

**PHY 410**

**HW# 11**

Assigned: April 14, 2010: Due April 21, 2010

- 11.1 In the first observation of Bose-Einstein condensation with atomic hydrogen a gas of approximately  $2 \times 10^{10}$  atoms was trapped in a region and cooled until its density was  $1.8 \times 10^{14}$  atoms/cm<sup>3</sup>.
- (i) Why can we treat the Hydrogen atom as a boson?
  - (ii) What is the average distance between the hydrogen atoms at the above density
  - (iii) Calculate the condensation temperature  $T_E$  and compare to the measured value of 50  $\mu$ K.
- 11.2 Calculate the condensation temperature for liquid helium-4 <sup>4</sup>He assuming that the atoms in the liquid can be treated as an ideal gas. (A physical justification for this is the large zero-point motion of the light He atoms). The mass density you can use is 0.145 g/cm<sup>3</sup>. Compare this with the experimental value of 2.17K.
- 11.3 Find an expression for the entropy of a 3-dimensional ideal bose gas (mass m) confined in a volume V as a function of temperature  $\tau$  in the region  $0 < \tau < \tau_E$ .
- 11.4 Problem 7.14 of the text. (Kittel and Kroemer)
- 11.5 Problem 8.1 of the text. (Kittel and Kroemer)