

Physics 831 Quiz #5 - Wednesday, Oct. 19, 2016

You may work in groups of three, but the group must contain nobody with whom you worked on a previous quiz. This quiz is open-note, open-book, open internet, and open-mind. However, you should not contact individuals outside your working group.

A large number of N diatomic molecules of mass m are confined to a region by a harmonic-oscillator potential,

$$V(\vec{r}) = \frac{1}{2}kr^2.$$

The system is at a sufficient temperature T so that the gas can be considered dilute. The temperature is in the range where rotational modes are routinely excited, but vibrational modes can be neglected.

- (10 pts) What is the energy per particle? Give your answer in terms of m , T , and k .
- (10 pts) Derive an expression for the entropy per particle in terms of the same variables. Begin with the expression,

$$S = \ln Z + \beta E,$$

where

$$Z = \frac{z^N}{N!},$$

and z is the partition function of a single molecule.

- (10 pts) If the spring constant is adiabatically changed from k_i to k_f , and if the initial temperature is T_i , find T_f .

Fun facts to know and tell: $\lim_{N \rightarrow \infty} \ln(N!) = N \ln N - N$

Solutions

First calculate partition function of single particle

$$\begin{aligned} z &= \int \frac{d^3p d^3r}{(2\pi\hbar)^3} \exp \left\{ -\beta \frac{p^2}{2m} - \beta \frac{kr^2}{2} \right\} \int d\ell (2\ell + 1) e^{-\beta\hbar^2\ell(\ell+1)/2I}, \\ &\approx \frac{1}{\hbar^3} \left(\frac{mT^2}{k} \right)^{3/2} \frac{2IT}{\hbar^2}, \\ &= \frac{2I}{\hbar^5} \left(\frac{m}{k} \right)^{3/2} T^4. \end{aligned}$$

where the approximation is that $\beta\hbar^2/I \ll 1$, so that the rotational states can be treated like a continuum.

- a) The energy per particle is

$$\begin{aligned} \frac{E}{N} &= -\partial_\beta \ln z \\ &= 4T \end{aligned}$$

b) The entropy is

$$\begin{aligned}\frac{S}{N} &= \ln z - \ln N + 1 + \beta \frac{\langle E \rangle}{N} \\ &= \ln \left[\left(\frac{m}{k} \right)^{3/2} \frac{2I}{N\hbar^5} T^4 \right] + 5\end{aligned}$$

c) Keep $T^4/k^{3/2}$ constant,

$$\begin{aligned}\frac{T_i^4}{k_i^{3/2}} &= \frac{T_f^4}{k_f^{3/2}} \\ T_f &= T_i \left(\frac{k_f}{k_i} \right)^{3/8}.\end{aligned}$$