# PHY 410 - Spring 2010 <br> Exam \#2 <br> (1 Hour) 

## PLEASE WAIT UNTIL YOU ARE TOLD TO BEGIN THE EXAM

While waiting, carefully fill in the information requested below
$\qquad$
Your Name:

Your Student Number:

There are 4 problems. Please answer them all showing your work clearly (for partial credit).

## USEFUL CONSTANTS AND INTEGRALS

> Thermal wavelength $\lambda_{t h}=\sqrt{\frac{2 \pi \hbar^{2}}{M \tau}}$
> Quantum concentration $n_{Q}=\left(\frac{M \tau}{2 \pi \hbar^{2}}\right)^{3 / 2}=\frac{1}{\lambda_{t h}^{3}}$
> Boltzmann constant $k_{B}=1.38066 \times 10^{-23} \mathrm{JK}^{-1}$
> Planck's constant $\hbar=1.05459 \times 10^{-34} \mathrm{JS}$

Energy of a photon in mode $\vec{n}$ inside a cubic box of volume $L^{3}: \hbar \omega_{\vec{n}}=\hbar n \pi c / L$

$$
\int_{0}^{\infty} \frac{x^{3}}{e^{x}-1} d x=\frac{\pi^{4}}{15}
$$

## Problem 1 (10 points)

A solid surface has 2 binding sites to each of which one Ar atom can be bound with energy $\varepsilon$. Ar atoms bound to different sites do not interact with each other. The solid surface is in contact with a bath of Ar atoms at temperature $\tau$ and chemical potential $\mu$. Let $\lambda=e^{\mu / \tau}$ be the absolute activity of the Ar atoms.
(1) What is the probability of finding 2 Ar atoms bound to the surface?
(2) What is the probability that the surface has no bound Ar atoms?
(3) What is the probability that the surface has 1 bound Ar atom?
(4) What is the average number of Ar atoms bound to the surface?

$$
\begin{aligned}
& \widetilde{Z}=\sum_{N=0}^{2} \lambda^{N} \sum_{s(N)} e^{-\varepsilon_{s(N)} / \tau} \\
& =1+2 \lambda e^{-\varepsilon / \tau}+\lambda^{2} e^{-2 \varepsilon / \tau} \\
& \text { (1) } P(N=2)=\frac{\lambda^{2} e^{-2 \varepsilon / \tau}}{\widetilde{Z}} \\
& \text { (2) } P(N=0)=\frac{1}{\widetilde{Z}} \\
& \text { (3) } P(N=1)=\frac{2 \lambda e^{-\varepsilon / \tau}}{\widetilde{Z}} \\
& \text { (4) }\langle N\rangle=\sum_{N=0}^{2} N P(N)=\frac{2 \lambda e^{-\varepsilon / \tau}+2 \lambda^{2} e^{-2 \varepsilon / \tau}}{\widetilde{Z}}=\frac{2 \lambda e^{-\varepsilon / \tau}+2 \lambda^{2} e^{-2 \varepsilon / \tau}}{1+2 \lambda e^{-\varepsilon / \tau}+\lambda^{2} e^{-2 \varepsilon / \tau}}
\end{aligned}
$$

## Problem 2 ( 15 points)

The energy density of black body radiation confined in a box of volume $V$ at temperature $\tau$ is given by

$$
u=\frac{U}{V}=\frac{\pi^{2}}{15 c^{3} \hbar^{3}} \tau^{4}
$$

(i) What is its heat capacity per unit volume? (3 points)
(ii) Starting from the expression relating heat capacity and entropy $\sigma$ (use thermodynamic identity) calculate $\sigma$. (6 points)
(iii) If the black body radiation undergoes an adiabatic expansion by a factor of 8 then by what factor does its temperature change? (3 points)
(iv) If the temperature of the black body increases by a factor of 2 then by what factor its entropy/volume changes? (3 points)
(i) $U=V \frac{\pi^{2}}{15 c^{3} \hbar^{3}} \tau^{4}$
$C_{V}=\left(\frac{\partial U}{\partial \tau}\right)_{V}=V \frac{4 \pi^{2}}{15 c^{3} \hbar^{3}} \tau^{3}$
(ii) $C_{V}=\left(\frac{\partial U}{\partial \tau}\right)_{V}=\tau\left(\frac{\partial \sigma}{\partial \tau}\right)_{V}$
$\int_{0}^{\sigma} d \sigma=\int \frac{C_{V}\left(\tau^{\prime}\right)}{\tau^{\prime}} d \tau^{\prime}$
$\sigma=V \frac{4 \pi^{2}}{45 c^{3} \hbar^{3}} \tau^{3} ;$ Used $\sigma(\tau=0)=0$
(iii) Adiabatic $\operatorname{Pr}$ ocess $\sigma=$ const
$V \tau^{3}=$ const
If $V$ increases by $2^{3}$ then $\tau$ shoud decrease by 2
(iv) If $\tau$ increases by 2 then
$\sigma / V$ increases by $2^{3}=8$

## Problem 3 (10 points)

(i) A classical ideal gas of $N$ atoms of mass $M$ is confined inside a cubic box of volume $V=L^{3}$ at temperature $\tau$. What is the chemical potential of the gas?(4 points)
(ii) Each atom has a charge $Q$. A potential $V$ is applied to the top plate of the box and the bottom plate is kept at zero potential. Derive an expression for the density of the gas as a function of the distance from the bottom plate, $n(z)$, where $z=0(L)$ for the bottom (top) plate.
Treat the charged gas as ideal and assume that the electric field inside is uniform. (6 points)
(i) $\mu=\tau \ln \left(\frac{n}{n_{Q}}\right) ; n=N / V$ or $<N>/ V, n_{Q}=\frac{1}{\lambda_{t h}^{3}}$
(ii) In the presence of external potential energy (due to charge and potentials $\mu_{t o t}=\mu_{\mathrm{int}}+\mu_{e x t}=\tau \ln \left(\frac{n(z)}{n_{Q}}\right)+Q E z=\tau \ln \left(\frac{n(z)}{n_{Q}}\right)+\frac{Q V z}{L}$
Since in diffusive equilibrium $\mu_{\text {tot }}=$ const $=C$
$\tau \ln \left(\frac{n(z)}{n_{Q}}\right)+\frac{Q V z}{L}=C$
$n(z)=n_{Q} e^{\frac{c-Q V z / L}{\tau}}=n(0) e^{-\frac{Q V z}{L \tau}}$

## Problem 4 ( 15 points)

(i) Write down the Fermi-Dirac and Bose-Einstein distribution functions for the occupation of a single orbital of energy $\varepsilon$ in terms of temperature $\tau$ and chemical potential $\mu$ (Don't derive it). Plot these as a function of $\varepsilon$.
(2 points)
(ii) What is the distribution function in the classical regime? When is the classical regime applicable? (3 points)
(iii) Using the above classical limit of the distribution function find $\mu$ in terms of the average number of particles $\langle N\rangle$, volume $V$, and the quantum concentration $n_{Q}$ for a 3-dimensional ideal classical gas. Use this equation to find the Helmholtz's free energy.
(8 points)
(iv) What is the value of $\mu$ when the concentration $\frac{\langle N\rangle}{V}=n_{Q}$ ? (2 points)
(i)

$$
\begin{aligned}
& f_{F D}(\varepsilon)=\langle N(\varepsilon)\rangle=\frac{1}{e^{(\varepsilon-\mu) / \tau}+1} \\
& f_{B E}(\varepsilon)=\langle N(\varepsilon)\rangle=\frac{1}{e^{(\varepsilon-\mu) / \tau}-1}
\end{aligned}
$$

(ii)

$$
f_{\text {classical }}(\varepsilon)=e^{-(\varepsilon-\mu) / \tau} ; \text { when } e^{(\varepsilon-\mu) / \tau} \gg 1 ;\left(n \ll n_{Q}\right)
$$

(iii)

$$
\begin{aligned}
& \langle N\rangle=\sum_{\varepsilon}\langle N(\varepsilon)\rangle=e^{\mu / \tau} \sum_{\varepsilon} e^{-\varepsilon / \tau}=e^{\mu / \tau} Z_{1}=e^{\mu / \tau} \frac{V}{\lambda_{t h}^{3}} \\
& \mu=\tau \ln \left(\frac{\left\langle N>\lambda_{t h}^{3}\right.}{V}\right) \equiv \tau \ln \left(\frac{N \lambda_{t h}^{3}}{V}\right)=\tau \ln \left(\frac{n}{n_{Q}}\right)
\end{aligned}
$$

Since chemical potentil $\mu$ and Helmholtz Free energy $F(N, \tau, V)$ are related by $\mu=\left(\frac{\partial F}{\partial N}\right)_{\tau, V}$
$F(N, \tau, V)=\int_{0}^{N} \mu\left(N^{\prime}, \tau, V\right) d N^{\prime}=N \tau\left[\ln \frac{n}{n_{Q}}-1\right]$
(iv)When $n=n_{Q}, \mu=\tau \ln (1)=0$

