Summary of the basic things covered in the lecture (Phy 410) Feb 18, 2009

- 1. Microstates and Macrostates (*N* spins and *N* quantum harmonic oscillators)
- 2. Multiplicity of a given <u>macrostate</u> (g(N,s)) or g(N,U), or g(N,M) for magnets, g(N,n) or $g(N,\varepsilon)$ for oscillators where U is the energy, M is total magnetic moment).

Spins or magnets with 2 states/magnet

$$g(N,s) = \frac{N!}{(\frac{N}{2} + s)!(\frac{N}{2} - s)!}$$
$$\sum_{s} g(N,s) = 2^{N}$$
$$M = 2sm, U = -MB = -2smB$$

N quantum harmonic oscillators, each with same frequency ω , energy $\varepsilon_s = s\hbar\omega$. Multiplicity factor of the <u>macrostate</u> with energy $\varepsilon = n\hbar\omega$

$$g(N,n) = \frac{(N-1+n)!}{(N-1)! \, n!}$$

3. Multiplicity of two systems (N_1 and N_2) in thermal contact with total energy U fixed.

$$g(N,U) = \sum_{U_1,U_2;U_1+U_2=U} g_1(N_1,U_1) \bullet g_2(N_2,U_2)$$

For spin systems (2s is the spin excess etc)

$$g(N,s) = \sum_{s_1, s_2, s_1 + s_2 = s} g_1(N_1, s_1) \bullet g_2(N_2, s_2)$$

4. Concept of entropy, thermal equilibrium and temperature (for system with fixed energy, closed system)

$$\sigma(N,U) = \ln g(N,U)$$

$$\frac{1}{\tau} = \left(\frac{\partial \sigma}{\partial U}\right)_{N}$$

Equilibrium corresponds to the most probable distribution when the two systems have the same temperature. This corresponds to a particular division of energy between the two systems, \overline{U}_1 and $\overline{U}_2 = U - \overline{U}_1$.

$$\frac{1}{\tau_1} = \left(\frac{\partial \sigma_1}{\partial U_1}\right)_{N_1, U_1 = \overline{U}_1} = \left(\frac{\partial \sigma_2}{\partial U_2}\right)_{N_2, U_2 = \overline{U}_2} = \frac{1}{\tau_2}$$

At equilibrium

$$g(N,U) = g_1(N_1, \overline{U}_1) \bullet g_2(N_2, \overline{U}_2)$$

$$\sigma(N,U) = \sigma_1(N_1, \overline{U}_1) + \sigma_1(N_2, \overline{U}_2)$$

For two independent systems Multiplicity Factor is a PRODUCT of multiplicity factor for each component. IN CONTRAST Entropy is additive.

5. For thermally open systems (system in equilibrium with a reservoir at temperature τ), probability of a given microstate s with energy ε_s is given by

$$P(\varepsilon_s) = \frac{e^{-\varepsilon_s/\tau}}{Z};$$

$$Z = \sum e^{-\varepsilon_s/\tau}; Partition function$$

 $Z = \sum_{s} e^{-\varepsilon_{s}/\tau}; Partition function$ Two non interacting systems A and B the partition function $Z_{A+B} = Z_{A} \bullet Z_{B}$

For N non interacting systems (spins, osillators, particles etc) $Z_N = Z_1 \bullet Z_1 \bullet Z_1 \bullet Z_1 \dots = (Z_1)^N \cdot Z_1$ will also depend on the physical system. Also for indistinguishable particles we need to divide Z_N by N!

6. Helmholtz free energy $F = U - \tau \sigma = -\tau \ln Z$

$$U = \sum_{s} \varepsilon_{s} P(\varepsilon_{s}) = \tau^{2} \frac{\partial}{\partial \tau} \ln Z$$

$$C_{V} = \left(\frac{\partial U}{\partial T}\right)_{V,N}$$

For two non-interacting systems A and B, $F_{A+B} = F_A + F_B$

7. Thermodynamic identities (when N is constant)

$$\tau d\sigma = dU + p dV$$
$$dF = -\sigma d\tau - p dV$$

You can use these identities to calculate temperature, entropy and pressure under various types of external constraints.

8. Maxwell velocity (\vec{v}) and speed (v) distribution

$$P(\vec{v}) = \left(\frac{M}{2\pi\tau}\right)^{3/2} e^{\frac{M\vec{v}^2}{2\tau}}; D(v) = \left(\frac{M}{2\pi\tau}\right)^{3/2} 4\pi v^2 e^{\frac{Mv^2}{2\tau}}$$

- 9. Equipartition principle in classical systems
- 10. Thermal properties of black body radiation (Quantum Harmonic Oscillator model). Average total energy U/V and energy density u_{ω} as a function of the radiation frequency ω .