

## PHY 410

Final Examination, Spring 2009

May 4, 2009 (5:45-7:45 p.m.)

**PLEASE WAIT UNTIL YOU ARE TOLD TO BEGIN THE EXAM.**

While waiting, carefully fill in the information requested below

Your Name: \_\_\_\_\_

Your Student Number: \_\_\_\_\_

**DO NOT TURN THIS PAGE UNTIL THE EXAM STARTS**

### USEFUL CONSTANTS AND INTEGRALS

$$\text{Avogadro's Number } N_A = 6.022 \times 10^{23}$$

$$\begin{aligned} \text{Boltzmann's constant } k &= 1.381 \times 10^{-23} \text{ J/K} \\ &= 8.617 \times 10^{-5} \text{ eV/K} \end{aligned}$$

$$\text{Planck constant } \hbar = 1.054 \times 10^{-34} \text{ J}\cdot\text{s}$$

$$\text{Electron charge (magnitude) } e = 1.602 \times 10^{-19} \text{ C}$$

$$\text{Electron mass } m = 9.109 \times 10^{-31} \text{ kg}$$

$$\text{Speed of light } = 2.998 \times 10^8 \text{ m/s}$$

$$1 \text{ atm} = 1.013 \text{ bar}$$

$$1 \text{ bar} = 10^5 \text{ N/m}^2$$

$$1 \text{ eV} = 1.602 \times 10^{-19} \text{ J}$$

**There are 7 problems. To receive full credit for each answer, you must work neatly, show your work and simplify your answer to the extent possible.**

**Problem 1 (5 points)**

A power plant produces 1 GW of electricity at an efficiency of 40%.

(i) What is the rate at which the power plant expels waste heat to the environment?

$$\eta = \frac{W}{Q_h}; W = \eta Q_h = \eta(W + Q_l) \Rightarrow W = \frac{\eta}{1-\eta} Q_l; Q_l = \frac{1-\eta}{\eta} W$$

$$\text{Given } \frac{dW}{dt} = 1GW \text{ we have } \frac{dQ_l}{dt} = \frac{1-\eta}{\eta} \frac{dW}{dt} = \frac{1-0.4}{0.4} 1GW = 1.5GW$$

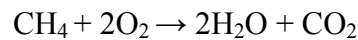
(ii) What is the rate at which it uses up energy at the hot end?

$$\frac{dQ_h}{dt} = \frac{1}{\eta} \frac{dW}{dt} = \frac{1}{0.4} 1GW = 2.5GW$$

$$\text{Alternatively } \frac{dQ_h}{dt} = \frac{dQ_l}{dt} + \frac{dW}{dt} = 1.5GW + 1GW = 2.5GW$$

**Problem 2 (7.5 points)**

Consider a fuel cell that uses methane (gas) as fuel. The reaction is



The difference in Gibb's free energy (final – initial) is -800 kJ per mole of CH<sub>4</sub>.

(i) Assuming ideal performance, how much electrical work can be produced by the cell for each mole of CH<sub>4</sub>.

$$800 \text{ kJ}$$

(ii) The two steps of this reaction are

At -ve electrode  $\text{CH}_4 + 2\text{H}_2\text{O} \rightarrow \text{CO}_2 + 8\text{H}^+ + 8\text{e}^-$  and At +ve electrode  $2\text{O}_2 + 8\text{H}^+ + 8\text{e}^- \rightarrow 4\text{H}_2\text{O}$

What is the maximum voltage generated by the cell?

$$(8e)N_A V_0 = 800 \text{ kJ}$$

$$eV_0 = \frac{800 \times 10^3 \text{ J}}{8N_A} = \frac{800 \times 10^3 \text{ J}}{8 \times 6 \times 10^{23}} \frac{1 \text{ eV}}{1.6 \times 10^{-19} \text{ J}} = 1.04 \text{ eV}$$

$$\text{Therefore max voltage } V_0 = 1.04 \text{ V}$$

**Problem 3 (10 points)**

An ideal gas of  $N$  bosons of mass  $M$  is in a cubical box of volume  $V$ . The bosons are nonrelativistic. The density of orbitals (or states) is given by

$$D_3(\varepsilon) = \frac{V}{4\pi^2} \left( \frac{2M}{\hbar^2} \right)^{3/2} \varepsilon^{1/2}; \varepsilon \geq 0.$$

(i) Write down an expression for the number of particles in terms of  $D_3(\varepsilon)$  and the Bose-Einstein distribution function  $f_{BE}(\varepsilon)$ .

$$N = \int_0^{\infty} D_3(\varepsilon) \frac{1}{e^{(\varepsilon-\mu)/\tau} - 1} d\varepsilon$$

(ii) Derive an expression for the BE condensation temperature  $\tau_E$  in terms  $M$  and the density

$$N/V. \text{ (Use } \int_0^{\infty} \frac{x^{1/2}}{e^x - 1} dx = 1.306\sqrt{\pi} \text{)}$$

$$N = \int_0^{\infty} D_3(\varepsilon) \frac{1}{e^{\varepsilon/\tau_E} - 1} d\varepsilon = \frac{V}{4\pi^2} \left( \frac{2M}{\hbar^2} \right)^{3/2} \int_0^{\infty} \frac{\varepsilon^{1/2}}{e^{\varepsilon/\tau_E} - 1} d\varepsilon$$

$$\text{Put } x = \frac{\varepsilon}{\tau_E} : N = \frac{V}{4\pi^2} \left( \frac{2M}{\hbar^2} \right)^{3/2} \tau_E^{3/2} \int_0^{\infty} \frac{x^{1/2}}{e^x - 1} dx$$

$$\tau_E = \frac{\hbar^2}{2M} \left( \frac{4\pi^{3/2} N}{1.306 V} \right)^{2/3}$$

(iii) How does  $\tau_E$  change when the mass of the boson doubles?

It reduces by a factor of 2

**Problem 4 (7.5 points)**

The measured constant volume heat capacity of a metal at low temperature ( $\tau \ll \tau_F = \varepsilon_F$  and  $\tau \ll k_B \theta_{Debye}$ ) is given by

$$C_V = A\tau + B\tau^3, \text{ where } A \text{ and } B \text{ are constants.}$$

(i) What are the physical origins of the above two different contributions to  $C_V$ ?

The linear terms comes from the Fermi Dirac statistics of electrons in a metal.

The cubic terms comes from the phonons (lattice vibrations) of the atoms

(ii) What is the temperature dependence of entropy of the metal?

$$C_V = \frac{dQ}{d\tau} = \tau \frac{d\sigma}{d\tau}$$

$$\int d\sigma = \int \frac{C_V(\tau')}{\tau'} d\tau'$$

$$\sigma(\tau) - \sigma(0) = \int_0^\tau \frac{C_V(\tau')}{\tau'} d\tau' = A \int_0^\tau d\tau' + B \int_0^\tau (\tau')^2 d\tau' = A\tau + \frac{B\tau^3}{3}$$

**Usually**

$$\sigma(0) = 0$$

**Problem 5 (10 points)**

Consider a defect in a solid which has two energy levels with energy 0 and  $\varepsilon > 0$ . The degeneracy of the ground level is 2 and that of the excited level is 1. The solid is at a temperature  $\tau$ .

- (i) What is the partition function for this defect?
- (ii) What is its average energy  $U(\tau)$ ? Plot  $U$  as a function of  $\tau$ .
- (iii) What is  $U(\tau)$  at  $\tau = \infty$ ?
- (iv) Find the temperature  $\tau_0$  when the average energy  $U(\tau_0)$  is half the average energy at  $\tau = \infty$ ,  $U(\infty)$ .

$$z = 2 + e^{-\varepsilon/\tau}$$

$$U(\tau) = \frac{\varepsilon e^{-\varepsilon/\tau}}{2 + e^{-\varepsilon/\tau}}$$

$$U(\infty) = \frac{\varepsilon}{3}$$

$$U(\tau_0) = \frac{U(\infty)}{2} = \frac{\varepsilon}{6} = \frac{\varepsilon e^{-\varepsilon/\tau_0}}{2 + e^{-\varepsilon/\tau_0}}$$

$$\text{Solve for } \frac{\varepsilon}{\tau_0} \text{ and then } \tau_0 = -\frac{\varepsilon}{\ln 2/5}$$

**Problem 6 (15 points)**

The energy per unit volume of a black body radiation and the radiation pressure  $p$  are given by

$$u = \frac{U}{V} = A\tau^4; \text{ where } A \text{ is a constant, } p = \frac{u}{3}$$

A gas of photon is used as the working medium of a heat engine. The gas undergoes an isothermal expansion at temperature  $\tau_h$  from volume  $V_1$  to  $V_2 = 2V_1$ .

(i) What is the work done by the photon gas,  $W_{12}$ ? If instead of a photon gas we use an ideal gas of  $N$  Ar atoms, what is the work done in this case?

$$\text{PHOTONS: } W_{12} = \int_1^2 p dV = \frac{1}{3} A \tau_h^4 \int_{V_1}^{V_2} dV = \frac{A \tau_h^4}{3} (V_2 - V_1) = \frac{A \tau_h^4 V_1}{3}$$

$$\text{ARGON ATOMS: } W_{12} = \int_1^2 p dV = N \tau_h \int_{V_1}^{V_2} \frac{dV}{V} = N \tau_h \ln\left(\frac{V_2}{V_1}\right) = N \tau_h \ln(2)$$

(ii) Now the photon gas undergoes an adiabatic expansion from  $(\tau_h, V_2)$  to  $(\tau_l, V_3)$

(where  $\tau_l = \tau_h / 2, V_3 = 2V_2$ ), and the work done by the photon gas is  $W_{23}$ . What is  $W_{23}/W_{12}$ ?

*Start from the 1st law of thermodynamics (Energy conservation)*

*$dU = dQ + dW$  where  $dW$  is the work done ON the gas*

*Work done BY the gas =  $-dW = dQ - dU$*

*For an adiabatic process  $dQ = 0$*

*Work done BY the gas =  $-dW = -dU$*

$$W_{23} = \int_2^3 -dU = U(2) - U(3) = AV_2 \tau_h^4 - AV_3 \tau_l^4$$

*Use result for  $W_{12}$  from (i) above*

$$\frac{W_{23}}{W_{12}} = \frac{AV_2 \tau_h^4 - AV_3 \tau_l^4}{(A \tau_h^4 V_1)/3} = \frac{3V_2}{V_1} \left( 1 - \left( \frac{\tau_l}{\tau_h} \right)^4 \frac{V_3}{V_2} \right)$$

*NOW Plug in the different ratios.*

**Problem 7 (15 points)**

Consider an ideal 2-dimensional gas of non-relativistic electrons (spin  $\frac{1}{2}$ ) of density  $n=N/A$  at zero temperature. The density of states for this system is given by  $D_2(\varepsilon) = \frac{A}{2\pi} \left( \frac{2m}{\hbar^2} \right); \varepsilon \geq 0$ .

(i) Derive an equation relating the Fermi energy  $\varepsilon_F$  as a function of  $n$ , mass  $m$ , and other fundamental constant(s).

$$N = \int_0^{\infty} D_2(\varepsilon) \frac{1}{e^{(\varepsilon-\mu)/\tau} + 1} d\varepsilon = \int_0^{\varepsilon_F} D_2(\varepsilon) d\varepsilon = \frac{A}{2\pi} \left( \frac{2m}{\hbar^2} \right) \varepsilon_F$$

$$\varepsilon_F = \frac{\hbar^2}{2m} \left( \frac{2\pi N}{A} \right)$$

(ii) Calculate the average energy (energy/particle) in terms of the Fermi energy  $\varepsilon_F$ .

$$U = \int_0^{\varepsilon_F} \varepsilon D_2(\varepsilon) d\varepsilon = \frac{A}{2\pi} \left( \frac{2m}{\hbar^2} \right) \frac{\varepsilon_F^2}{2}$$

$$\frac{U}{N} = \frac{\varepsilon_F}{2}$$

(iii) What is the Fermi energy  $\varepsilon_F$  (in electron volts) if  $n = 10^{18} \text{ m}^{-2}$

$$\varepsilon_F = 2\pi \times 10^{18} \frac{1}{\text{m}^2} \times \frac{(1.054 \times 10^{-34} \text{ J}\cdot\text{s})^2}{2 \times 9.1 \times 10^{-31} \text{ kg}} \times \frac{1 \text{ eV}}{1.6 \times 10^{-19} \text{ J}}$$

$$= 0.24 \text{ eV}$$

