths of Radio Waves

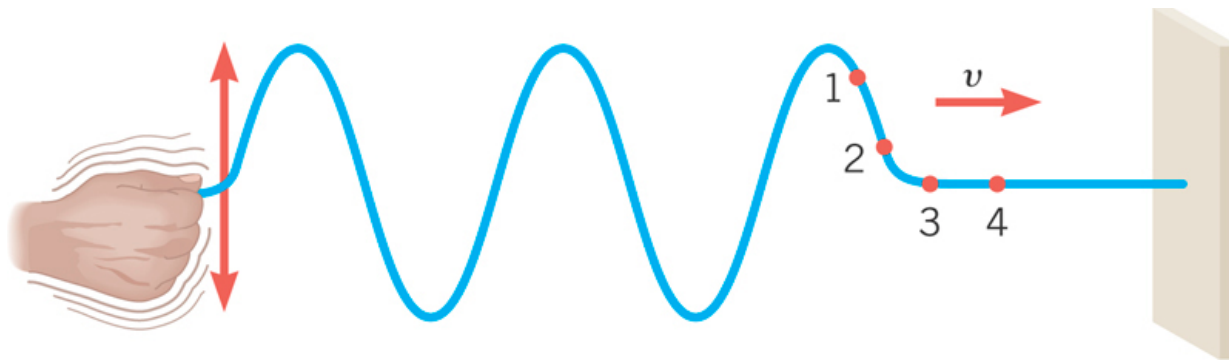
AM and FM radio waves are transverse waves consisting of electric and magnetic field disturbances traveling at a speed of $3.00 \times 10^8 \text{ m/s}$. A station broadcasts AM radio waves whose frequency is $1230 \times 10^3 \text{ Hz}$ and an FM radio wave whose frequency is $91.9 \times 10^6 \text{ Hz}$. Find the distance between adjacent crests in each wave.

$$\lambda_{\text{AM}} = \frac{v}{f} = \frac{3.00 \times 10^8 \text{ m/s}}{1230 \times 10^3 \text{ Hz}} = 244 \text{ m}$$

$$\lambda_{\text{FM}} = \frac{v}{f} = \frac{3.00 \times 10^8 \text{ m/s}}{91.9 \times 10^6 \text{ Hz}} = 3.26 \text{ m}$$

16.3 The Speed of a Wave on a String

The speed at which the wave moves to the right depends on how quickly one particle of the string is accelerated upward in response to the net pulling force.



$$v = \sqrt{\frac{F}{m/L}}$$

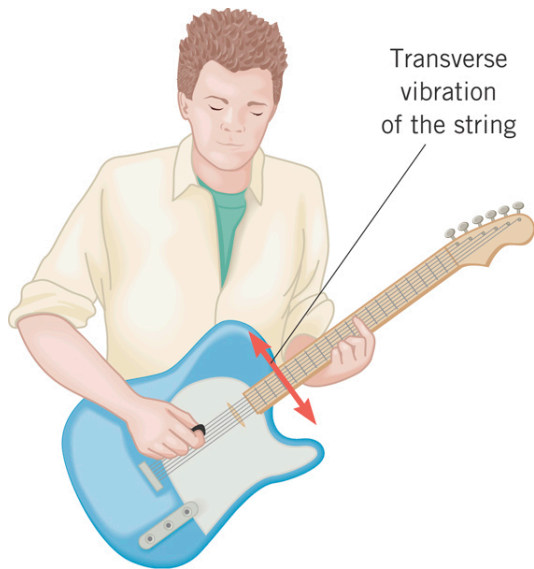
Tension: F

Linear mass density: m/L

16.3 The Speed of a Wave on a String

Example 2 Waves Traveling on Guitar Strings

Transverse waves travel on each string of an electric guitar after the string is plucked. The length of each string between its two fixed ends is 0.628 m, and the mass is 0.208 g for the highest pitched E string and 3.32 g for the lowest pitched E string. Each string is under a tension of 226 N. Find the speeds of the waves on the two strings.



High E

$$v = \sqrt{\frac{F}{m/L}} = \sqrt{\frac{226 \text{ N}}{(0.208 \times 10^{-3} \text{ kg}) / (0.628 \text{ m})}} = 826 \text{ m/s}$$

Low E

$$v = \sqrt{\frac{F}{m/L}} = \sqrt{\frac{226 \text{ N}}{(3.32 \times 10^{-3} \text{ kg}) / (0.628 \text{ m})}} = 207 \text{ m/s}$$

16.4 The Mathematical Description of a Wave

Traveling wave:

Moving toward +x

$$y = A \sin\left(2\pi ft - 2\pi \frac{x}{\lambda}\right)$$
$$= A \sin(\omega t - kx)$$

$$\omega = 2\pi f, k = \frac{2\pi}{\lambda}$$

Moving toward -x

$$y = A \sin\left(2\pi ft + 2\pi \frac{x}{\lambda}\right)$$
$$= A \sin(\omega t + kx)$$

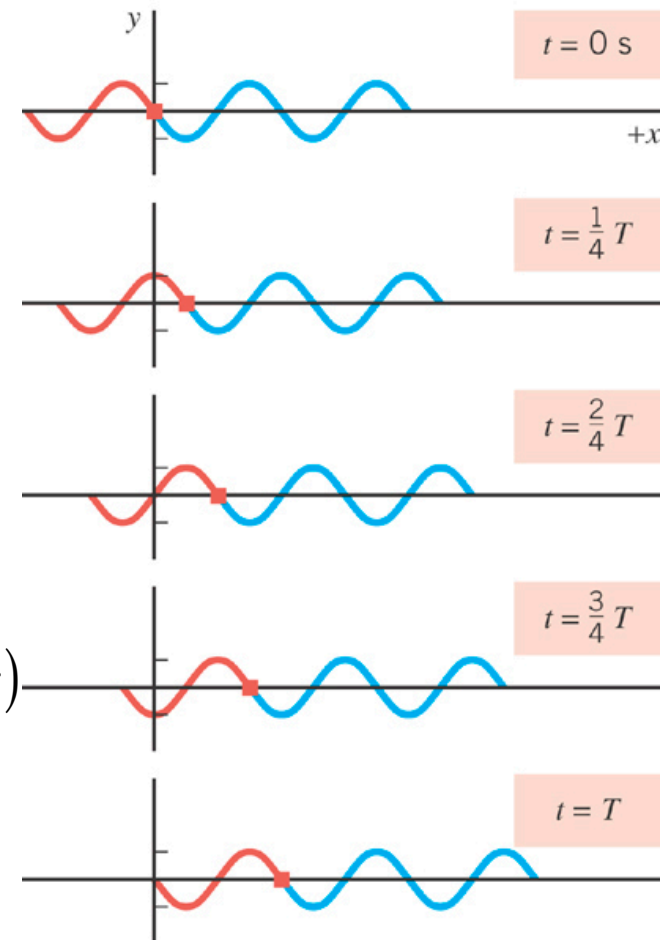
$$y_{t=0} = A \sin(0 - kx)$$

$$y_{t=T/4} = A \sin\left(\frac{\pi}{2} - kx\right)$$

$$y_{t=T/2} = A \sin(\pi - kx)$$

$$y_{t=3T/4} = A \sin\left(\frac{3\pi}{2} - kx\right)$$

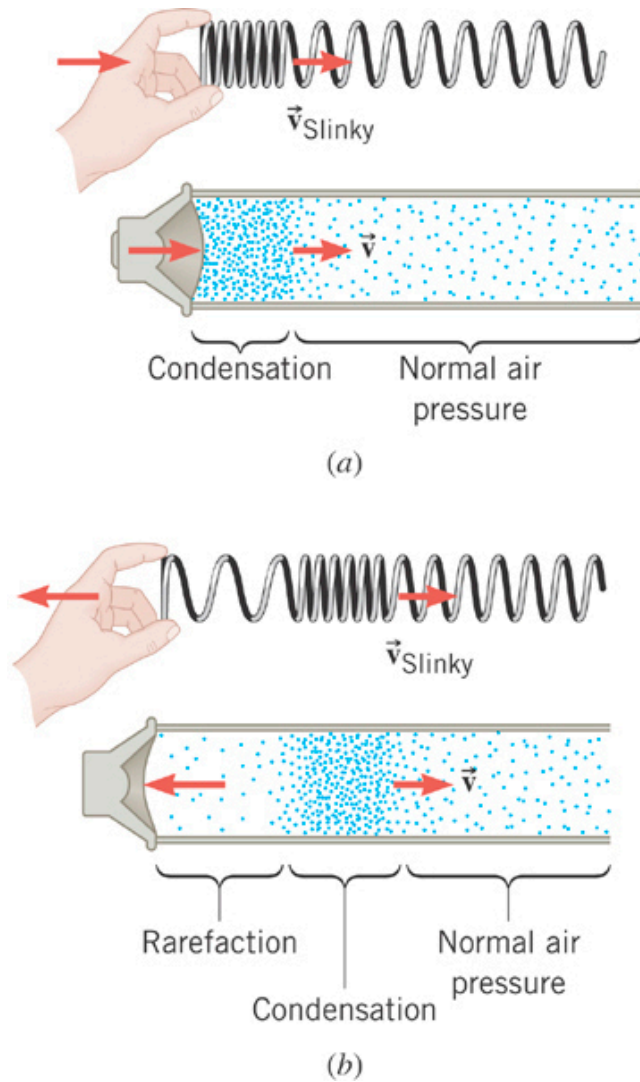
$$y_{t=T} = A \sin(2\pi - kx)$$



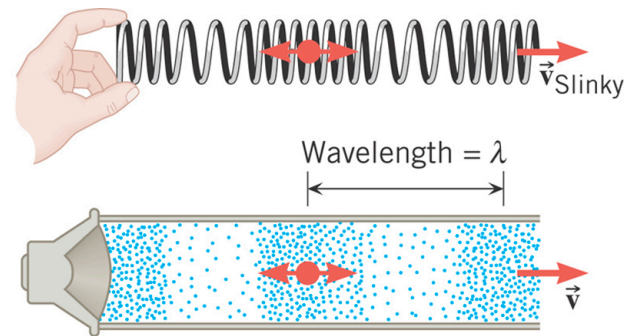
$$\omega = 2\pi f, k = \frac{2\pi}{\lambda}$$

16.5 The Nature of Sound Waves

LONGITUDINAL SOUND WAVES



The distance between adjacent condensations is equal to the wavelength of the sound wave.



16.5 The Nature of Sound Waves

THE FREQUENCY OF A SOUND WAVE

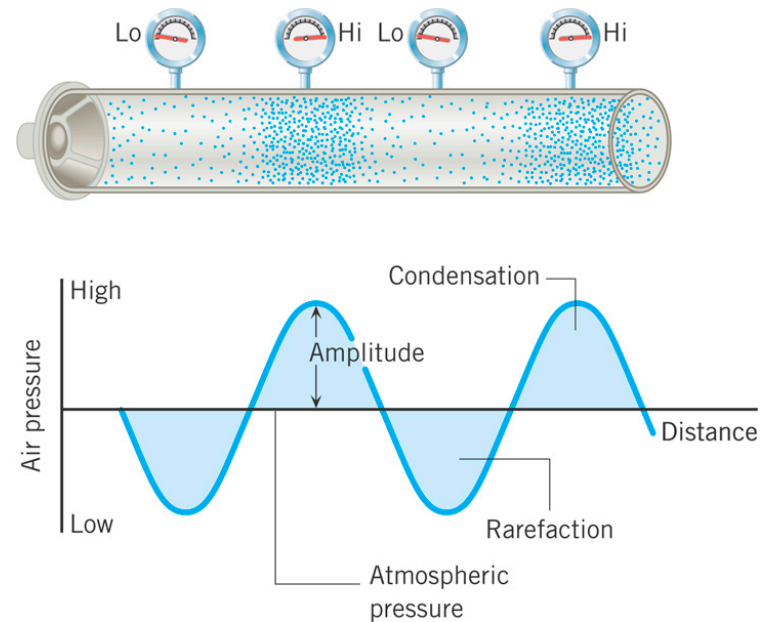
The **frequency** is the number of cycles per second.

A sound with a single frequency is called a **pure tone**.

The brain interprets the frequency in terms of the subjective quality called **pitch**.

THE AMPLITUDE OF A SOUND WAVE

Loudness is an attribute of a sound that depends primarily on the pressure amplitude of the wave.



16.6 The Speed of Sound

Sound travels through gases, liquids, and solids at considerably different speeds.

Compressions and rarefactions can move from place to place by collisions of gas molecules

**Ideal Gas
molecular
velocities**

$$v_{rms} = \sqrt{\frac{3kT}{m}}$$

Sound wave speed

$$v_{sound} = \sqrt{\frac{\gamma kT}{m}}$$

$$k = 1.38 \times 10^{-23} \text{ J/K}$$

$$\gamma = \frac{5}{3} \text{ or } \frac{7}{3}$$

Table 16.1 Speed of Sound in Gases, Liquids, and Solids

| Substance | Speed (m/s) |
|-----------------------|-------------|
| Gases | |
| Air (0 °C) | 331 |
| Air (20 °C) | 343 |
| Carbon dioxide (0 °C) | 259 |
| Oxygen (0 °C) | 316 |
| Helium (0 °C) | 965 |
| Liquids | |
| Chloroform (20 °C) | 1004 |
| Ethyl alcohol (20 °C) | 1162 |
| Mercury (20 °C) | 1450 |
| Fresh water (20 °C) | 1482 |
| Seawater (20 °C) | 1522 |
| Solids | |
| Copper | 5010 |
| Glass (Pyrex) | 5640 |
| Lead | 1960 |
| Steel | 5960 |

16.6 The Speed of Sound

Table 11.1 Mass Densities^a
of Common Substances

| Substance | Mass Density ρ (kg/m ³) |
|----------------------|---|
| Solids | |
| Aluminum | 2700 |
| Brass | 8470 |
| Concrete | 2200 |
| Copper | 8890 |
| Diamond | 3520 |
| Gold | 19 300 |
| Ice | 917 |
| Iron (steel) | 7860 |
| Lead | 11 300 |
| Quartz | 2660 |
| Silver | 10 500 |
| Wood (yellow pine) | 550 |
| Liquids | |
| Blood (whole, 37 °C) | 1060 |
| Ethyl alcohol | 806 |
| Mercury | 13 600 |
| Oil (hydraulic) | 800 |
| Water (4 °C) | 1.000×10^3 |
| Gases | |
| Air | 1.29 |
| Carbon dioxide | 1.98 |
| Helium | 0.179 |
| Hydrogen | 0.0899 |
| Nitrogen | 1.25 |
| Oxygen | 1.43 |

^a Unless otherwise noted, densities are given at 0 °C and 1 atm pressure.

LIQUIDS

$$v = \sqrt{\frac{B_{ad}}{\rho}}$$

Table 10.3 Values for the Bulk Modulus of Solid and Liquid Materials

| Material | Bulk Modulus B [N/m ² (=Pa)] |
|----------------|--|
| Solids | |
| Aluminum | 7.1×10^{10} |
| Brass | 6.7×10^{10} |
| Copper | 1.3×10^{11} |
| Diamond | 4.43×10^{11} |
| Lead | 4.2×10^{10} |
| Nylon | 6.1×10^9 |
| Osmium | 4.62×10^{11} |
| Pyrex glass | 2.6×10^{10} |
| Steel | 1.4×10^{11} |
| Liquids | |
| Ethanol | 8.9×10^8 |
| Oil | 1.7×10^9 |
| Water | 2.2×10^9 |

SOLID BARS

$$v = \sqrt{\frac{Y}{\rho}}$$

Table 10.1 Values for the Young's Modulus of Solid Materials

| Material | Young's Modulus Y (N/m ²) |
|-------------|--|
| Aluminum | 6.9×10^{10} |
| Bone | |
| Compression | 9.4×10^9 |
| Tension | 1.6×10^{10} |
| Brass | 9.0×10^{10} |
| Brick | 1.4×10^{10} |
| Copper | 1.1×10^{11} |
| Mohair | 2.9×10^9 |
| Nylon | 3.7×10^9 |
| Pyrex glass | 6.2×10^{10} |
| Steel | 2.0×10^{11} |
| Teflon | 3.7×10^8 |
| Titanium | 1.2×10^{11} |
| Tungsten | 3.6×10^{11} |

16.7 Sound Intensity

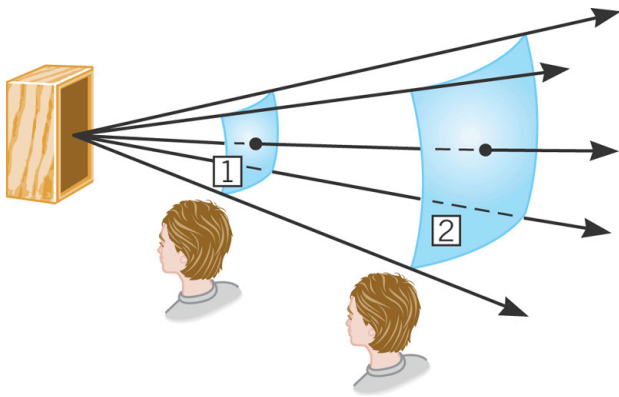
The amount of energy transported per second is called the **power** of the wave.

The **sound intensity** is defined as the power that passes perpendicularly through a surface divided by the area of that surface.

$$I = P/A; \quad \text{power: } P \text{ (watts)}$$

Example 6 Sound Intensities

$12 \times 10^{-5} \text{ W}$ of sound power passed through the surfaces labeled 1 and 2. The areas of these surfaces are 4.0 m^2 and 12 m^2 . Determine the sound intensity at each surface.



$$I_1 = \frac{P}{A_1} = \frac{12 \times 10^{-5} \text{ W}}{4.0 \text{ m}^2} = 3.0 \times 10^{-5} \text{ W/m}^2$$

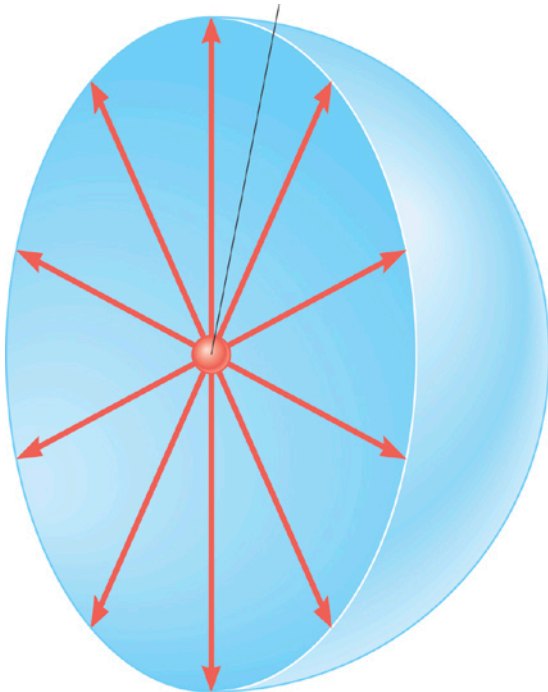
$$I_2 = \frac{P}{A_2} = \frac{12 \times 10^{-5} \text{ W}}{12 \text{ m}^2} = 1.0 \times 10^{-5} \text{ W/m}^2$$

16.7 Sound Intensity

For a 1000 Hz tone, the smallest sound intensity that the human ear can detect is about $1 \times 10^{-12} \text{ W/m}^2$. This intensity is called the ***threshold of hearing***.

On the other extreme, continuous exposure to intensities greater than 1 W/m^2 can be painful.

If the source emits sound *uniformly in all directions*, the intensity depends on the distance from the source in a simple way.



$$I = \frac{P}{4\pi r^2}$$

Intensity depends inversely on the **square of the distance** from the source.

16.8 Decibels

The **decibel** (dB) is a measurement unit used when comparing two sound Intensities.

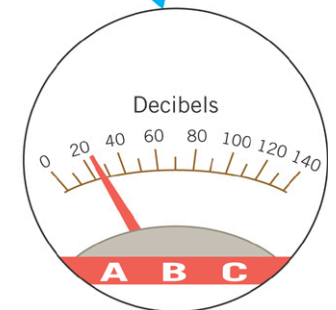
Human hearing mechanism responds to sound **intensity level** , logarithmically.

$$\beta = (10 \text{ dB}) \log\left(\frac{I}{I_o}\right)$$

Note that $\log(1) = 0$

dB (decibel)

$$I_o = 1.00 \times 10^{-12} \text{ W/m}^2$$



| | Intensity I (W/m^2) | Intensity Level β (dB) |
|-------------------------------|----------------------------------|------------------------------|
| Threshold of hearing | 1.0×10^{-12} | 0 |
| Rustling leaves | 1.0×10^{-11} | 10 |
| Whisper | 1.0×10^{-10} | 20 |
| Normal conversation (1 meter) | 3.2×10^{-6} | 65 |
| Inside car in city traffic | 1.0×10^{-4} | 80 |
| Car without muffler | 1.0×10^{-2} | 100 |
| Live rock concert | 1.0 | 120 |
| Threshold of pain | 10 | 130 |

16.8 Decibels

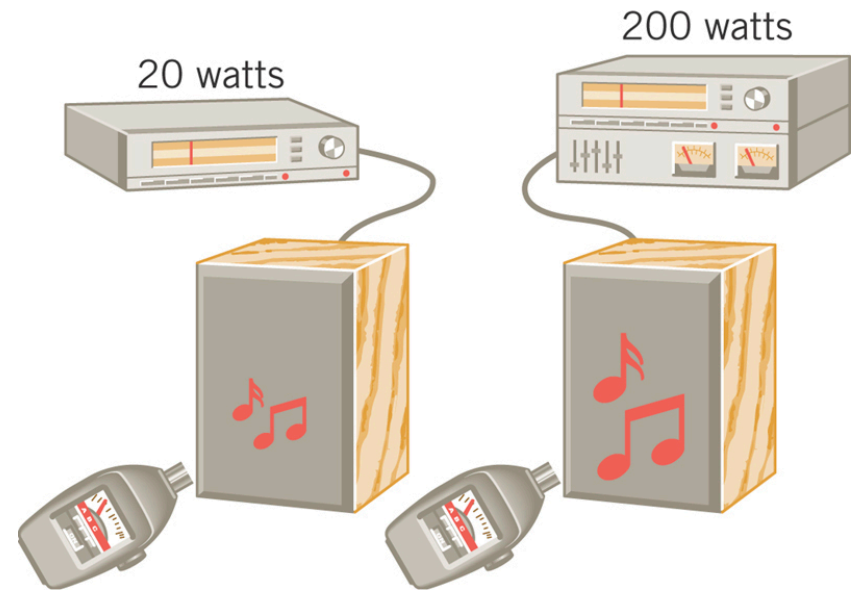
Example 9 Comparing Sound Intensities

Audio system 1 produces a sound intensity level of 90.0 dB, and system 2 produces an intensity level of 93.0 dB. Determine the ratio of intensities.

$$\beta = (10 \text{ dB}) \log\left(\frac{I}{I_o}\right)$$

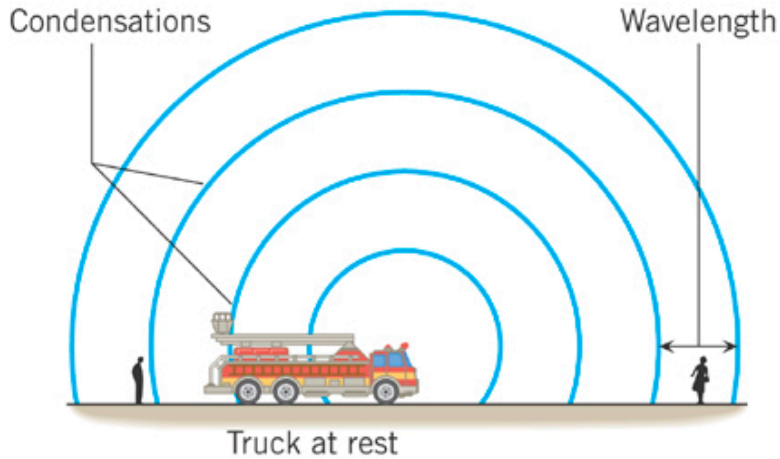
$$\begin{aligned} 90 \text{ dB} &= (10 \text{ dB}) \log(I/I_o) \\ \log(I/I_o) &= 9; \\ I &= I_o \times 10^9 = (1 \times 10^{-12} \text{ W/m}^2) \times 10^9 \\ &= 1 \times 10^{-3} \text{ W/m}^2 \end{aligned}$$

$$\begin{aligned} 93 \text{ dB} &= (10 \text{ dB}) \log(I/I_o) \\ \log(I/I_o) &= 9.3; \\ I &= I_o \times 10^{9.3} = (1 \times 10^{-12} \text{ W/m}^2) \times 10^{9.3} \\ &= 1 \times 10^{-3.3} \text{ W/m}^2 = 1 \times 10^{-3} (10^{0.3}) \text{ W/m}^2 \\ &= 1 \times 10^{-3} (2) \text{ W/m}^2 = 2 \times 10^{-3} \text{ W/m}^2 \end{aligned}$$

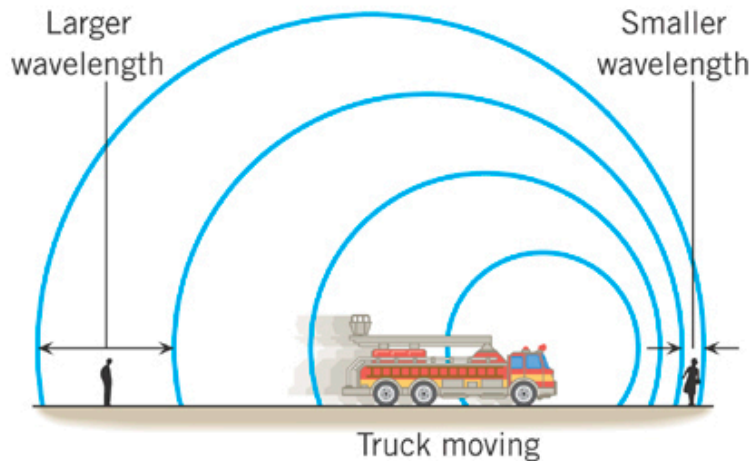


$$\begin{aligned} 93 \text{ dB} &= 90 \text{ dB} + 3 \text{ dB} \\ \text{Adding 3 dB results in a factor of 2} \\ 3 \text{ dB} &= (10 \text{ dB}) \log(I_2/I_1) \\ 0.3 &= \log(I_2/I_1); \\ I_2 &= 10^{0.3} I_1 = 2 I_1 \end{aligned}$$

16.9 The Doppler Effect



(a)



(b)

The **Doppler effect** is the change in frequency or pitch of the sound detected by an observer because the sound source and the observer have different velocities with respect to the medium of sound propagation.

**SOURCE (s) MOVING AT v_s
TOWARD OBSERVER (o)**

$$f_o = f_s \left(\frac{1}{1 - v_s/v} \right)$$

**SOURCE (s) MOVING AT v_s
AWAY FROM OBSERVER (o)**

$$f_o = f_s \left(\frac{1}{1 + v_s/v} \right)$$