PHY 410 – Spring 2010 Exam #2 (1 Hour)

PLEASE WAIT UNTIL YOU ARE TOLD TO BEGIN THE EXAM

While waiting, carefully fill in the information requested below

Your Name:.....

Your Student Number:

There are 4 problems. Please answer them all showing your work clearly (for partial credit).

USEFUL CONSTANTS AND INTEGRALS

Thermal wavelength $\lambda_{th} = \sqrt{\frac{2\pi\hbar^2}{M\tau}}$ Quantum concentration $n_Q = \left(\frac{M\tau}{2\pi\hbar^2}\right)^{3/2} = \frac{1}{\lambda_{th}^3}$ Boltzmann constant $k_B = 1.38066 \times 10^{-23} J K^{-1}$ Planck's constant $\hbar = 1.05459 \times 10^{-34} J s$ Energy of a photon in mode \vec{n} inside a cubic box of volume L^3 : $\hbar \omega_{\vec{n}} = \hbar n \pi c / L$ $\int_{0}^{\infty} \frac{x^3}{e^x - 1} dx = \frac{\pi^4}{15}$ 1

Problem 1 (10 points)

A solid surface has 2 binding sites to each of which one Ar atom can be bound with energy ε . Ar atoms bound to different sites do not interact with each other. The solid surface is in contact with a bath of Ar atoms at temperature τ and chemical potential μ . Let $\lambda = e^{\mu/\tau}$ be the absolute activity of the Ar atoms.

(1) What is the probability of finding 2 Ar atoms bound to the surface?

(2) What is the probability that the surface has no bound Ar atoms?

(3) What is the probability that the surface has 1 bound Ar atom?

(4) What is the average number of Ar atoms bound to the surface?

$$\begin{split} \widetilde{Z} &= \sum_{N=0}^{2} \lambda^{N} \sum_{s(N)} e^{-\varepsilon_{s(N)}/\tau} \\ &= 1 + 2\lambda e^{-\varepsilon/\tau} + \lambda^{2} e^{-2\varepsilon/\tau} \\ (1) \ P(N=2) &= \frac{\lambda^{2} e^{-2\varepsilon/\tau}}{\widetilde{Z}} \\ (2) \ P(N=0) &= \frac{1}{\widetilde{Z}} \\ (3) \ P(N=1) &= \frac{2\lambda e^{-\varepsilon/\tau}}{\widetilde{Z}} \\ (4) < N >= \sum_{N=0}^{2} NP(N) = \frac{2\lambda e^{-\varepsilon/\tau} + 2\lambda^{2} e^{-2\varepsilon/\tau}}{\widetilde{Z}} \\ &= \frac{2\lambda e^{-\varepsilon/\tau} + 2\lambda^{2} e^{-2\varepsilon/\tau}}{1 + 2\lambda e^{-\varepsilon/\tau} + \lambda^{2} e^{-2\varepsilon/\tau}} \end{split}$$

Problem 2 (15 points)

The energy density of black body radiation confined in a box of volume V at temperature τ is given by

$$u = \frac{U}{V} = \frac{\pi^2}{15 c^3 \hbar^3} \tau^4$$

(i) What is its heat capacity per unit volume?(3 points)

(ii) Starting from the expression relating heat capacity and entropy σ (use thermodynamic identity) calculate σ . (6 points)

(iii) If the black body radiation undergoes an adiabatic expansion by a factor of 8 then by what factor does its temperature change? (3 points)

(iv) If the temperature of the black body increases by a factor of 2 then by what factor its entropy/volume changes? (3 points)

(i)
$$U = V \frac{\pi^2}{15c^3\hbar^3} \tau^4$$

 $C_V = \left(\frac{\partial U}{\partial \tau}\right)_V = V \frac{4\pi^2}{15c^3\hbar^3} \tau^3$
(ii) $C_V = \left(\frac{\partial U}{\partial \tau}\right)_V = \tau \left(\frac{\partial \sigma}{\partial \tau}\right)_V$
 $\int_0^{\sigma} d\sigma = \int \frac{C_V(\tau')}{\tau'} d\tau'$
 $\sigma = V \frac{4\pi^2}{45c^3\hbar^3} \tau^3$; Used $\sigma(\tau = 0) = 0$
(iii) Adiabatic Process $\sigma = const$
 $V\tau^3 = const$
If V increases by 2³ then τ shoud decrease by 2
(iv) If τ increases by 2³ = 8

Problem 3 (10 points)

(i) A classical ideal gas of N atoms of mass M is confined inside a cubic box of volume $V=L^3$ at temperature τ . What is the chemical potential of the gas?(4 points)

(ii) Each atom has a charge Q. A potential V is applied to the top plate of the box and the bottom plate is kept at zero potential. Derive an expression for the density of the gas as a function of the distance from the bottom plate, n(z), where z=O(L) for the bottom (top) plate.

Treat the charged gas as ideal and assume that the electric field inside is uniform. (6 points)

(i)
$$\mu = \tau \ln\left(\frac{n}{n_Q}\right); n = N/V \text{ or } < N > /V, n_Q = \frac{1}{\lambda_{th}^3}$$

(ii) In the presence of external potential energy (due to charge and potentials

$$\mu_{tot} = \mu_{int} + \mu_{ext} = \tau \ln\left(\frac{n(z)}{n_Q}\right) + QEz = \tau \ln\left(\frac{n(z)}{n_Q}\right) + \frac{QVz}{L}$$

Since in diffusive equilibrium $\mu_{tot} = const = C$

$$\tau \ln\left(\frac{n(z)}{n_Q}\right) + \frac{QVz}{L} = C$$
$$n(z) = n_Q e^{\frac{c - QVz/L}{\tau}} = n(0)e^{-\frac{QVz}{L\tau}}$$

Problem 4 (15 points)

(i) Write down the Fermi-Dirac and Bose-Einstein distribution functions for the occupation of a single orbital of energy ε in terms of temperature τ and chemical potential μ (Don't derive it). Plot these as a function of ε .

(2 points)

(ii) What is the distribution function in the classical regime? When is the classical regime applicable? (3 points)

(iii) Using the above classical limit of the distribution function find μ in terms of the average number of particles $\langle N \rangle$, volume V, and the quantum concentration n_Q for a 3-dimensional ideal classical gas. Use this equation to find the Helmholtz's free energy.

(8 points)

(iv) What is the value of μ when the concentration $\frac{\langle N \rangle}{V} = n_Q$?(2 points)

(i)

$$f_{FD}(\varepsilon) = \langle N(\varepsilon) \rangle = \frac{1}{e^{(\varepsilon-\mu)/\tau} + 1}$$

$$f_{BE}(\varepsilon) = \langle N(\varepsilon) \rangle = \frac{1}{e^{(\varepsilon-\mu)/\tau} - 1}$$
(ii)

$$f_{classical}(\varepsilon) = e^{-(\varepsilon-\mu)/\tau}; \text{ when } e^{(\varepsilon-\mu)/\tau} \gg 1; (n < n_Q)$$
(iii)

$$\langle N \rangle = \sum_{\varepsilon} \langle N(\varepsilon) \rangle = e^{\mu/\tau} \sum_{\varepsilon} e^{-\varepsilon/\tau} = e^{\mu/\tau} Z_1 = e^{\mu/\tau} \frac{V}{\lambda_{th}^3}$$

$$\mu = \tau \ln\left(\frac{\langle N \rangle \lambda_{th}^3}{V}\right) = \tau \ln\left(\frac{N\lambda_{th}^3}{V}\right) = \tau \ln\left(\frac{n}{n_Q}\right)$$
Since chemical potentil μ and Helmholtz Free energy $F(N, \tau, V)$ are related by $\mu = \left(\frac{\partial F}{\partial N}\right)_{\tau, V}$

$$F(N, \tau, V) = \int_{0}^{N} \mu(N', \tau, V) dN' = N\tau \left[\ln\frac{n}{n_Q} - 1\right]$$

 $F(N,\tau,V) = \int_{0}^{\infty} \mu(N',\tau,V) dN' = N\tau \left[\ln \frac{1}{n_Q} - 1 \right]$ (*iv*) When $n = n_Q$, $\mu = \tau \ln(1) = 0$