

your name(s) \_\_\_\_\_

*Physics 831 Quiz #4*  
Friday, Sep. 29, 2017

Work in groups of three to four.

Consider the equation of state

$$P(\rho, T) = \rho T \left( 1 + \frac{\rho^2}{\rho_0^2} \right) - a \frac{\rho^2}{\rho_0}.$$

1. (5 pts) Solve for the critical density  $\rho_c$  and the critical temperature  $T_c$  in terms of  $\rho_0$  and  $a$ .
2. (5 pts) Using the Maxwell relation,

$$\left. \frac{\partial(P/T)}{\partial\beta} \right|_{N,V} = - \left. \frac{\partial E}{\partial V} \right|_{N,T},$$

Find the energy per particle as a function of temperature and density.

3. (5 pts) If the system expands at constant temperature from volume per particle  $v_a$  to  $v_b$ , find the change in entropy per particle  $s$ .
4. (5 pts) Using  $Ts = e + Pv - \mu$ , find the change in chemical potential between the two points  $a$  and  $b$ .
5. (5 pts) Find the density of the liquid on the coexistence curve as  $T \rightarrow 0$ .
6. (5 pts) Find the latent heat per particle as  $T \rightarrow 0$ .

## Solutions

1.

$$\begin{aligned}
 P &= \rho T \left( 1 + \frac{\rho^2}{\rho_0^2} \right) - a \frac{\rho^2}{\rho_0}, \\
 \partial P &= 0 = T \left( 1 + 3 \frac{\rho^2}{\rho_0^2} \right) - 2a \frac{\rho}{\rho_0}, \\
 \partial^2 P &= 0 = 6T \frac{\rho}{\rho_0^2} - \frac{2a}{\rho_0}, \\
 T_c &= \frac{a\rho_0}{3\rho_c}, \\
 \frac{a\rho_0}{3} \left( 1 + 3 \frac{\rho_c^2}{\rho_0^2} \right) - 2a \frac{\rho_c^2}{\rho_0} &= 0, \\
 \rho_c^2 \left( \frac{3}{\rho_0} - \frac{2}{\rho_0} \right) + \frac{\rho_0}{3} &= 0, \\
 \rho_c &= \frac{\rho_0}{\sqrt{3}}, \quad T_c = \frac{a}{\sqrt{3}}.
 \end{aligned}$$

2.

$$\begin{aligned}
 E &= \frac{3}{2}T - \int_{\infty}^V dv \frac{a\rho^2}{\rho_0} \\
 &= \frac{3}{2}T - \int_0^{\rho} \frac{d\rho}{\rho^2} \frac{a\rho^2}{\rho_0} \\
 &= \frac{3}{2}T - a \frac{\rho}{\rho_0}.
 \end{aligned}$$

3.

$$\begin{aligned}
 dS &= \beta dE + (P/T)dV \\
 \Delta(S/N) &= -\frac{a}{T} \left( \frac{1}{\rho_0 v_b} - \frac{1}{\rho_0 v_a} \right) + \int dv \left[ \rho \left( 1 + \frac{\rho^2}{\rho_0^2} \right) - \frac{a}{T} \frac{\rho^2}{\rho_0} \right] \\
 dv &= -d\rho/\rho^2, \\
 \Delta(S/N) &= \ln(v_b/v_a) - \frac{1}{2} \left( \frac{1}{v_b^2 \rho_0^2} - \frac{1}{v_a^2 \rho_0^2} \right).
 \end{aligned}$$

4.

$$\begin{aligned}
 \Delta\mu &= \Delta e + \Delta(Pv) - T\Delta(S/N) \\
 &= -2a \left( \frac{1}{\rho_0 v_b} - \frac{1}{\rho_0 v_a} \right) + \frac{3T}{2} \left( \frac{1}{\rho_0^2 v_b^2} - \frac{1}{\rho_0^2 v_a^2} \right) - T \ln \left( \frac{v_b}{v_a} \right).
 \end{aligned}$$

5. As  $T \rightarrow 0$ , one needs to find liquid in coexistence with a vapor at zero density, thus at zero pressure.

$$P = \rho T \left( 1 + \frac{\rho^2}{\rho_0^2} \right) - a \frac{\rho^2}{\rho_0}, \quad (1)$$

$$\frac{\rho^2}{\rho_0^2} - \frac{a\rho}{T\rho_0} + 1 = 0 \quad (2)$$

$$\rho = \rho_0 \left[ \frac{a}{2T} \pm \sqrt{\frac{a^2}{T^2} - 4} \right] \quad (3)$$

This goes to  $\infty$  as  $T \rightarrow 0$ .

6. The energy has a term  $-a\rho/\rho_0$ , so it also blows up.