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nagytimo@msu

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Nagy,

Tibor

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Keep this exam **CLOSED** until advised by the instructor.

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50 minute long closed book exam.

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Fill out the bubble sheet: last name, first initial, **student number (PID)**. Leave the section, code, form and signature areas empty.

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Three two-sided handwritten 8.5 by 11 help sheets are allowed.

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When done, hand in your **test** and your **bubble sheet**.

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Thank you and good luck!

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Possibly useful constants:

- $g = 9.81 \text{ m/s}^2$
- $G = 6.67 \times 10^{-11} \text{ Nm}^2/\text{kg}^2$
- $\rho_{\text{water}} = 1000 \text{ kg/m}^3 = 1 \text{ kg/l} = 1 \text{ g/cm}^3$
- $1 \text{ atm} = 101.3 \text{ kPa} = 760 \text{ mmHg}$
- $N_A = 6.02 \times 10^{23} \text{ 1/mol}$
- $R = 8.31 \text{ J/(molK)}$
- $k_B = 1.38 \times 10^{-23} \text{ J/K}$
- $0 \text{ }^\circ\text{C} = 273.15 \text{ K}$

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Please, sit in row G.

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1 pt Are you sitting in the seat assigned?

1.A  Yes, I am.

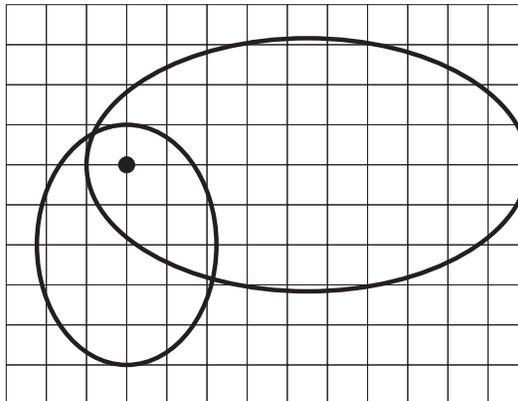
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**4 pt** A 210 kg satellite is orbiting on a circular orbit 5155 km above the Earth's surface. Determine the speed of the satellite. (The mass of the Earth is  $5.97 \times 10^{24}$  kg, and the radius of the Earth is 6370 km.)

(in km/s)

2.    **A**  2.29    **B**  2.68    **C**  3.14    **D**  3.67  
       **E**  4.29    **F**  5.02    **G**  5.88    **H**  6.88

**4 pt** The paths of two small satellites,  $M_L = 6.00$  kg and  $M_R = 9.00$  kg, are shown below, drawn to scale, with  $M_L$  corresponding to the orbit on the left hand side in the figure. They orbit in the same plane around a massive star, as shown below.



The period of  $M_L$  is 26.0 years. Calculate the period of  $M_R$ , in years.

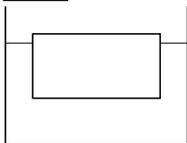
3.    **A**   $4.45 \times 10^1$     **B**   $6.45 \times 10^1$     **C**   $9.36 \times 10^1$   
       **D**   $1.36 \times 10^2$     **E**   $1.97 \times 10^2$     **F**   $2.85 \times 10^2$   
       **G**   $4.14 \times 10^2$     **H**   $6.00 \times 10^2$

**4 pt** Glucose solution is administered to a patient in a hospital. The density of the solution is 1.336 kg/l. If the blood pressure in the vein is 39.2 mmHg, then what is the minimum necessary height of the IV bag above the position of the needle?

(in cm)

4.    **A**  16.9    **B**  22.5    **C**  30.0    **D**  39.9  
       **E**  53.0    **F**  70.5    **G**  93.8    **H**  124.7

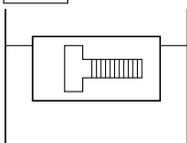
3 pt A large ice cube floats in a glass of water.



What happens to the water level, when the ice cube melts? (No water is lost due to evaporation.)

5.  A The water level will fall.  
 B The water level will rise.  
 C It depends on how much water we have in the glass, and how big the ice cube is.  
 D The water level will not change.

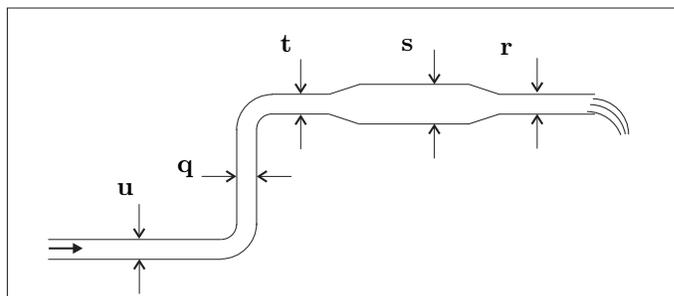
3 pt A large ice cube floats in a glass of water.



There is a steel bolt frozen inside the ice cube. What happens to the water level when all the ice melts? (No water is lost due to evaporation.)

6.  A The water level will fall.  
 B The water level will rise.  
 C Without knowing the mass of the bolt, we cannot answer this question.  
 D The water level will not change.

8 pt The figure illustrates the flow of an ideal fluid through a pipe of circular cross section, with diameters of 1 cm and 2 cm and with different elevations.  $p_x$  is the pressure in the pipe, and  $v_x$  is the speed of the fluid at locations  $x = q, r, s, t, \text{ or } u$ .



7.  $p_r$  is ...  $p_s$   
 A Greater than       B Less than  
 C Equal to

8.  $p_u$  is ...  $p_q$   
 A Greater than       B Less than  
 C Equal to

9.  $v_q$  is ...  $2v_s$   
 A Greater than       B Less than  
 C Equal to

10.  $v_u$  is ...  $v_r$   
 A Greater than       B Less than  
 C Equal to

**4 pt** A rock band uses a wall built out of 37 identical speakers. If one single speaker can produce a sound level of 95.5 dB in the front row area, then what is the sound level produced by the whole wall?

(in dB)

11. **A**  59.3      **B**  69.4      **C**  81.2      **D**  95.0  
**E**  111.2      **F**  130.1      **G**  152.2      **H**  178.1
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**4 pt** A stationary horn emits a sound with a frequency of 205 Hz. A car is moving away from the horn on a straight road with constant speed. If the driver of the car hears the horn at a frequency of 185 Hz, then what is the speed of the car? Use 340 m/s for the speed of the sound.

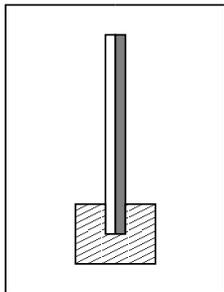
(in m/s)

12. **A**   $2.30 \times 10^1$       **B**   $2.60 \times 10^1$       **C**   $2.94 \times 10^1$   
**D**   $3.32 \times 10^1$       **E**   $3.75 \times 10^1$       **F**   $4.24 \times 10^1$   
**G**   $4.79 \times 10^1$       **H**   $5.41 \times 10^1$
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**4 pt** A pipe is 1.70 m long and it is open at both ends. What are the frequencies of the lowest three harmonics produced by this pipe? The speed of sound is 340 m/s. Only one answer is correct.

13. **A**  100 Hz, 200 Hz, 300 Hz  
**B**  100 Hz, 300 Hz, 500 Hz  
**C**  50 Hz, 100 Hz, 150 Hz  
**D**  50 Hz, 100 Hz, 200 Hz  
**E**  50 Hz, 150 Hz, 250 Hz  
**F**  200 Hz, 400 Hz, 600 Hz  
**G**  200 Hz, 600 Hz, 1000 Hz  
**H**  200 Hz, 300 Hz, 400 Hz
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**3 pt** A bimetallic strip is held fixed at the bottom end as shown in the figure.



The metal on the left has a coefficient of linear heat expansion of  $\alpha_{\text{left}} = 1.50 \times 10^{-5} \text{ 1/K}$ , the metal on the right has  $\alpha_{\text{right}} = 3.85 \times 10^{-5} \text{ 1/K}$ . When the strip is heated, it will ... (complete the sentence)

14. **A**  ... bend right.  
**B**  ... remain straight.  
**C**  ... bend left.

**4 pt** 5.20 liters of Nitrogen gas at  $70.0^\circ\text{C}$  temperature and 2.90 atm pressure contains how many moles?

15. **A**  0.257    **B**  0.291    **C**  0.329    **D**  0.371  
**E**  0.420    **F**  0.474    **G**  0.536    **H**  0.605

**2 pt** A gas bottle contains  $8.44 \times 10^{23}$  Methane molecules at a temperature of 349 K. What is the thermal energy of the gas?

(in J)

16. **A**   $1.22 \times 10^4$     **B**   $1.38 \times 10^4$     **C**   $1.56 \times 10^4$   
**D**   $1.76 \times 10^4$     **E**   $1.99 \times 10^4$     **F**   $2.25 \times 10^4$   
**G**   $2.54 \times 10^4$     **H**   $2.87 \times 10^4$

**2 pt** On average how much energy is stored by ONE degree of freedom for ONE single molecule?

(in J)

17. **A**   $2.41 \times 10^{-21}$     **B**   $3.01 \times 10^{-21}$   
**C**   $3.76 \times 10^{-21}$     **D**   $4.71 \times 10^{-21}$   
**E**   $5.88 \times 10^{-21}$     **F**   $7.35 \times 10^{-21}$   
**G**   $9.19 \times 10^{-21}$     **H**   $1.15 \times 10^{-20}$