PHY215-00: Physical (SI) Units

C.-P. Yuan

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1 Physical (SI) Units

1. Classical Mechanics

- Length: The SI unit is the meter (m). It is used to measure distances, displacements, lengths of objects, etc. For example, the length of a car might be measured in meters.
- Mass: The SI unit is the kilogram (kg). It is used to quantify the amount of matter in an object. The mass of a person or a box can be expressed in kilograms.
- **Time**: The SI unit is the second (s). It is used to measure the duration of events or the intervals between them. The time it takes for a pendulum to swing back and forth once can be measured in seconds.
- Force: The SI unit is the newton (N). One newton is defined as the force that causes a 1-kilogram mass to accelerate at a rate of 1 meter per second squared ($1 \text{ N} = 1 \text{ kg} \cdot 1 \text{ m/s}^2$). When you push or pull an object, the force you exert can be measured in newtons.
- Work and Energy: The SI unit for both work and energy is the joule (J). One joule is equal to the work done when a force of one newton acts through a distance of one meter (1 J = 1 N · 1 m). The energy stored in a spring

or the work done in lifting an object can be measured in joules.

- **Power**: The SI unit is the watt (W). One watt is equal to one joule per second (1 W = 1 J/s). It measures the rate at which work is done or energy is transferred. The power of a light bulb or an electric motor can be measured in watts.
- Velocity and Speed: The SI unit is meters per second (m/s). It tells how fast an object is moving and in what direction (for velocity). The speed of a car traveling on a highway might be 30 m/s.
- Acceleration: The SI unit is meters per second squared (m/s^2) . It measures the rate of change of velocity. When a car speeds up or slows down, its acceleration can be measured in m/s^2 .

2. Electromagnetism

- Electric Charge: The SI unit is the coulomb (C). It quantifies the amount of electric charge. The charge on an electron is approximately -1.6×10^{-19} C.
- Electric Current: The SI unit is the ampere (A). One ampere is defined as the flow of one coulomb of charge per second (1 A = 1 C/s). The current in an electrical circuit can be measured in amperes.

- Electric Potential Difference (Voltage): The SI unit is the volt (V). One volt is equal to one joule per coulomb (1V = 1 J/C). The voltage across a battery or in an electrical circuit can be measured in volts.
- Electric Resistance: The SI unit is the ohm (Ω) . Resistance is defined as the ratio of voltage to current (R = V/I). A resistor in an electrical circuit might have a resistance of 100Ω .
- Capacitance: The SI unit is the farad (F). One farad is defined as the capacitance of a capacitor that stores a charge of one coulomb when a potential difference of one volt is applied across it (1 F = 1 C/V).
- Magnetic Field Strength: The SI unit is the tesla (T). The magnetic field around a bar magnet or produced by an electromagnet can be measured in teslas.
- Magnetic Flux: The SI unit is the weber (Wb). Magnetic flux is related to the magnetic field and the area through which it passes ($\boldsymbol{\Phi} = \mathbf{B} \cdot \mathbf{A}$, where **B** is the magnetic field and **A** is the area). One weber is equal to one tesla-square meter (1 Wb = 1 T \cdot m²).

3. Thermodynamics

• **Temperature**: The SI unit is the kelvin (K). The kelvin scale is an absolute temperature scale. Zero kelvin (0 K)

is the lowest possible temperature, also known as absolute zero. The temperature of a room might be around $293 \,\mathrm{K}$.

• **Heat**: The SI unit is the joule (J). Heat is a form of energy transfer. The amount of heat required to raise the temperature of a substance can be measured in joules.

Relation between calorie and joule

- The calorie (cal) is a non-SI unit of energy. One calorie is defined as the amount of heat energy required to raise the temperature of 1 gram of water by 1 degree Celsius. The SI unit of energy is the joule (J). The conversion factor between them is 1 cal = 4.184 J.

Relation between Calorie (with a capital C) and other units

- When we talk about the energy content of food, the "Calorie" (with a capital C) is actually a kilocalorie (1 Cal = 1 kcal). Since 1 kcal = 1000 cal, and 1 cal = 4.184 J, then 1 $Cal = 1000 \times 4.184 J = 4184 J$.

- **Entropy**: The SI unit is joules per kelvin (J/K). Entropy is a measure of the disorder or randomness of a system.
- Internal Energy: The SI unit is the joule (J). The internal energy of a gas or a thermodynamic system is the sum of the kinetic and potential energies of its molecules and can be measured in joules.
- **Pressure**: The SI unit is the pascal (Pa). One pascal is

equal to one newton per square meter $(1 \text{ Pa} = 1 \text{ N/m}^2)$. The pressure exerted by a gas or a fluid can be measured in pascals.

Volume: The liter (L) is a unit of volume in the metric system. One liter is defined as the volume of a cube that is 10 centimeters (0.1 meters) on each side. Mathematically, 1L = 1000 cm³ = 10⁻³ m³. It is commonly used to measure the volume of liquids.

4. Quantum Physics

- Energy: The SI unit is the joule (J). In quantum mechanics, the energy of a photon, for example, is related to its frequency by E = hf, where h is Planck's constant and f is the frequency.
- **Frequency**: The SI unit is the hertz (Hz). One hertz is equal to one cycle per second. The frequency of an electromagnetic wave or a quantum mechanical oscillator can be measured in hertz.
- Angular Momentum: The SI unit is joule-seconds $(J \cdot s)$. In quantum mechanics, angular momentum is quantized. Planck's constant $(h = 6.626 \times 10^{-34} \text{ J} \cdot \text{s})$ has the unit of angular momentum and is fundamental in quantum physics.

• Wave Function: The wave function in quantum mechanics is a complex-valued function and doesn't have a physical unit in the traditional sense. However, the square of the absolute value of the wave function $(|\psi|^2)$, which gives the probability density, is dimensionless.